

HT Combinatorial Screening of Novel Materials for High Capacity Hydrogen Storage

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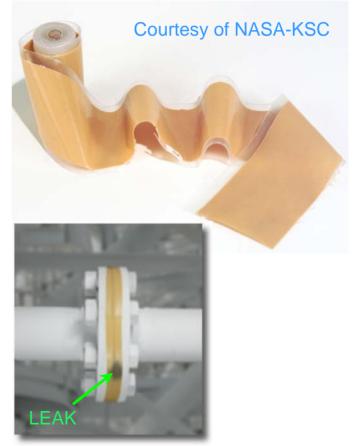
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- Develop (*i.e.* design, build, test and verify) a high throughput screening device based on FSECdeveloped H₂ sensing materials applicable to a broad range of adsorbents, including MOFs, various doped & undoped hydrides so that methods for increasing H₂ adsorption energies beyond 5 kJ/mol can be identified.
- Develop a rapid screening method & demonstrate the ability to reproducibly screen 10/100 hydrogen sorbing materials per run.
- Measure the amount of H₂ released for up to 10/100 solid adsorbents, simultaneously, from SA/LN2 to 423 K (150°C) and pressures from 10 torr to 50 bars.



- Industries producing or consuming H₂ (*e.g.* refineries) & others (*e.g.* NASA).
- Hydrogen Economy
 - Transportation
 - Storage
 - Fuel Cells





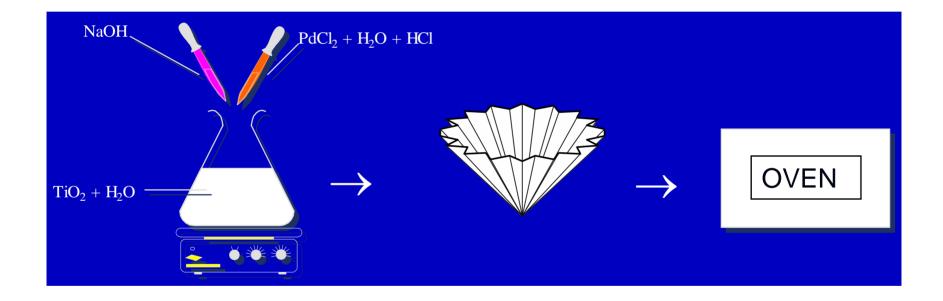
At FSEC, we have:

- Formulated, prepared & characterized many novel chemochromic materials for the visual detection of hydrogen gas.
- Developed a technique for the implementation of PdO-based & other "smart pigments". Bokerman, G., et al. "Gas Permeable Chemochromic Composition for Hydrogen Sensing," U.S. Patent and Trademark Office Serial No. 11/414,900, May 2006.
- Synthesized novel chemochromic materials based on POM complexes and <u>others</u>.



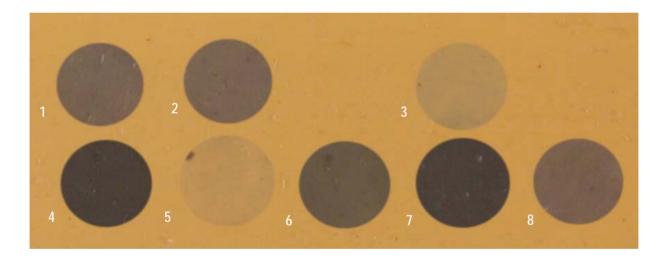
- Change color at temperatures as low as -40°C
- Employ gas permeable matrices for the pigment encapsulation that make them selective toward hydrogen detection
- Has been fielded by NASA.





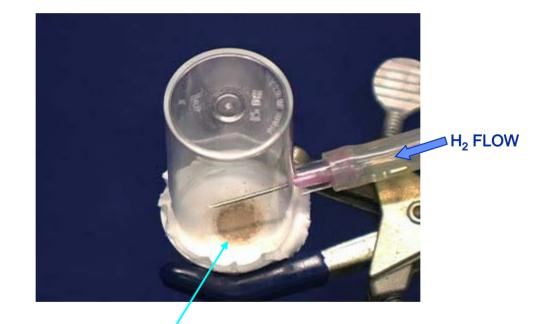
TiO ₂ Sample	Degussa P-25	Aldrich	Fisher	DuPont R103
Particle size (µm)	0.02-0.04	0.5	0.1-0.5	0.05-0.2





A 25-mil thick section of chemochromic film after exposure to various quantities of hydrogen gas.





CHEMOCHROMIC MARKER

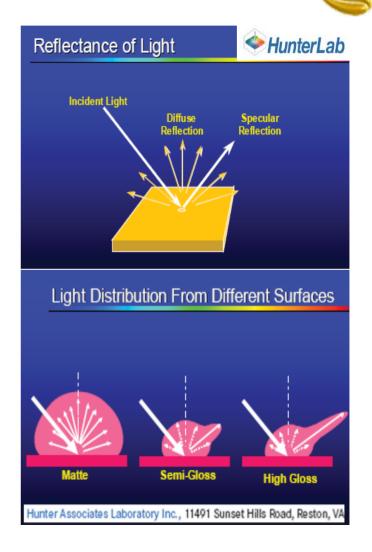


Quantifying Color Change

Parameters affecting the appearance of color are:

 Basic pigment formulation
Surface characteristics (gloss, texture, pattern, etc.)

0/45- or 45/0-degree instruments are most often used as they avoid the specular component, like a person does when examining the appearance of color.



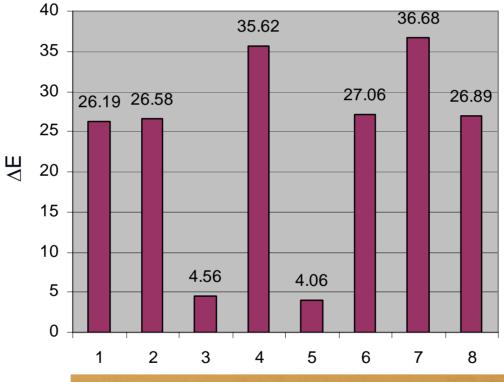


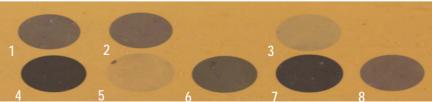
Using ColorTec-PCM colorimeter

 $\Delta E^* = [(\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2]^{\frac{1}{2}}$

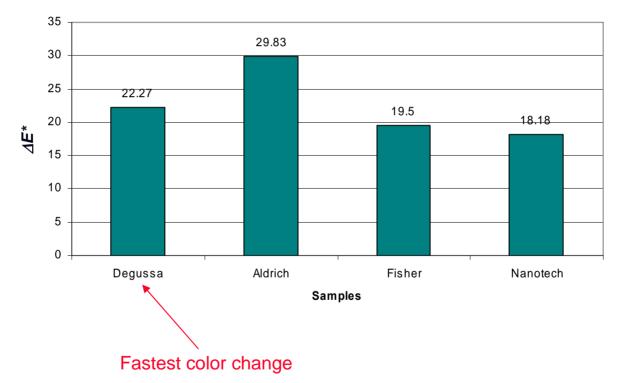
- L*- Lightness Value a*- position on red-green axis b*- position on yellow-blue axis





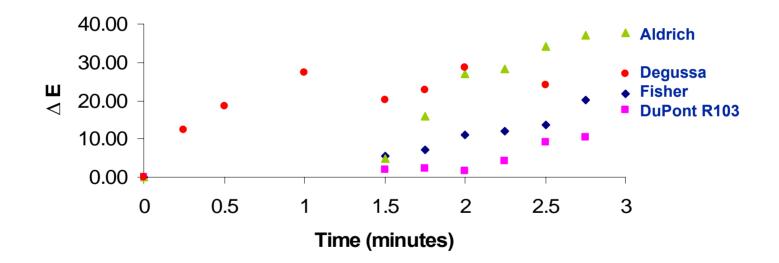






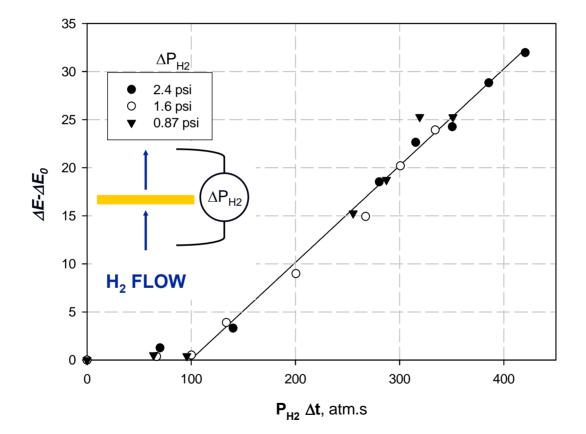
Captain, J.E., et al. "Chemochromic Hydrogen Detection," *Proceedings of SPIE Conference*, 17–21 April **2006**, Orlando, Florida.





Mohajeri, et al., "Effect of PdO on TiO₂ Loading on Chemochromic Detection of Hydrogen," submitted for publication in the Int. J. Hydrogen Energy.





Data courtesy of NASA-KSC & Arctic Slope Research Corp.

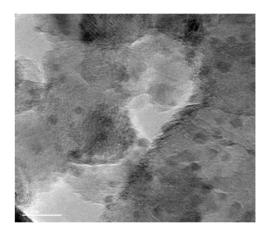


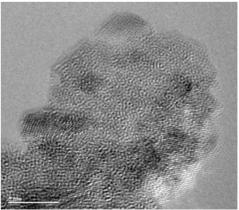


Strongly attached to TiO₂ support













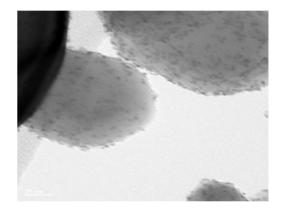
 \clubsuit Loosely attached to TiO₂ support

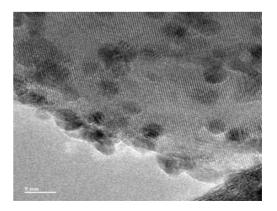


Non-homogenous

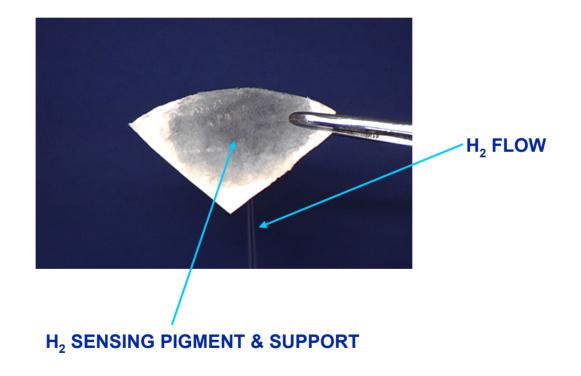


Particle size > 5 nm



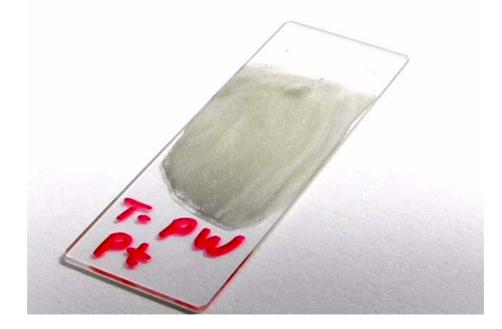


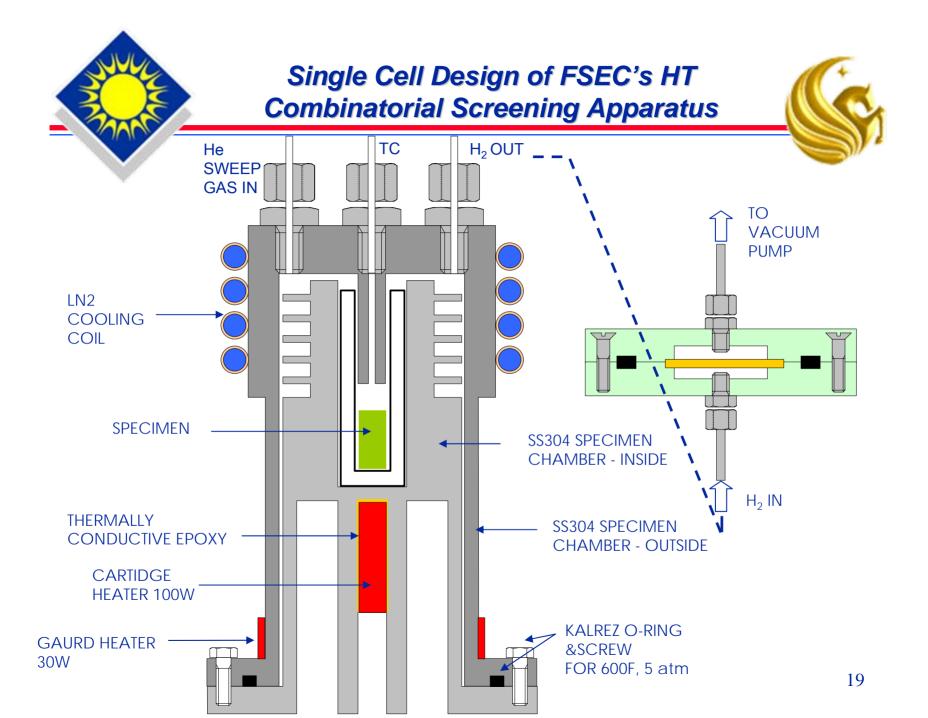






Several formulations developed at FSEC can be used in "repeated coloration/discoloration" applications including HT screening of hydrogen storage materials as the following video shows.











♦ A large number of hydrogen sensing pigments suitable for HT
H₂ storage materials screening have been synthesized using
PdO on TiO₂ supports and other compounds

The extent of color change has been determined as a function of time and amount of H_2 within the membranes

PdO on Degussa P-25 with smallest TiO₂ particles shows fastest discoloration and highest color contrast

✤ A HT combinatorial screening apparatus has been designed for testing up to 100 individual samples, in few minutes, at temperatures & pressures in the range of LN2-150°C and 10-38 ktorrs, respectively.



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U.S. Navy & DLA