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# *Membrane and MEA Accelerated Stress Test Protocols*

*Presented at High Temperature Membrane  
Working Group Meeting*

*Washington, DC*

*May 14, 2007*

*T.G. Benjamin*

*Argonne National Laboratory*

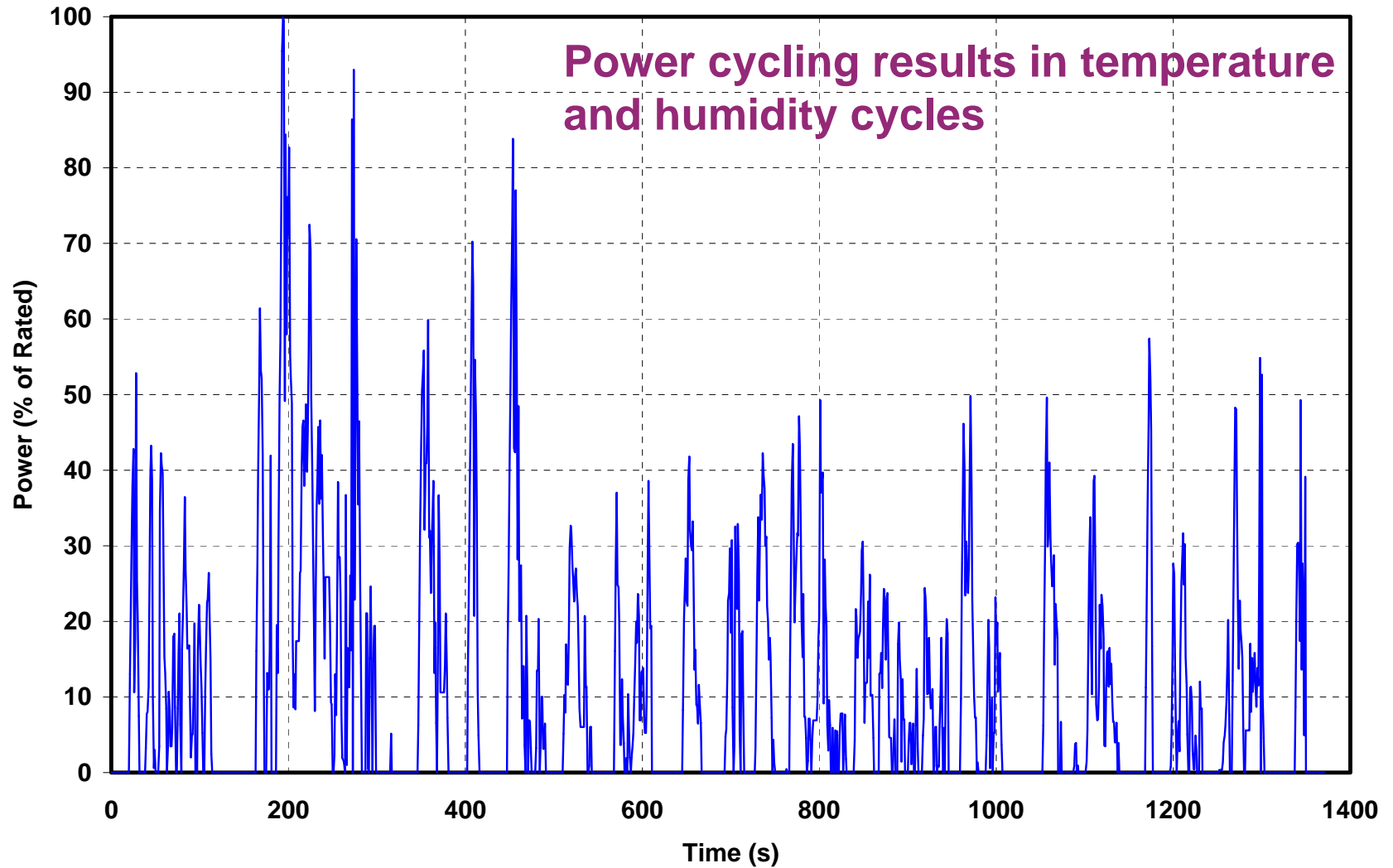


U.S. Department  
of Energy

UChicago ►  
Argonne<sub>LLC</sub>

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# Federal Urban Driving Schedule (FUDS)



# *Degradation Mechanisms*

- **Mechanical cycling**
  - Humidity cycling causes swelling and shrinkage, generating mechanical stresses in the membrane
- **Chemical attack (by peroxy radical) causes ionomer damage and loss of functionality and integrity**

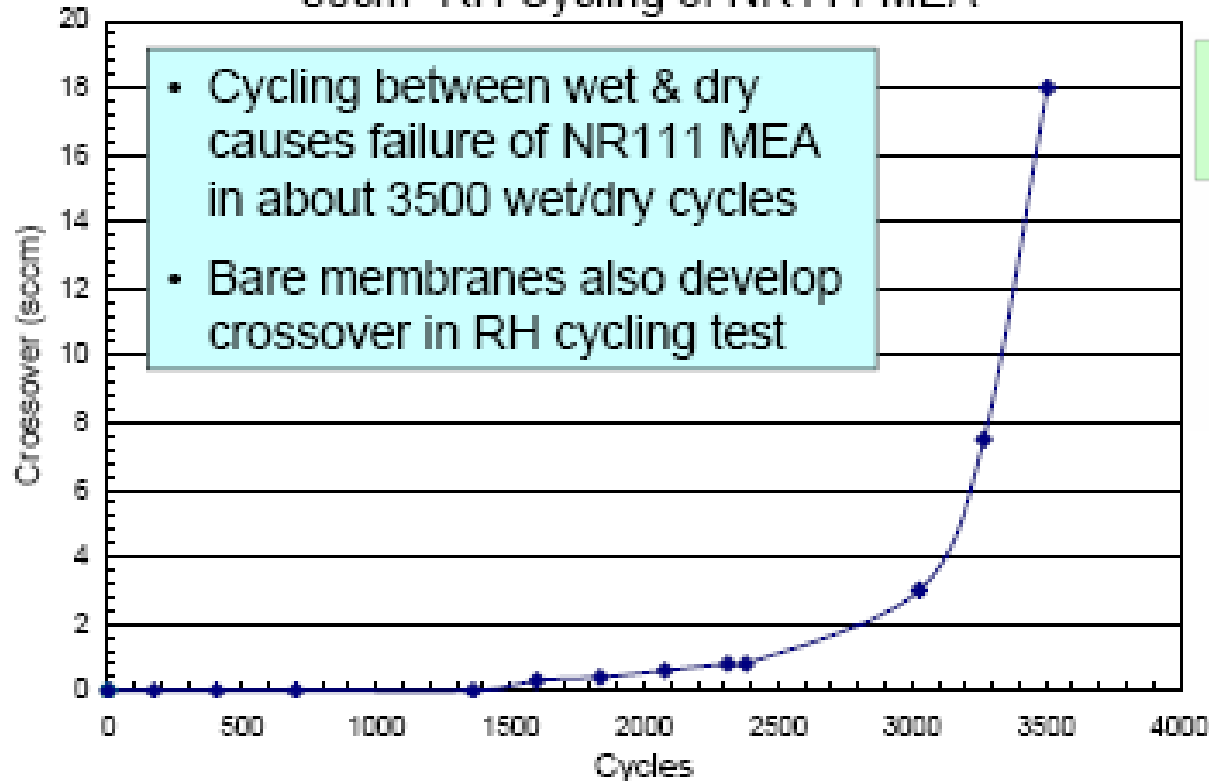
*Mike Hicks, formerly of 3M and now of IdaTech, said,*

- **Accelerated Stress Test Protocols are needed to reduce new product introduction cycle**
  - Can't wait 3 years to determine if product lasts 3 years
  - Need estimate of lifetime now
  
- **DOE 2010 stationary MEA lifetime target of 40,000 hours**
  - Should have started the life test in June 2005
  
- **Failure mode in accelerated test must be the same as the failure mode in “normal” operation.**

2<sup>nd</sup> MEA Manufacturing Symposium  
Dayton, OH  
August 23, 2006

# RH Cycling: Results

50cm<sup>2</sup> RH Cycling of NR111 MEA



- Cycling between wet & dry causes failure of NR111 MEA in about 3500 wet/dry cycles
- Bare membranes also develop crossover in RH cycling test

Characterization of Perfluorosulfonic Acid Membranes for PEM Fuel Cell Mechanical Durability

Mike Budinski, Craig Gittleman, Yeh-Hung Lai, Brian Litteer & Dan Miller

*General Motors Corporation  
Fuel Cell Activities  
Honeoye Falls, NY*

November 11, 2004

Annual AIChE meeting

Austin, TX

RH Test cycling accelerates mechanical failures in the absence of electrochemical degradation

Fuel Cell Activities

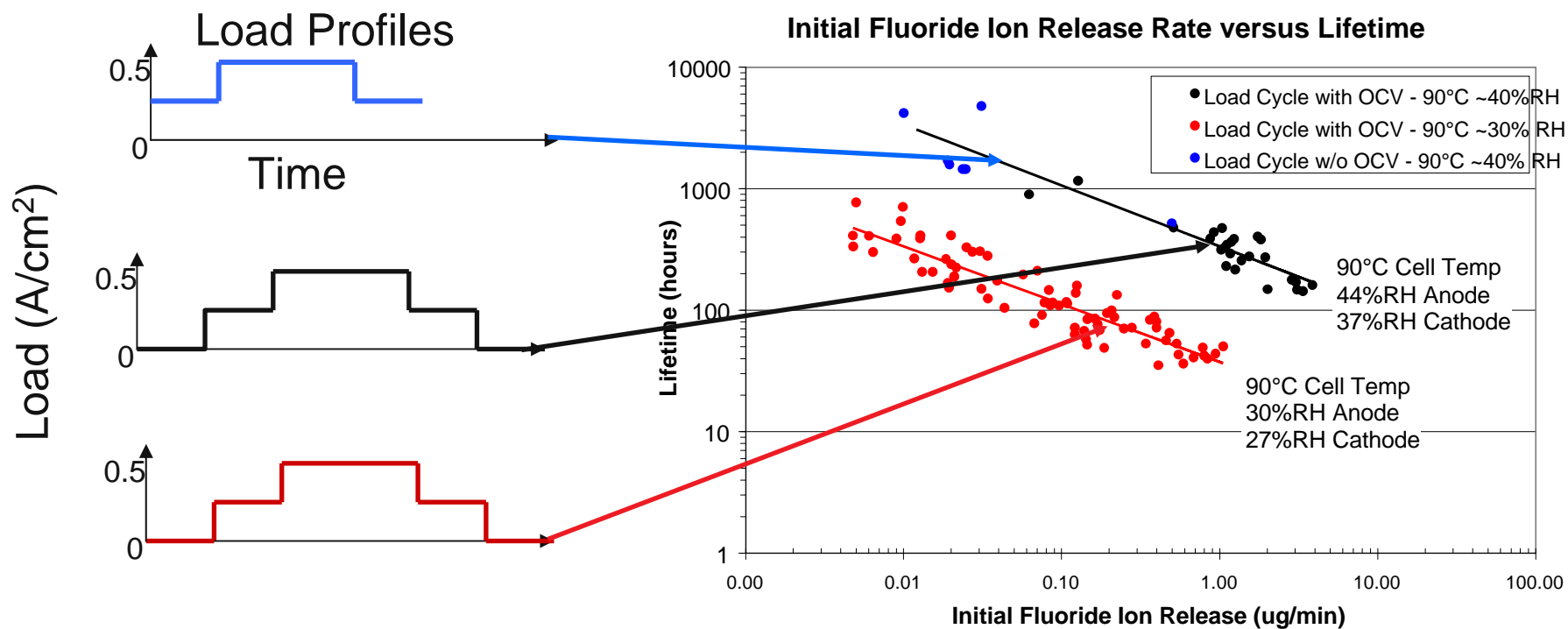
November 11, 2004



**General Motors**

# Accelerated Testing: Initial Fluoride Release vs. Lifetime

Cycle characteristics affect degradation.



Courtesy Mike Yandrasits, 3M

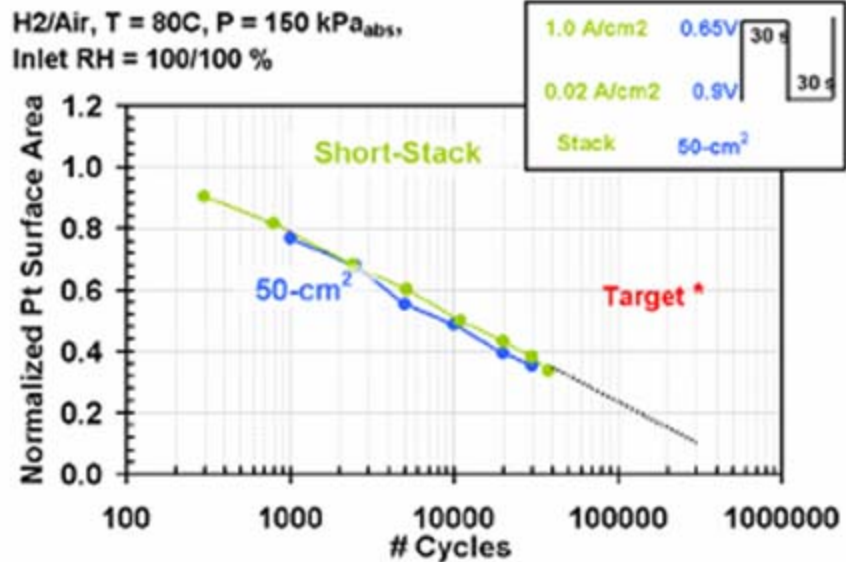
## Pt Surface Area Loss Target

↳ catalysts must have  $\leq 3 \mu\text{V/hr}$  degradation from Pt surface area loss during 5500 hrs of operation

→  $\leq 16.5 \text{ mV}$  degradation in 5500 hrs

→  $\leq 16.5 \text{ mV}$  degradation in 300,000 large- $\Delta V$

→ Nominally  $\leq 40\%$  Pt surface area loss in 300,000 large- $\Delta V$





# DOE MEA Chemical Stability

**Table 3**  
**MEA Chemical Stability and Metrics**

<b>Test Condition</b>	<b>Steady state OCV, single cell 25 - 50cm<sup>2</sup></b>
<b>Total time</b>	200 h
<b>Temperature</b>	90°C
<b>Relative Humidity</b>	Anode/Cathode 30/30%
<b>Fuel/Oxidant</b>	Hydrogen/Air at stoics of 10/10 at 0.2 A/cm <sup>2</sup> equivalent flow
<b>Pressure, inlet kPa abs (bara)</b>	Anode 250 (2.5), Cathode 200 (2.0)

<b>Metric</b>	<b>Frequency</b>	<b>Target</b>
<b>F<sup>-</sup> release or equivalent for non-fluorine membranes</b>	At least every 24 h	No target – for monitoring
<b>Hydrogen Crossover (mA/cm<sup>2</sup>)*</b>	Every 24 h	≤20 mA/cm <sup>2</sup>
<b>OCV</b>	Continuous	≤20% loss in OCV
<b>High-frequency resistance</b>	Every 24 h at 0.2 A/cm <sup>2</sup>	No target – for monitoring

\*Crossover current per USFCC “Single Cell Test Protocol” Section A3-2, electrochemical hydrogen crossover method



# DOE Membrane Mechanical Cycle

**Table 4**  
**Membrane Mechanical Cycle and Metrics**

<b>Cycle</b>	<b>Cycle 0% RH (2 min) to 90°C dewpoint (2 min), single cell 25 - 50cm<sup>2</sup></b>	
<b>Total time</b>	Until crossover >10 sccm or 20,000 cycles	
<b>Temperature</b>	80°C	
<b>Relative Humidity</b>	Cycle from 0% RH (2 min) to 90°C dewpoint (2 min)	
<b>Fuel/Oxidant</b>	Air/Air at 2 slpm on both sides	
<b>Pressure</b>	Ambient or no back-pressure	
<b>Metric</b>	<b>Frequency</b>	<b>Target</b>
<b>Crossover*</b>	Every 24 h	≤10 sccm

\*Crossover per USFCC “Single Cell Test Protocol” Section A3-1, pressure test method with 3 psig N<sub>2</sub>

M. Mathias et al., ECS Interface Vol. 14 No. 3, 2005, pp. 24-35

# *US Fuel Cell Council (USFCC) Durability Task Force*

**The Mission of the USFCC Durability Task Force is to establish standardized non-application specific, accelerated test protocols to evaluate the durability of various fuel cell components.**

## Initial Focus

### – Catalyst

- *Platinum sintering, agglomeration, and dissolution*
- *Carbon catalyst support oxidation/corrosion*

### – Membrane

- *Chemical degradation (Peroxy radical attack)*
- *Mechanical degradation*

*[www.usfcc.com](http://www.usfcc.com)*

# DOE MEA Chemical Stability – USFCC Differences

**Table 3**  
**MEA Chemical Stability and Metrics**

<b>Test Condition</b>	<b>Steady state OCV, single cell 25 - 50cm<sup>2</sup></b>	
<b>Total time</b>	200 h	
<b>Temperature</b>	90°C	
<b>Relative Humidity</b>	Anode/Cathode 30/30%	
<b>Fuel/Oxidant</b>	Hydrogen/Air at <b>H<sub>2</sub>/O<sub>2</sub> or H<sub>2</sub>/40% O<sub>2</sub></b> m <sup>2</sup> equivalent flow	
<b>Pressure, inlet kPa abs (bara)</b>	Anode 250 (2.5), Cathode 200 (2.0)	
<b>Metric</b>	<b>Frequency</b>	<b>Target</b>
<b>F<sup>-</sup> release or equivalent for non-fluorine membranes</b>	At least every 24 h	No target – for monitoring
<b>Hydrogen Crossover (mA/cm<sup>2</sup>)*</b>	Every 24 h	≤20 mA/cm <sup>2</sup>
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*Crossover current per USFCC “Single Cell Test Protocol” Section A3-2, electrochemical hydrogen crossover method		



# DOE Membrane Mechanical Cycle – USFCC Differences

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<b>Pressure</b>	Ambient or no back-pressure	
<b>Metric</b>	<b>Frequency</b>	<b>Target</b>
<b>Crossover*</b>	Every 24 h	≤10 sccm

**0 - 150% RH**

\*Crossover per USFCC “Single Cell Test Protocol” Section A3-1, pressure test method with 3 psig N<sub>2</sub>

## ***USFCC and DOE Additional Differences***

- **USFCC Draft Protocol includes Fenton's test as an *ex-situ* membrane chemical stability assessment.**
- **USFCC Draft Protocol also includes DuPont DD-4 which is a combined (alternating) humidity and load cycle.**

## **Ex-situ *Chemical Stability Test: Fenton's Test* from USFCC Materials and Components Working Group meeting at the Fuel Cell Seminar 2006**

Goal: assess the relative oxidative stability of PEMs

The degree of degradation measured by polymer weight loss, fluoride ion evolution, change in ion exchange capacity (IEC), etc.

Recommended test conditions for PFSA membranes:

- 30% H<sub>2</sub>O<sub>2</sub>
- 20 ppm Fe<sup>+2</sup>
- 85<sup>0</sup>C
- 3 Cycles with fresh reagent, 18 hours per cycle
- Measure fluoride and weight loss

## **Chemical + Mechanical Stability Test - from USFCC Materials and Components Working Group meeting at the Fuel Cell Seminar 2006**

Goal: combine chemical and mechanical degradation mechanisms in a single accelerated test

Test protocol recommended by DuPont

**Two cycle modes are interchanged every 24 hrs:**

***Humidity cycle:***

N<sub>2</sub> / N<sub>2</sub>, 80 °C

RH of inlet gases is cycled between 0 and 100% RH every 30 minutes

***Load cycle:***

H<sub>2</sub> / O<sub>2</sub>; 50%RH, 80 °C

Load cycled between 10 and 800 mA/cm<sup>2</sup> (7min/3 min)

**Monitor crossover current density as a function of time**

***Stop test when > 10 mA/cm<sup>2</sup>***

## *Thanks for your attention and ... Remember*

- **The accelerated stress test protocols have not been generally correlated with actual life under “normal” operating conditions.**
- **The protocols are test cycles only. Conditioning procedures and analysis techniques are not described.**
- **Membranes other than PFSA may need different cycles.**
- **Visit [www.USFCC.com](http://www.USFCC.com) for:**
  - Existing USFCC Single Cell Test Protocol
  - Future USFCC Catalyst Stability Accelerated Stress Test Protocols (first draft)
  - Future Membrane/MEA Accelerated Stress Test Protocols



## *And Remember*

- **DOE Solicitation DE-PS36-GO95020 – “High Temperature, Low Relative Humidity, Polymer-Type Membranes” – says**

“Applicants should also show that the material can be expected to meet durability targets in the aggressive environment of a fuel cell, i.e., the material must have good chemical stability and be resistant to oxidation by peroxide.”

- **Funding Opportunity DE-PS36-06GO96017 - Research and Development of Fuel Cell Technology for the Hydrogen Economy - says**

“Additionally, the material must demonstrate the ability to meet the cost and durability targets in the aggressive environment of the fuel cell, and have good mechanical and chemical stability under highly oxidizing conditions.”