

Hydrogen Storage by the Reversible Hydrogenation of Liquid and Solid Substrates

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Objectives

- **New Project (3 years) – Agreement signed 2/04**
 - **Overall project objectives:**
 - **Develop carbon-based solid and liquid hydrogen storage materials with capacities of >6 wt. % and >45 g H₂/L**
 - **Develop hydrogen storage system prototype with 6 wt. % and 45 g H₂/L capacity in the range of –40 to 90-120 °C and less than 1000 psia H₂ pressure**
- **Our current primary technical objectives:**
 - **Discovery of novel substrates using experimental and computational methods**
 - **Demonstration of reversible hydrogen storage with liquid substrates**
 - **new liquid development**
 - **catalyst development**
 - **1st year milestone: 3 wt. % hydrogen storage at temperatures at or below 150 °C**
 - **preliminary regeneration cost analysis**

Budget

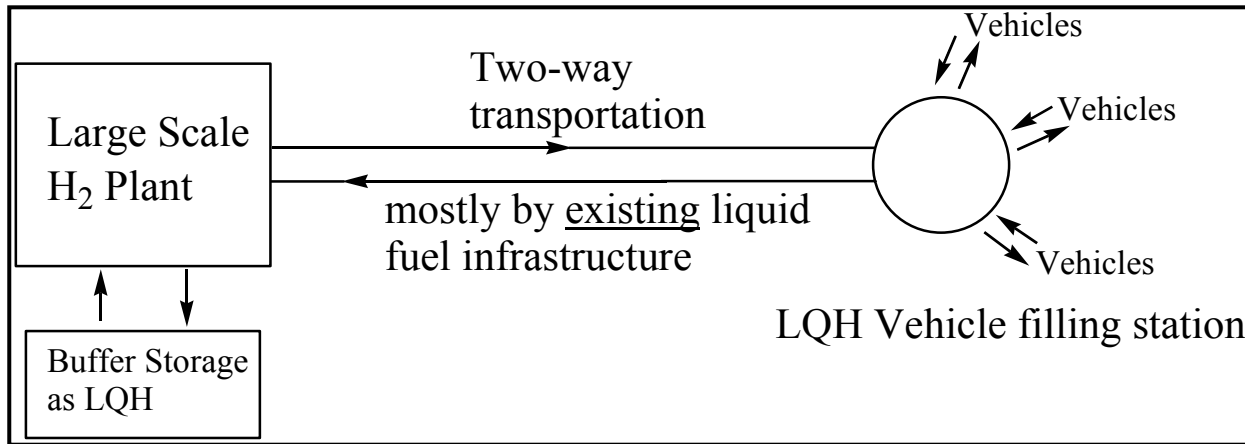
- **Total funding for the project (3 yrs.)**
 - **\$6,121,279**
 - **DOE share** **\$4,346,108 (71%)**
 - **Air Products share** **\$1,775,171 (29%)**
- **Funding FY04:**
 - **DOE funding estimated at \$780,000**

Technical Barriers and Targets

- **Technical Barriers- Hydrogen Storage**
 - A. Cost: \$3.00 per gallon gasoline equiv. (2005), \$1.50 per gallon gasoline equiv. (2010)
 - B. Weight and Volume: 4.5 wt. % and 36 g H₂/L (2005), 6.0 wt. % and 45 g H₂/L (2010)
 - C. Efficiency
 - G. Life Cycle & Efficiency Analyses
 - Q. Regeneration Processes
 - R. Byproduct Removal
- A. Cost: Facile regeneration of liquid substrates by hydrogenation at large central facilities
- B. Weight and Volume: Ongoing discovery effort for liquid and solid substrates with high useable capacity
- C. Efficiency: Total cycle – use and regeneration
- G. Life Cycle & Efficiency Analyses: Modeling of regeneration economics

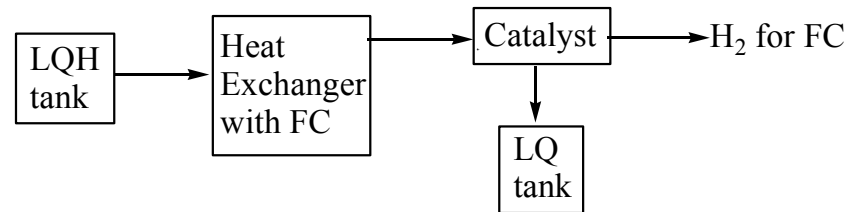
Approach

Utilizing Temperature/Pressure-Reversible “Liquid Hydride” (LQ*H₂) Carriers



Prior art LQ*H₂ compositions include cyclohexane and decalin.

In fuel cell (FC) vehicle or stationary power source:



Maximum energy efficiency: by (a) recovering the exothermic ($-\Delta H$) of hydrogenation and (b) utilizing the waste heat from the power source to supply the ΔH for the endothermic dehydrogenation.

Assuming an LQ*H₂ ($d=1\text{g/cc}$) carrying 6 wt. % H₂, 4 Kg H₂ (sufficient for compact FC vehicle 300-400 mile range) is equivalent to 18 gallons of the “liquid hydride” fuel.

Performance Limits and Comparison of “Chemical Hydride” Liquid Carriers

	Inorganic Hydride Hydrolysis	LQ*H ₂
Max. Capacity (wt. % H ₂)	6.5-11 (for current chemical hydrides)	6-7 No reagents – H ₂ O, etc.
ΔH (kcal/mol H ₂) for H ₂ release	-13 to -33 (exothermic) Spontaneous generation of H ₂ . Reaction heat has to be dissipated to environment.	10 to 15 (endothermic) Allows utilization of waste heat from the power source.
ΔG (kcal/mol H ₂) for H ₂ release	-19 to -38 Spontaneous at ambient temperature, cannot be reversed with H ₂ alone.	~0 (at reaction temp.) Spontaneous upon heating, but only in the presence of appropriate catalysts.
Regeneration	Requires multi-step chemical processing or a direct electrochemical synthesis.	LQ* is easily regenerated with H ₂ alone, using known, inexpensive catalytic hydrogenation technology.

Project Safety

- **Air Products: A world leader in hydrogen safety. Safety is an integral part of our corporate culture**
- **We utilize detailed hazard reviews for all lab operations and lab equipment**
 - **Use of written, peer-reviewed hazard reviews**
 - **Corporate EH&S oversight**
- **Specific engineering controls:**
 - **Hydrogen monitors that automatically shut down equipment in the event of a hydrogen leak**
 - **Overpressure and over-temperature controls**
- **We have strict chemical hygiene rules for synthesis work on solid and liquid substrates**

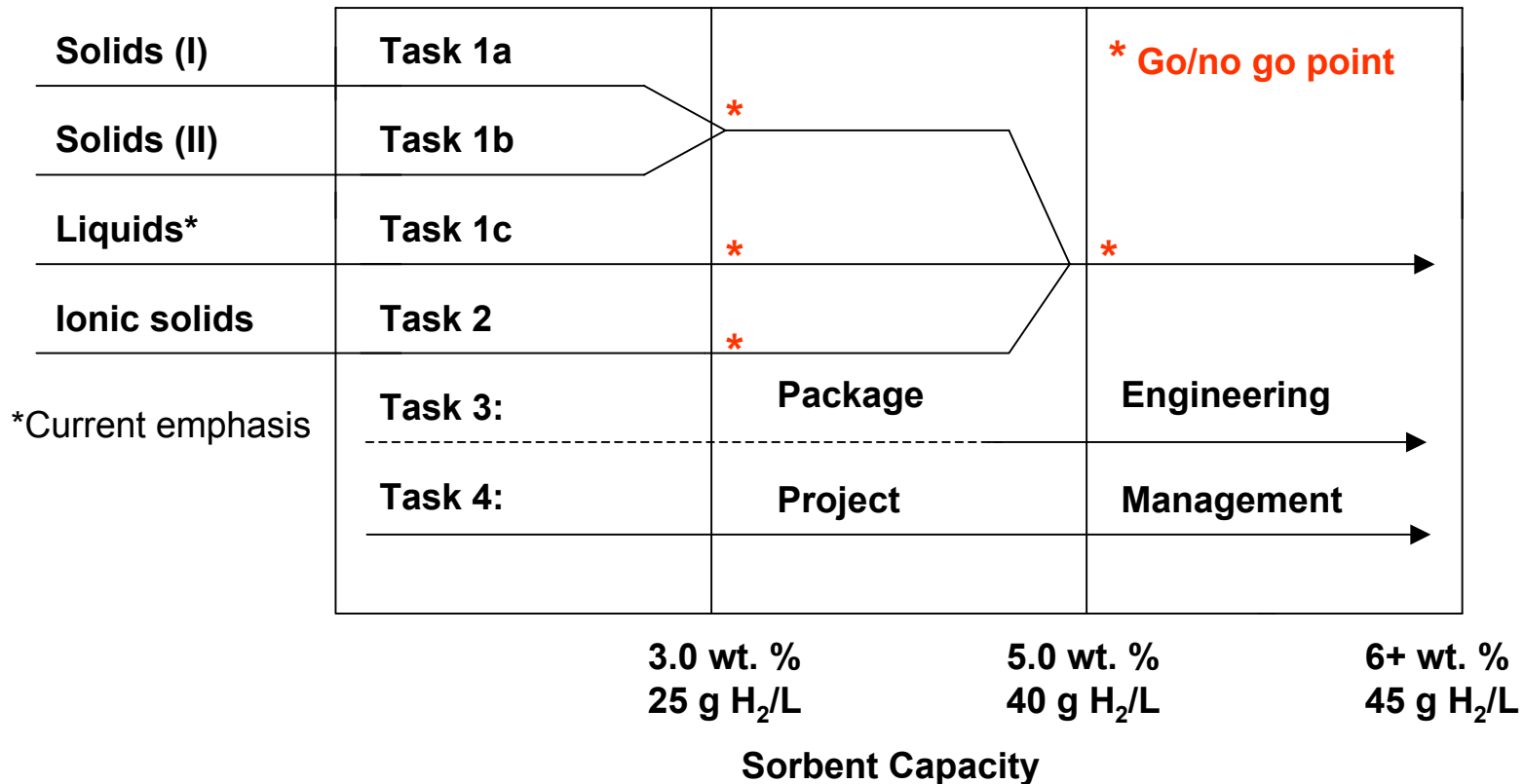
Project Timeline

Year: Discovery 1 Cycling & Optimization 2 Integration 3

Mar 1, 2004 –
Feb 28, 2005

Mar 1, 2005 –
Feb 28, 2006

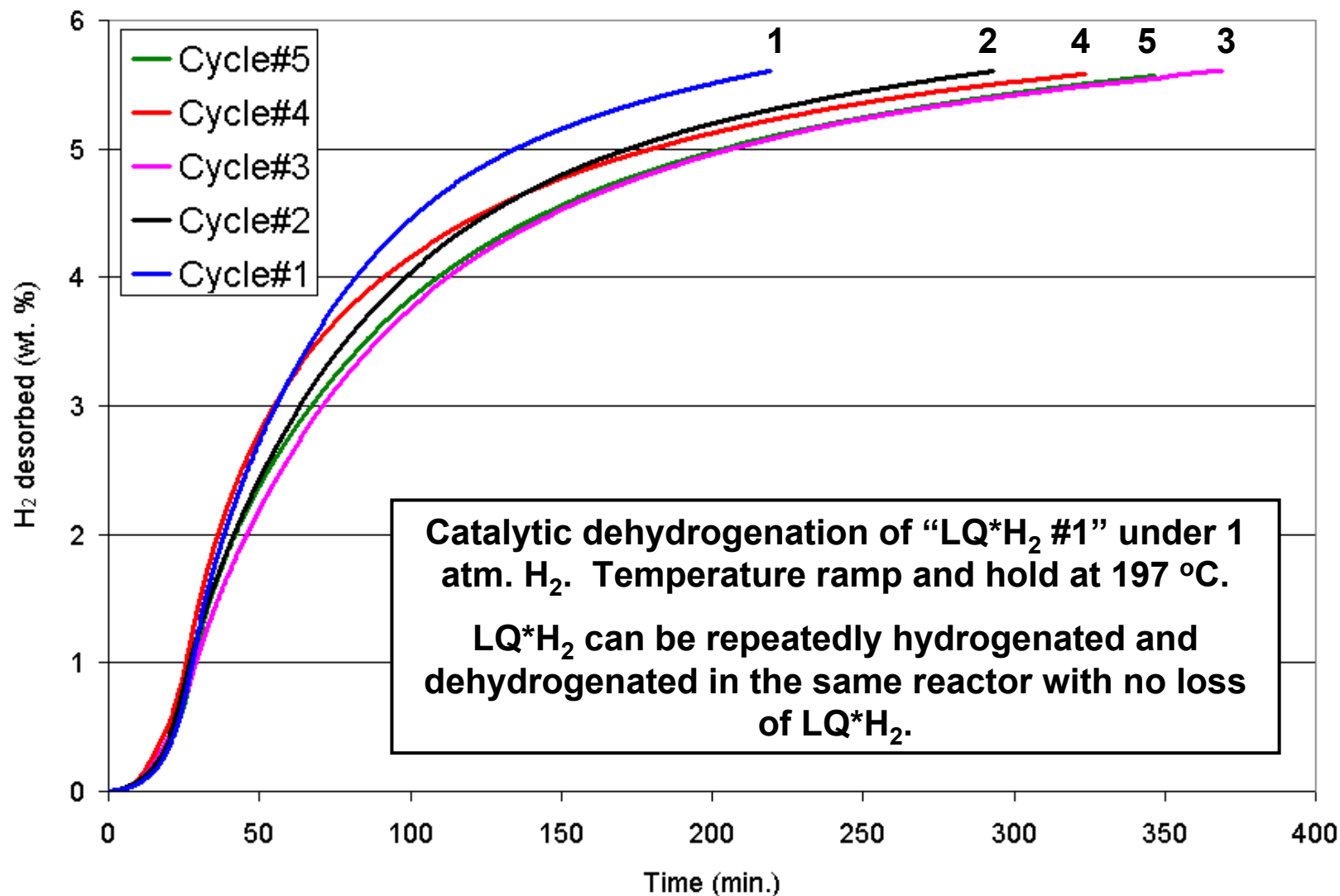
Mar 1, 2006 –
Feb 28, 2007



Technical Accomplishments

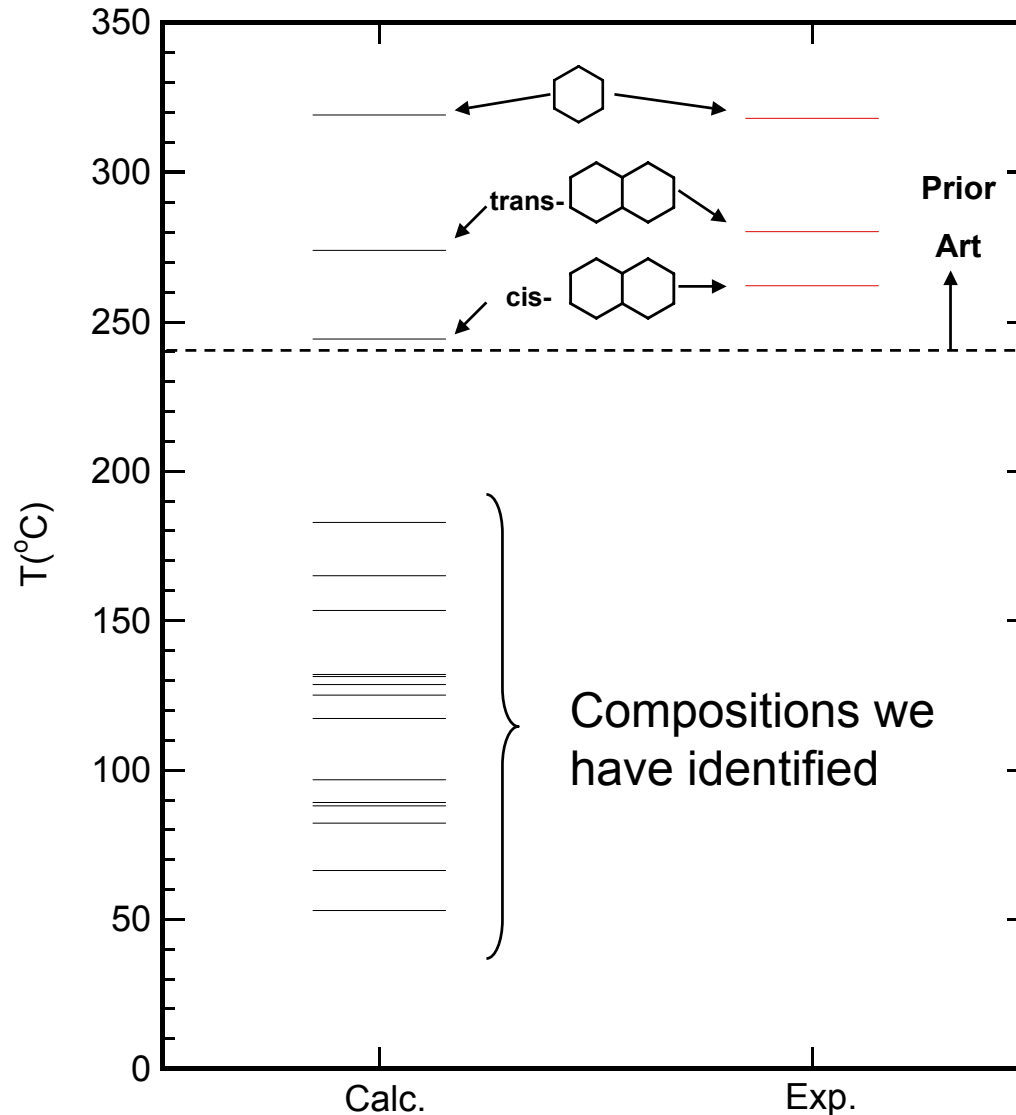
- We have developed a number of novel liquid-phase hydrogen carriers. One example is “LQ*H₂ composition #1”:
 - 5-7 wt. % H₂ gravimetric capacity
 - >50 g H₂/l volumetric capacity
 - Experimental heat of hydrogenation, ΔH , is -12.1 kcal/mol H₂ (cf. decalin ΔH = -15.8 kcal/mol H₂)
 - Low volatility
 - vapor pressure is approximately 10 torr at 200 °C
 - LQ*H₂ can be dehydrogenated to provide hydrogen free of liquid carrier contamination with no need for condensers or membranes

Reversible LQ*H₂ Dehydrogenation to LQ*



5 cycles with no capacity loss or systematic changes in hydrogenation/dehydrogenation rate

Dehydrogenation Temperatures for LQ*H₂ Liquid Substrates



Temperatures for a 95% dehydrogenation of LQ*H₂ in equilibrium with LQ* and 1 atm. H₂: (a) As calculated by high level predictive models and (b) As derived from published thermodynamic experimental data.

Preliminary Projection of LQ* to LQ*H₂ Regeneration Cost

- Primary cost drivers include:
 - Capital for plant and LQ*H₂
 - Catalyst and LQ*H₂ operating lifetime
 - Energy recovery from LQ*H₂ regeneration
- Scale used in projections:
 - 15MM - 90MM kg H₂/year
- Regeneration cost range:
 - \$0.19 - \$0.73/kg H₂

Interactions and Collaborations

- Fuel cell manufacturers: discussions of heat integration of dehydrogenation reactor with fuel cells
- Military fuel-cell developers: discussions of potential uses for liquid hydrogen storage substrates with existing fuel cell applications
- Penn State University: discussions of possible collaborative research with faculty
- Southwest Research Institute hydrogen storage measurement laboratory: assistance provided for instrument selection and high pressure H₂ handling issues
- Carbon H₂ Storage Materials Working Group: Standard sample testing and discussions of measurement standardization
- Publications (3) and presentations (2) in last year

Response to FreedomCAR Tech Team Reviewers Comments (2/19/04)

- **Scope of project** is too ambitious-
 - We will downselect at the end of year 1 and year 2 to focus work on the most promising hydrogen storage substrates
- **Technical Feasibility:** Do liquid compositions exist than can provide a distinct improvement over the simple aromatics (benzene, naphthalene) previously described for hydrogen storage/delivery? What about catalyst and liquid lifetimes?
 - We have shown experimental and computational data that proves that significant advances are possible
 - Extended liquid cycling and catalyst lifetime studies slated for year 2
- **Regeneration Cost / Energy Efficiency:**
 - We have preliminary economic evaluations that suggest that the “liquid carrier” concept could meet the DOE cost targets for onboard delivered hydrogen.
 - Off-board H₂ storage and H₂ delivery possibilities are being addressed with DOE.

Future Work

- **Remainder of Year 1: Continue discovery effort towards “ideal” liquid and solid hydrogen carriers**
 - predictive computational modeling
 - experimental efforts on selected substrates
 - catalyst screening
- **Year 2:**
 - More emphasis on cycling/lifetime studies
 - Preliminary engineering of hydrogen storage system (dehydrogenation reactor)
 - Catalyst development and hydrogenation/dehydrogenation rate optimization