Acrylic Materials in PV Applications:

Making an Informed Choice

Evonik Cyro LLC Acrylic Polymers Peter Colburn



NREL PV Reliability Workshop Feb 17, 2011

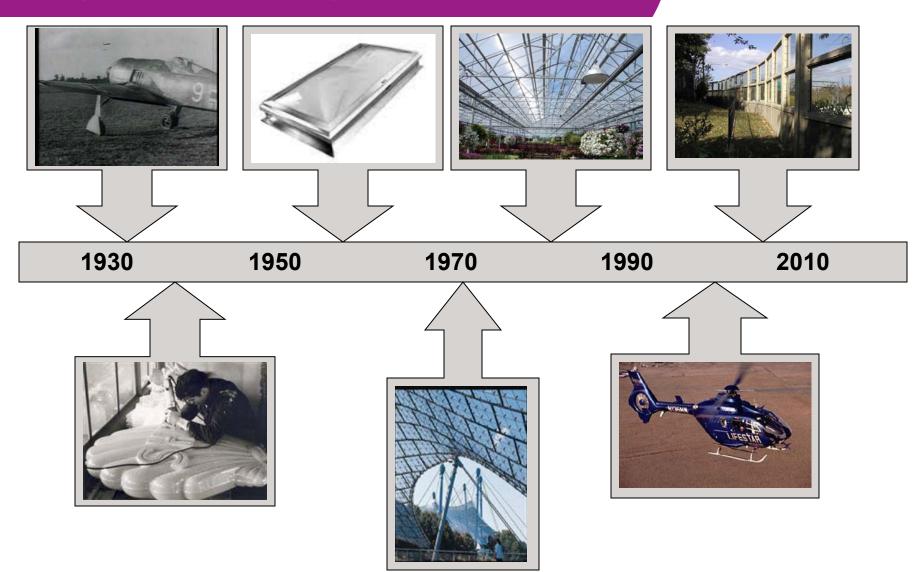


Topics

- **1. Outdoor Acrylic Applications**
- 2. What is Acrylic?
- **3. UV Absorption Characteristics**
- 4. Degradation Mechanisms
- **5. Protection Systems**
- 6. Weathering Performance

Acrylic Use History







Automotive

- > Lighting
- Pillar covers
- Mirror housings









Architectural Glazing





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Aircraft Glazing

- Helicopter glazing
- General aircraft glazing
- Commercial aircraft cabin windows
- Fighter canopies and windshields
- Glider canopies











Highway Noise Barriers

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HID & Outdoor Lighting Applications



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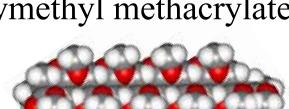
Topics

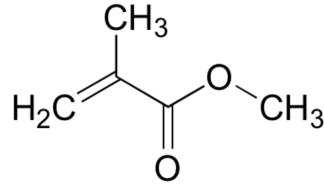
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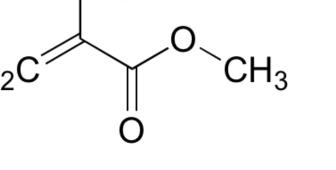
What is Acrylic?

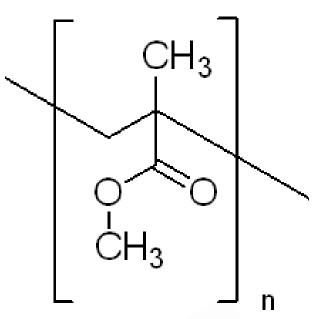
methyl methacrylate

Polymethyl methacrylate





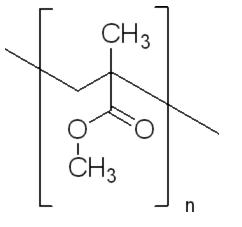






What is "acrylic"?

- Polymethylmethacrylate: -(MMA)_n-
 - PMMA is an ester
- Monomer Methyl methacrylate
- Comonomer (methyl or ethyl –acrylate)
- Additives
 - Lubricants
 - Stabilizers
 - Colorants





Key Performance Characteristics



- Long Term Stability in Outdoor Environments
- Additional Protection Systems
- Excellent Surface Hardness
- Light Weight
- Outstanding Transmittance & Optical Clarity
- Optical Design Flexibility and Control



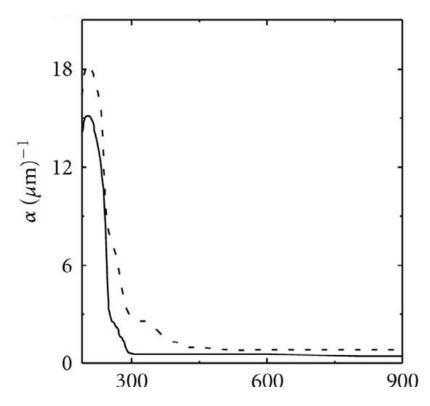
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UV Stability



The outstanding stability of PMMA to sunlight results primarily from weak absorption of acrylic in the UV-region.

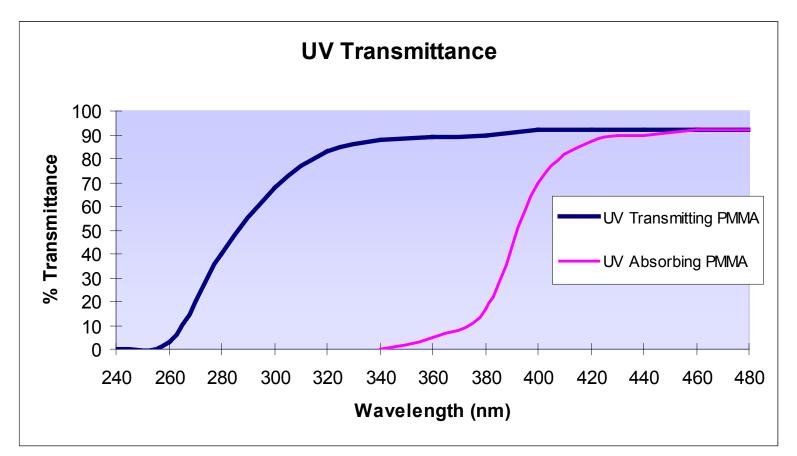


R.M. Ahmed, International Journal of Photoenergy, Volume 2009 (2009), Article ID 150389, "Research ArticleOptical Study on Poly(methyl methacrylate)/Poly(vinyl acetate) Blends

UV Stability



Acrylic is naturally UV transmitting, and can be modified with UV protection systems.





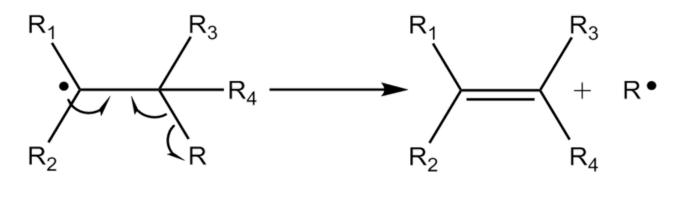
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Chain scission by exposure to extreme temperature or radiation conditions produces alkyl radicals that can react with oxygen to form peroxy radicals, chain termination can lead to creation of aldehydes, ketones or alcohols. Activation energy is ~260 kcal/mole at 130,000 mw. Thermal degradation of polymeric materials, Krzysztof Pielichowski, James Njuguna, 2005, pg 101

Alkoxy radicals may abstract H or undergo beta-scission. (unlikely).

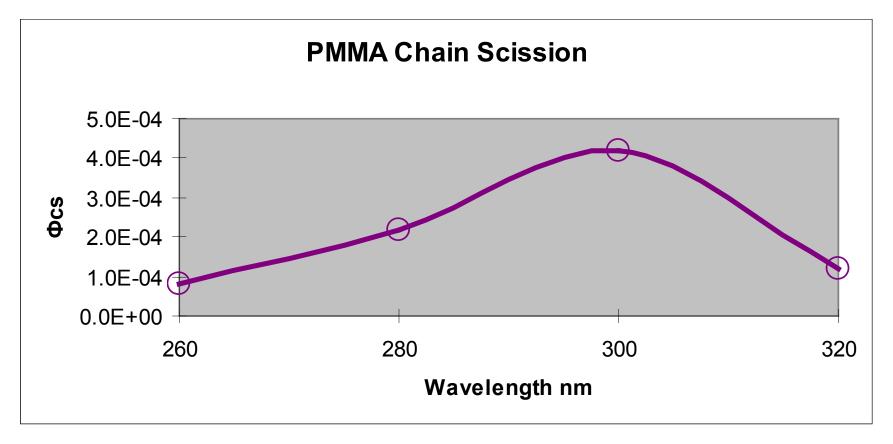


Beta scission mechanism

Degradation Mechanisms Main Chain Scission - PMMA



The quantum yield for main-chain scission of polymethylmethacrylate shows a maximum at 300 nm.



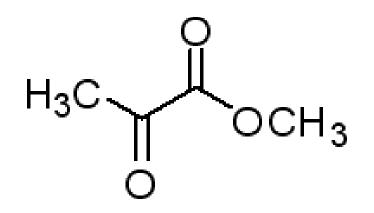
Mitsuoka, T.; Torikai, A.; Fueke, K. J. Appl. Pol. Sci 1993, 47, 1027

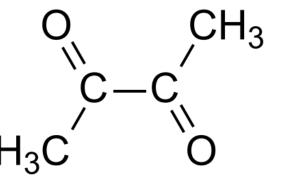


- Initiation of PMMA photo-oxidation is attributed to the decomposition of the ester groups by radiation in the range of 300-330 nm.
- In the presence of oxygen, radical-initiated oxidation occurs with the onset of photolysis.
- Another initiation scenario is the sensitizer effect of MMA-monomer: A photo-excited residual monomer transfers its energy to oxygen yielding singlet O₂ which reacts with monomer to form a hydroperoxide.



In pyrolysis, polymethylmethacrylate can degrade via oxidation of byproducts into methyl pyruvate or diacetyl.¹ Oxidation of residual MMA can produce these products as well. These materials are chromophores that can cause yellowing.





Methyl pyruvate

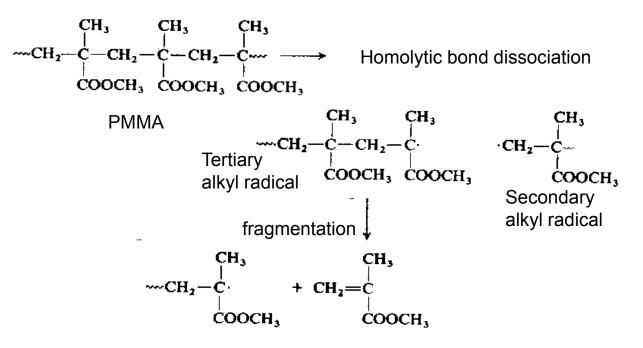
Diacetyl (2,3-butanedione)

1. Mukundan, Thekkekara and Kishore, Kaushal (1987) *Structure and pyrolysis of poly(methyl methacrylate peroxide): a thermochemical approach.* In: Macromolecules, 20 (10). pp. 2382-2385.

Thermal Effects



- Degradation occurs in processing above 300°C.
- Oxidation and chain scission can be accelerated at temperatures above 90°C in the presence of intense UV radiation in the range of 300-330nm.





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Protection Systems



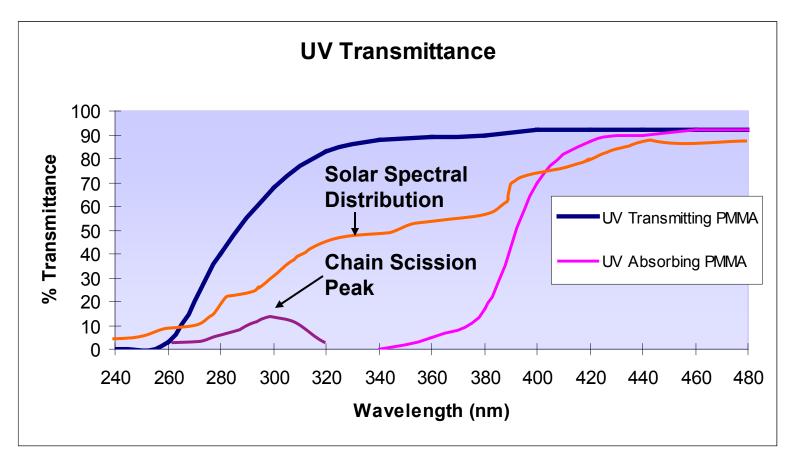
UV Absorbers

STABILIZERS

Combination Systems



UV Absorbing PMMA blocks UV light in the region at which chain scission peaks.

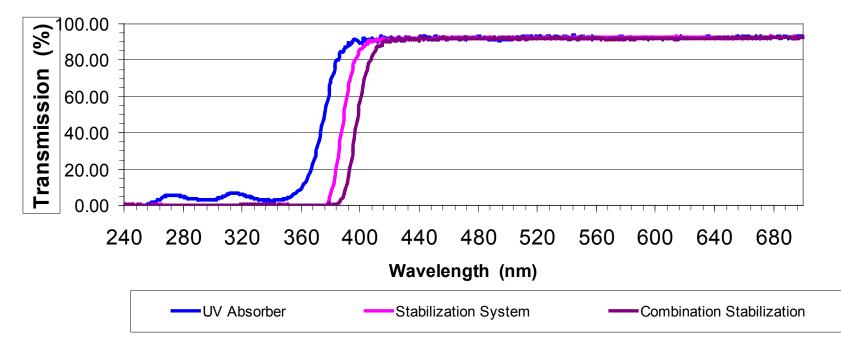


Protection Systems



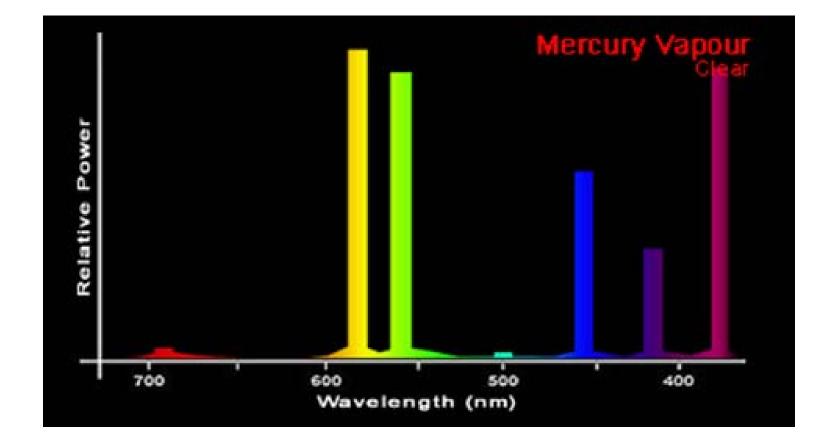
UV_Vis Transmission Curve

Polymer Products Technical Laboratory



High Pressure Hg Vapor Lamp

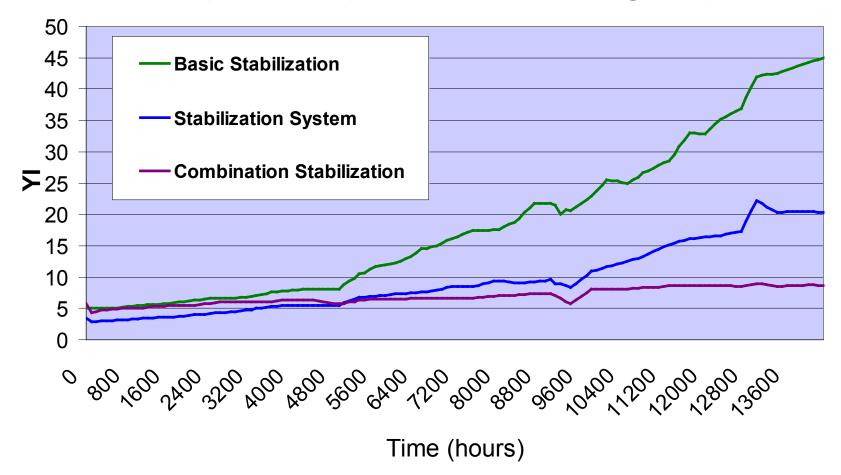




High Intensity UV and Heat Exposure



400W High Pressure Hg Vapor Lamp UV Exposure @ 90 Degrees C





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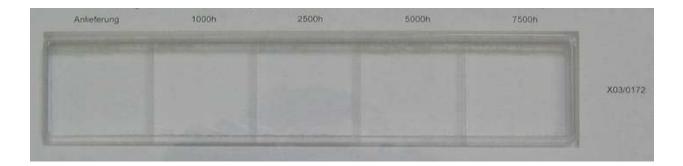
sun radiation (UV + vis + IR)

- + water (moisture)
- + mechanical stress (from diffusion, temperature, ...)
- + temperature changes
- + microorganisms (algae, fungi, ...)
- + time (1a, 2a, 3a, 5a, 10a and more)

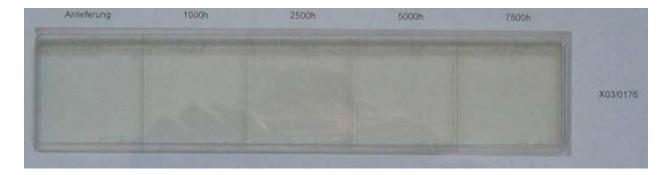


7500 hrs Xenon Arc Weathering

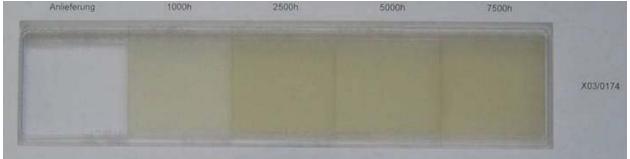




ACRYLITE[®]



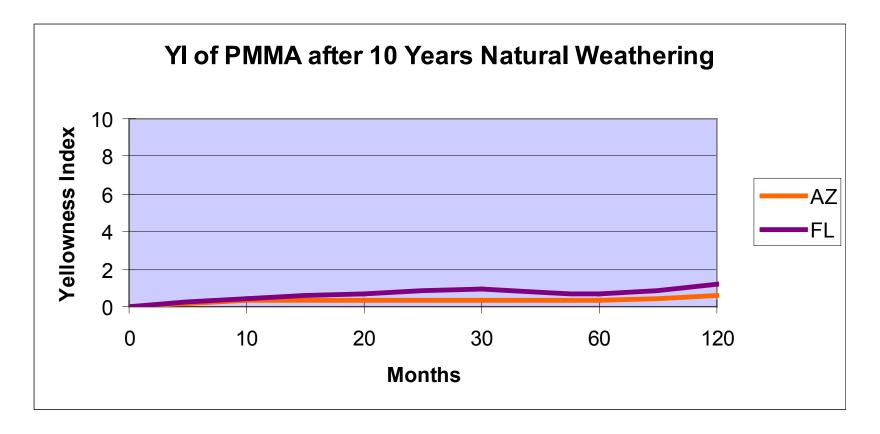




PC

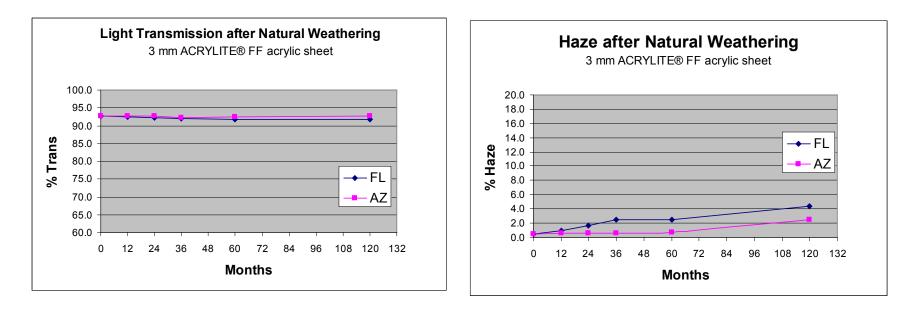


- Virtually no yellowing of acrylic in 10-year outdoor weathering in AZ and FL
- YI = 1.0 is nearly imperceptible to the eye when viewed through the material



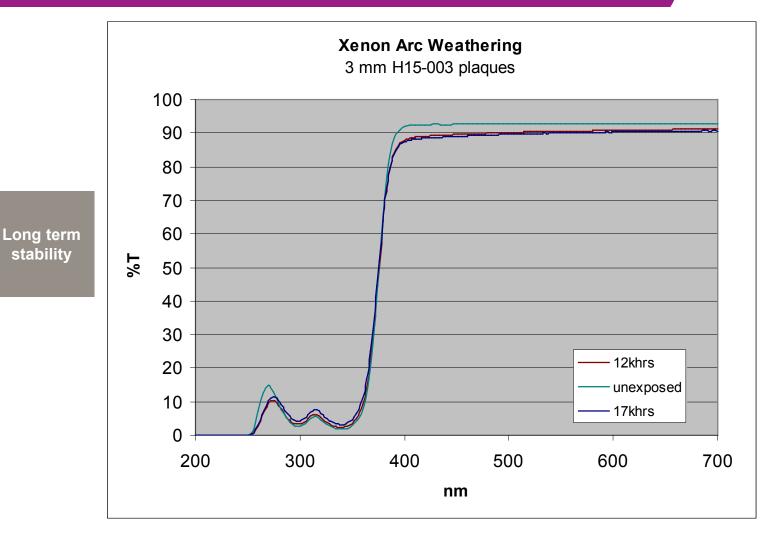


- Overall light transmission remains high and haze remains low after 10 years in outdoor weathering environments
- Florida environment has much more moisture and tends to increase haze somewhat



Xenon Arc Weathering Results





1khr = 1 year; "17-year" Xenon Arc Weathering %T = 88.1% @ 410 nm

Probability of Failure



Failure Mode	Mechanism	Probability
Yellowing from UV Exposure	Oxidation of residual Monomer	High if monomer is present in sufficient quantities
Weakness from UV Exposure	Chain Scission	Low – Insufficient UV Exposure at Earth's Surface
Yellowing from processing	Chain Scission and oxidation of low molecular weight fractions from processing at temperatures above 300°C	Low – Depends on process conditions
Yellowing from UV exposure at High temperatures	Chain Scission and oxidation of low Mw fractions with intense UV radiation and heat	Depends on material, design and ambient temperature conditions
Mechanical (cracks)	High stress level and stress concentration point	Depends on application design and material preparation.



- Application design relative to material properties is important.
- High Residual MMA can cause yellowness through oxidation that creates chromaphores.
- Chain scission occurs at very low levels, if at all, creating minor low molecular weight fractions.
- Protective formulation systems can be used to minimize oxidative, thermal, and UV influences on molecular structure.
- All polymethyl methacrylate materials are not created equal.

