

Approaches to Barrier Coatings for the Prevention of Water Vapor Ingress



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Motivation for Thin Film Barrier Development



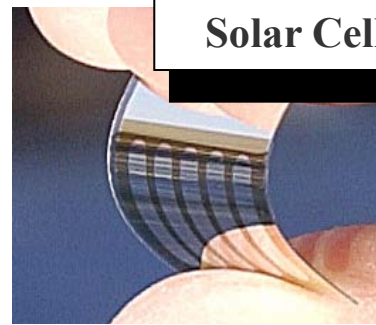
**Solid State
Lighting**



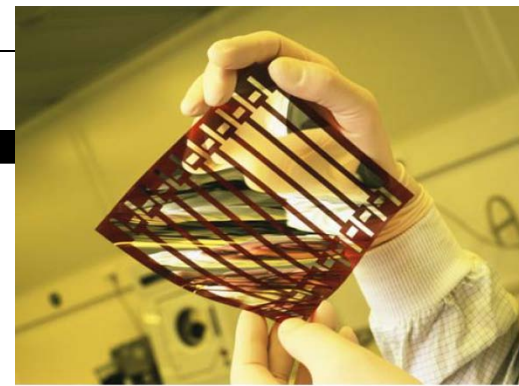
Displays



**Flexible
Transistors**



Solar Cells



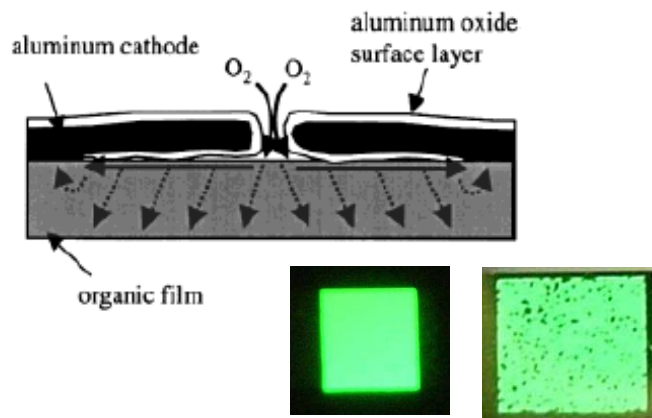
Reliability issues: device encapsulation

Need for Barrier Layers

- Highly reactive electrodes and active layers are very sensitive to water vapor and oxygen.
- Advancements in materials can reduce sensitivity, but not eliminate environmental degradation.

Must address:

- ⇒ **Development of high barrier performance films.**
- ⇒ **Process compatibility with device.**
- ⇒ **Extending technology to large areas and devices with topography.**



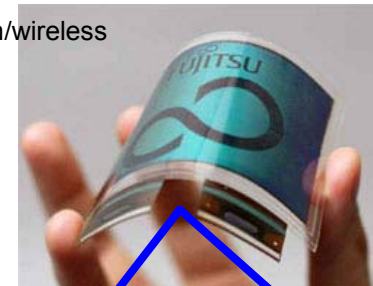
Mechanical Concerns

- Inorganic layers found in encapsulation are generally very brittle and may crack during bending.
- Internal stresses from processing can impact the reliability of the encapsulation.

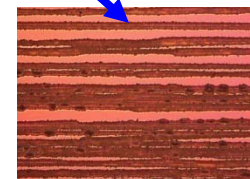
Must address:

- ⇒ **Mechanically robust barrier layers.**
- ⇒ **Adhesion and stress management.**

<http://wirelessmedia.ign.com/wireless/image/article>

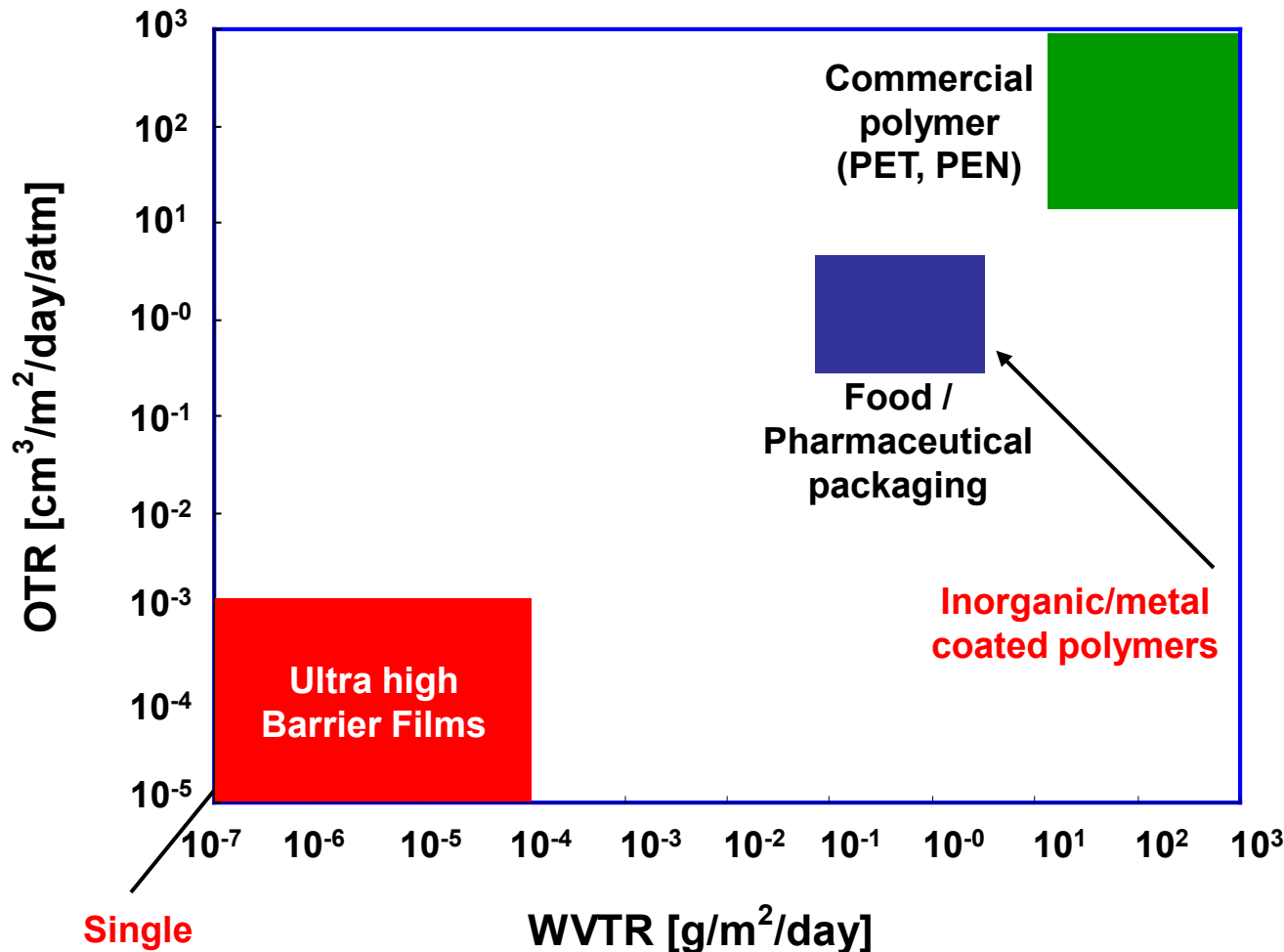


1 min

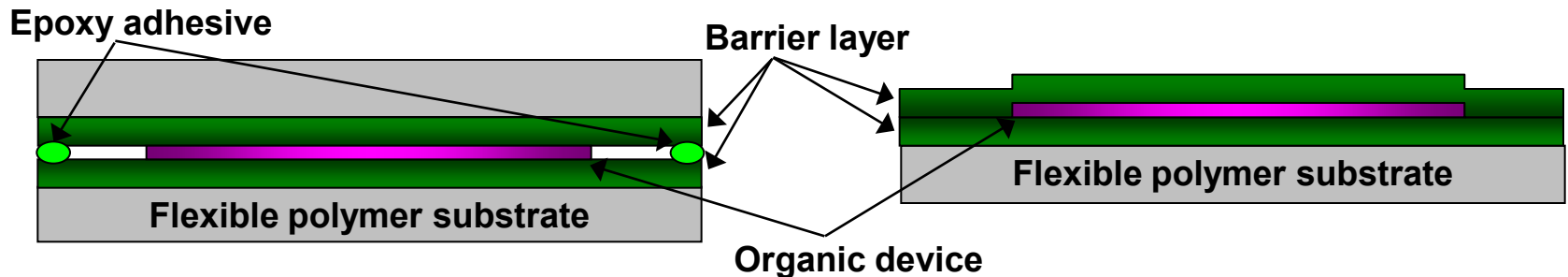
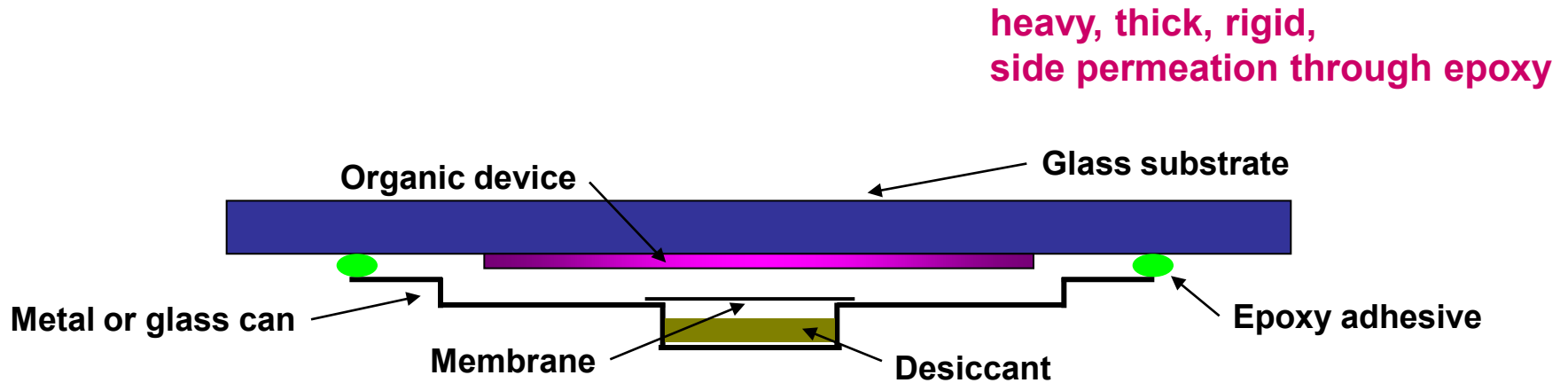


1 hr

Encapsulation Performance Needs



Encapsulation Structures



+ : flexible, light

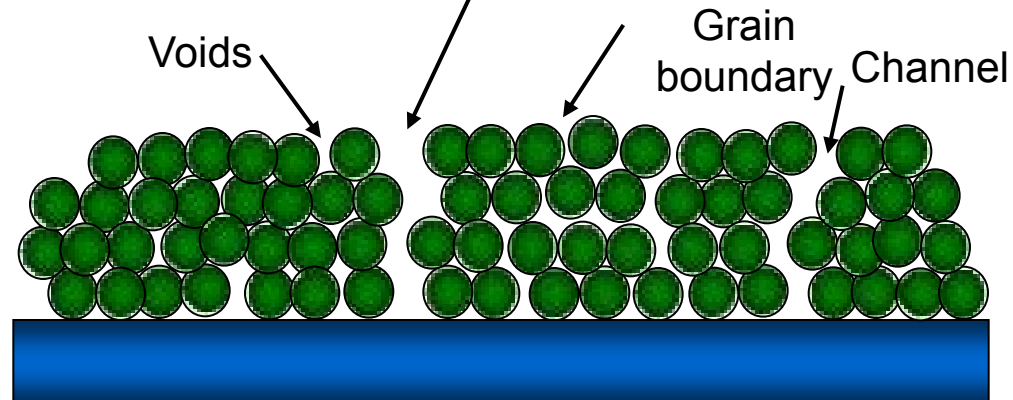
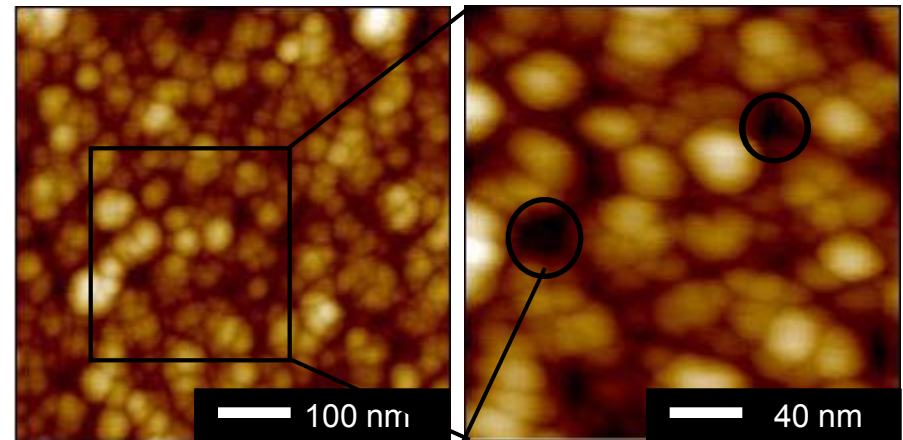
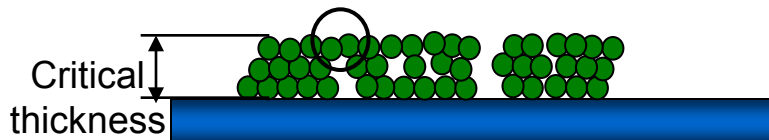
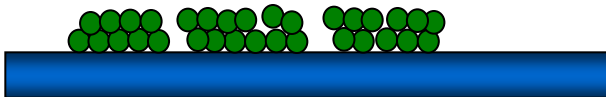
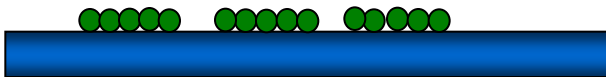
- : need to fabricate barrier layer on
both substrates, side permeation

+ : thin, very flexible, light,
no side permeation

- : need to fabricate barrier layer

Defects in Barrier Films

Single Layer Thin Films

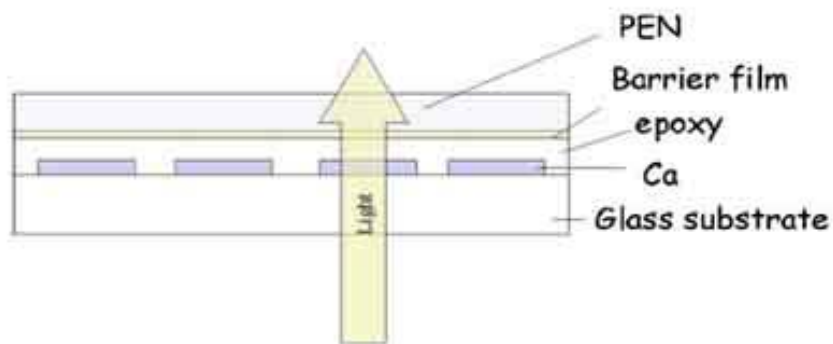


Thin Film Encapsulation Methods

High Quality Single Layer Encapsulation: Al_2O_3

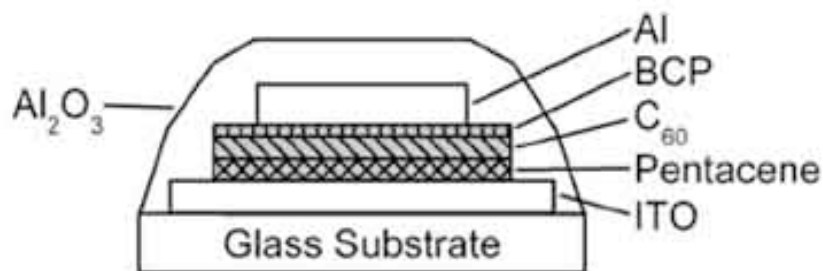
High density, pinhole free, conformal deposition

Permeation governed by nanoscale defects vs macrodefects.

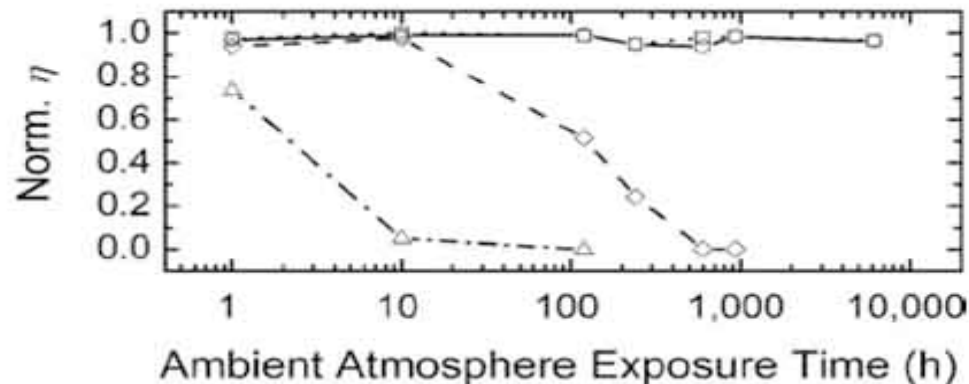


Structure	PEN/ Al_2O_3
Deposition	ALD
WVTR [g/m ² /day]	1.7×10^{-5}
Test condition	38°C and 85% R.H.

P. F. Carcia, R. S. McLean, M. H. Reilly, M. D. Groner, S. M. George, *Applied Physics Letters* **2006**, 89, 031915.



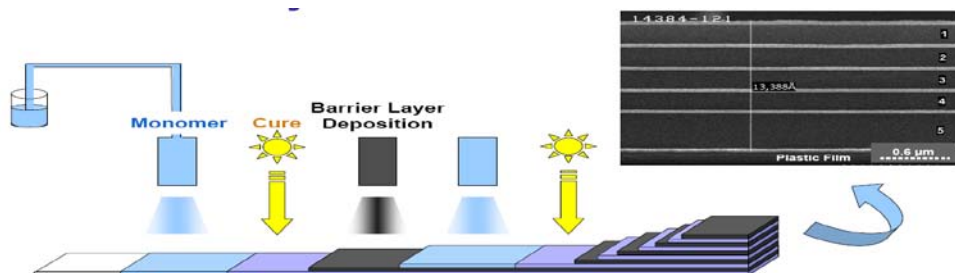
200 nm Al_2O_3 by ALD



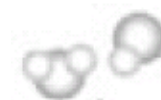
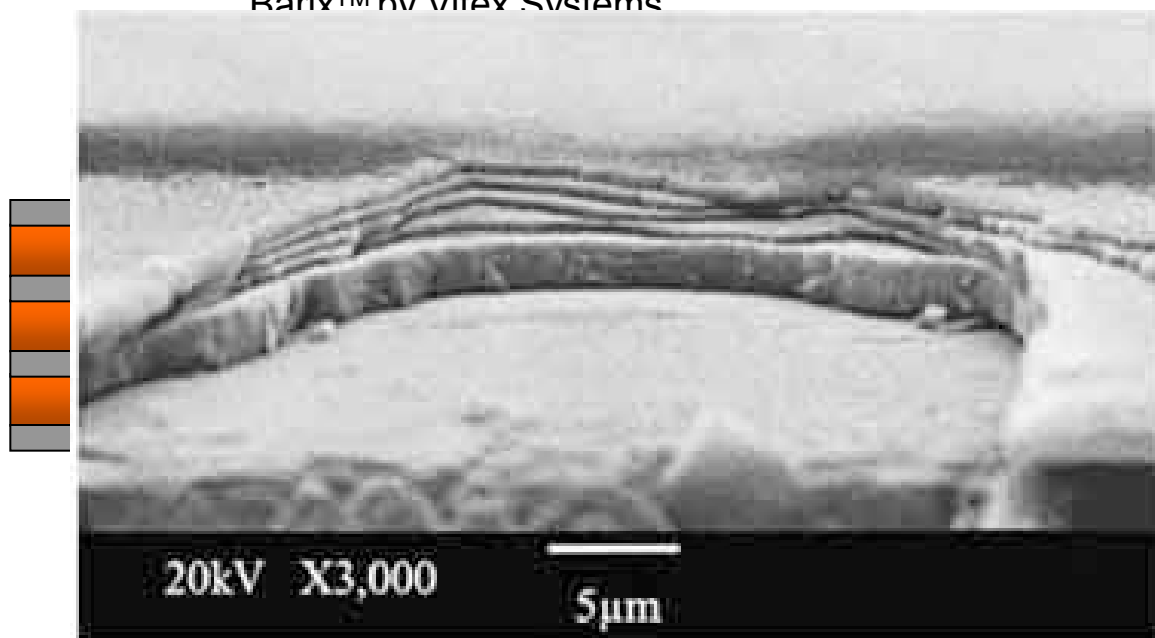
W. J. Potscavage, S. Yoo, B. Domercq, B. Kippelen, *Applied Physics Letters* **2007**, 90, 253511

Thin Film Encapsulation Methods

Multilayer Encapsulation



Rarix™ by Vitex Systems



Water



Organic

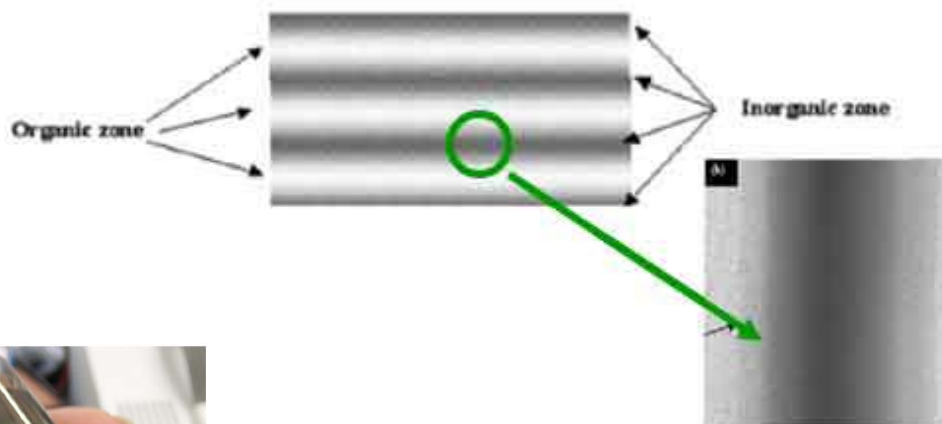


Inorganic

Structure	Al ₂ O ₃ / Polyacrylate
Deposition	DC Sputtering, Evaporation/ UV curing
WVTR [g/m ² /day]	Maximum: 2.1x10 ⁻⁶
Test condition	20°C, 50% R.H.

Thin Film Encapsulation Methods

Graded organic and inorganic layer (GE, Shaepkens, et al., JVST 2004)

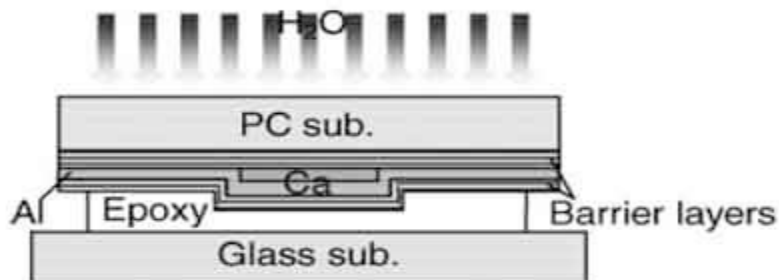


Structure	SiOxNy/ SiOxCy
Deposition	PECVD
WVTR [g/m ² /day]	5x10 ⁻⁵ ~ 5x10 ⁻⁶
Test condition	23°C, 50% R.H. for 20 days



IMRE, Singapore

Multi layer Plus Bonding (Chen, et al., Plasma Process and Polymers 2007)



Structure	3 pairs SiOx/SiNx/ + Parylene + 3 pairs SiOx/ SiNx
Deposition	PECVD· PVD
WVTR [g/m ² /day]	Maximum: 2.5x10 ⁻⁷
Test condition	23°C and 40% R.H. for 75 day

Processing of Barrier Films

Materials Used



PECVD: SiO_x, SiN_x



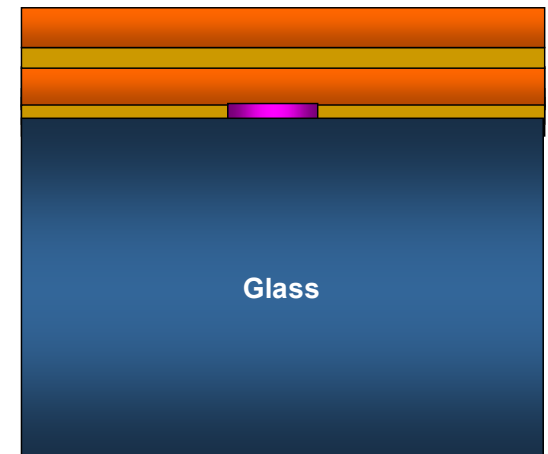
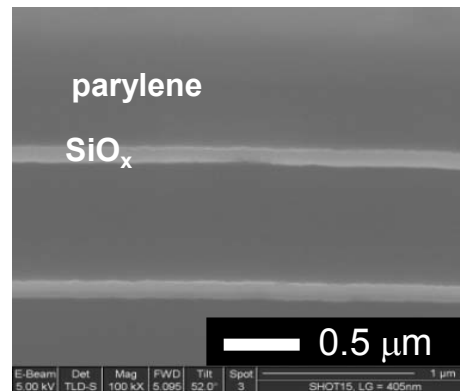
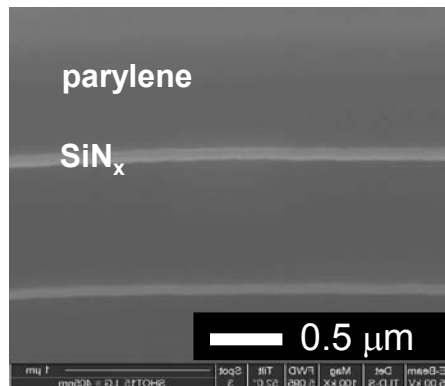
PVD: Parylene



ALD: Al_2O_3

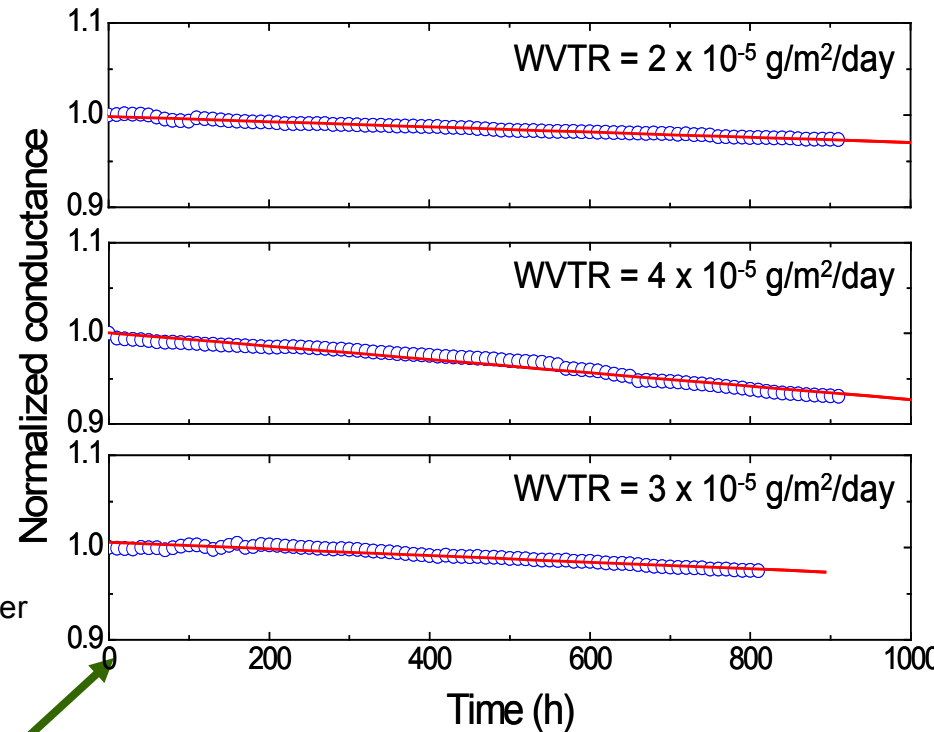
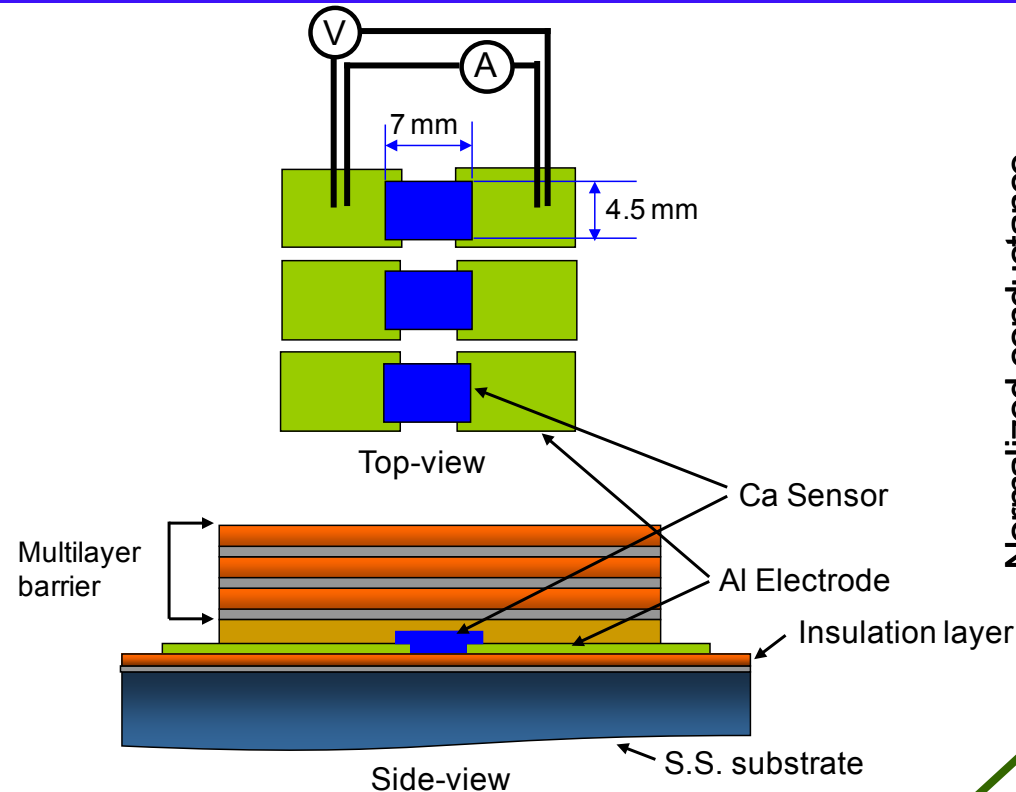


Environmental Chamber



Measured by Ca Corrosion Method at 50 % R.H. and 20 °C

Ca Corrosion Tests



$$WVTR[g / m^2 / day] = 2\delta_{Ca} \times \rho_{Ca} \times \frac{dG_s}{dt} \times \frac{M(H_2O)}{M(Ca)} \times \frac{Ca_Area}{Window_Area}$$

$$\rho_{Ca} \quad 1.55 \text{ g/cm}^3$$

$$\delta_{Ca} \quad 3.4 \times 10^{-6} \text{ cm } \Omega$$

$$G_s = 1/R_s = (W/L) \times (1/R)$$

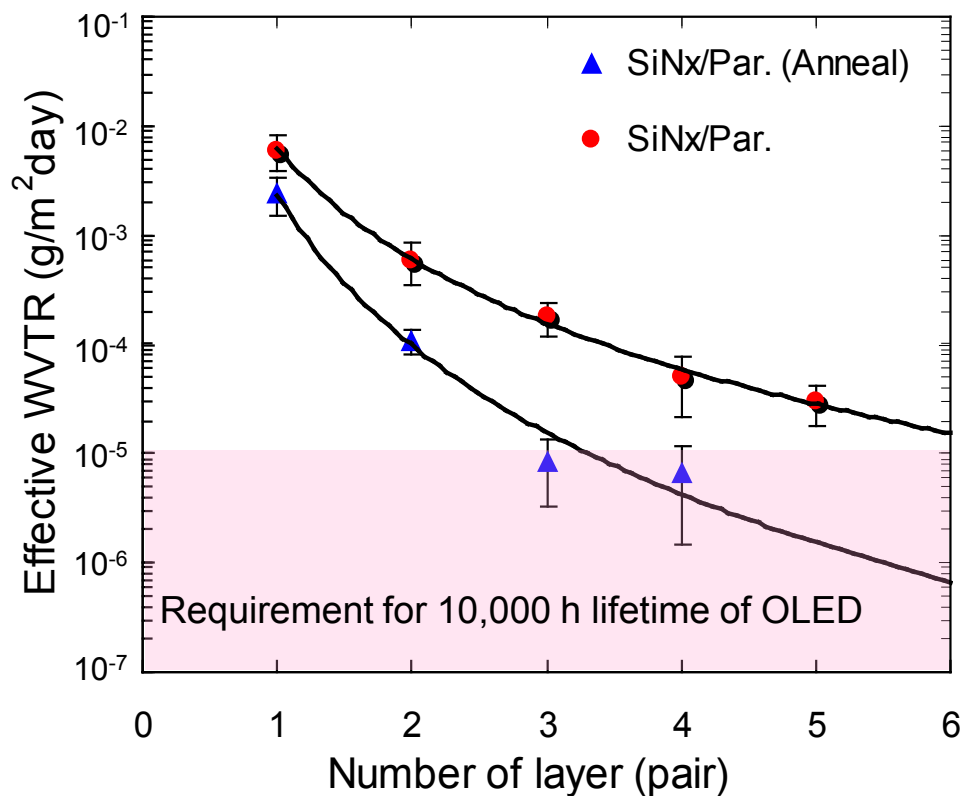
L : Length of Ca

W : Width of Ca

$$M(H_2O) \quad 18 \text{ amu}$$

$$M(Ca) \quad 40.1 \text{ amu}$$

Multilayer Results



No. of Layers [pairs]	WVTR [g/m ² /day]		Decrease in WVTR [%]
	Before annealing	After annealing	
1	4.3 × 10 ⁻³	2.4 × 10 ⁻³	44
2	4.4 × 10 ⁻⁴	1.3 × 10 ⁻⁴	70
3	1.3 × 10 ⁻⁴	7.3 × 10 ⁻⁶	94
4	4.4 × 10 ⁻⁵	6.6 × 10 ⁻⁶	85

SiOx/ Parylene (3 pairs)

8.4 × 10⁻⁴ ▶ 6.6 × 10⁻⁵ (g/m²/day), (85 %↓)

Mass Transport in Barrier Films

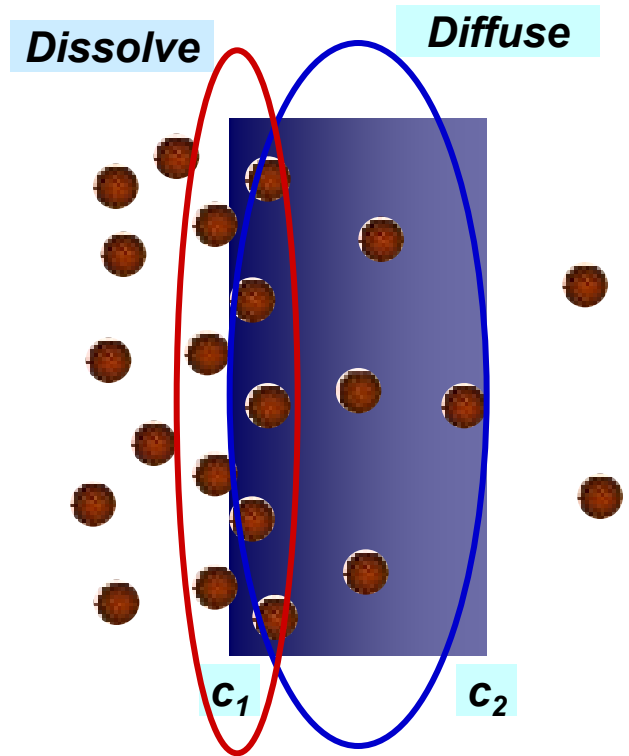
Principles of permeation

$$P = DS$$

P : Permeation coefficient (permeability)

D : Diffusion coefficient, determines how **fast** the permeant can **move** in the media

S : Solubility coefficient, determines how **much** of the permeant can **be dissolved in in the film**



Driving force

$$J = -D\left(\frac{\partial c}{\partial x}\right) \quad \text{Fick's first Law}$$

$$\rightarrow c = Sp \quad \text{Henry's Law}$$

$$J = DS \frac{\Delta p}{l}$$

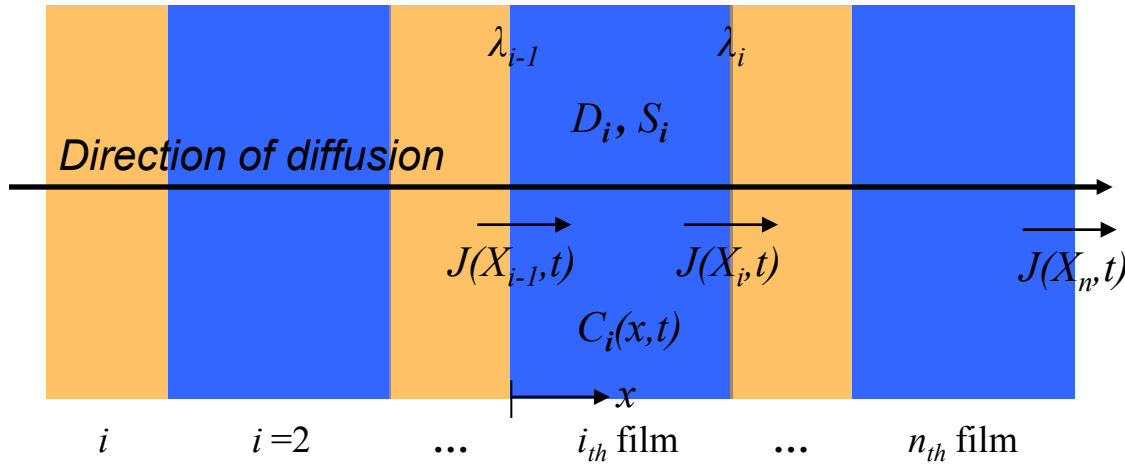
J : Flux of permeant

$\partial c / \partial x$: concentration gradient

p : Partial pressure of permeant

Mass Transport in Barrier Films

Diffusion in multilayer structure



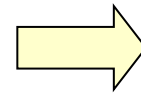
D_i (diffusivity),
 S_i (Solubility),
 L_i (thickness)-fixed
 Defect spacing

WVTR, Lag time

WVTR calculation

$$WVTR \Leftrightarrow \frac{P_{total} \cdot \Delta p}{L_{total}}$$

Δp : pressure drop
 L_{total} : thickness



For inorganic layers, use effective permeability

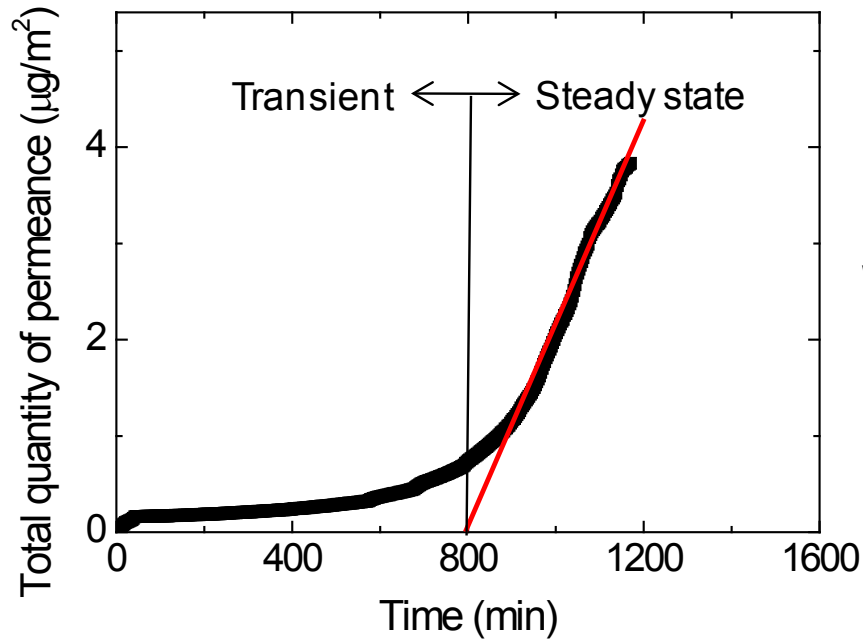
$$\frac{L_{total}}{P_{total}} = \frac{1}{\frac{L_1}{P_1} + \frac{L_2}{P_2} + \frac{L_3}{P_3} + \dots + \frac{L_n}{P_n}}$$

Lag time calculation

$$L = \left(\sum_{i=1}^n \frac{L_i}{D_i} \prod_{j=1}^{i-1} k_j \right)^{-1} \sum_{i=1}^n \left\{ \frac{L_i^2}{2D_i} \sum_{m=1}^n \left[\frac{L_m^2}{D_m} \prod_{j=1}^{m-1} k_j \right] - \frac{L_i^3}{3D_i^2} \prod_{j=1}^{i-1} k_j \right\} + \sum_{i=1}^n \left\{ \frac{L_i}{D_i} \prod_{j=1}^{i-1} k_j \sum_{\beta=i+1}^n \left[\frac{L_\beta}{D_\beta} \sum_{m=\beta}^n \left(\frac{L_m}{D_m} \prod_{j=1}^{m-1} k_j \right) - \frac{L_\beta^2}{2D_\beta} \right] \right\}$$

where, $k_j = \frac{k_j}{k_{i+1}}$

Mass Transport in Barrier Films



**We have seen lag times greater than 1000 hours in our films.
Impacts the WVTR measured.**

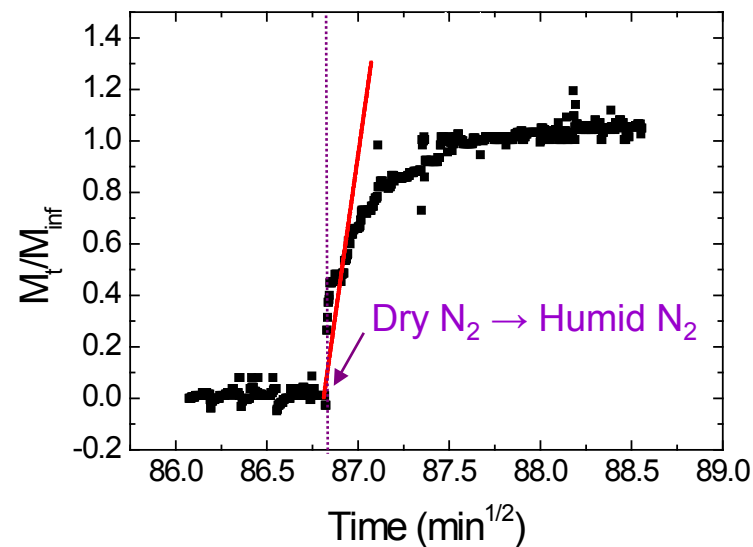
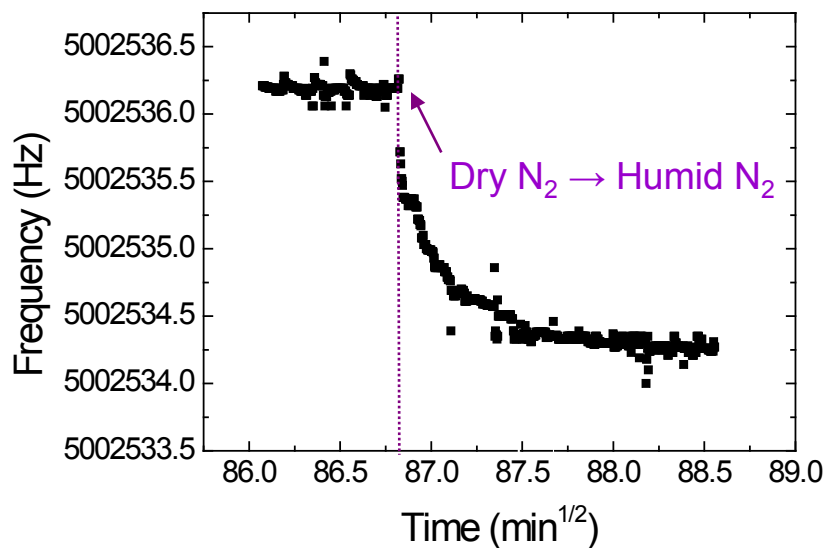
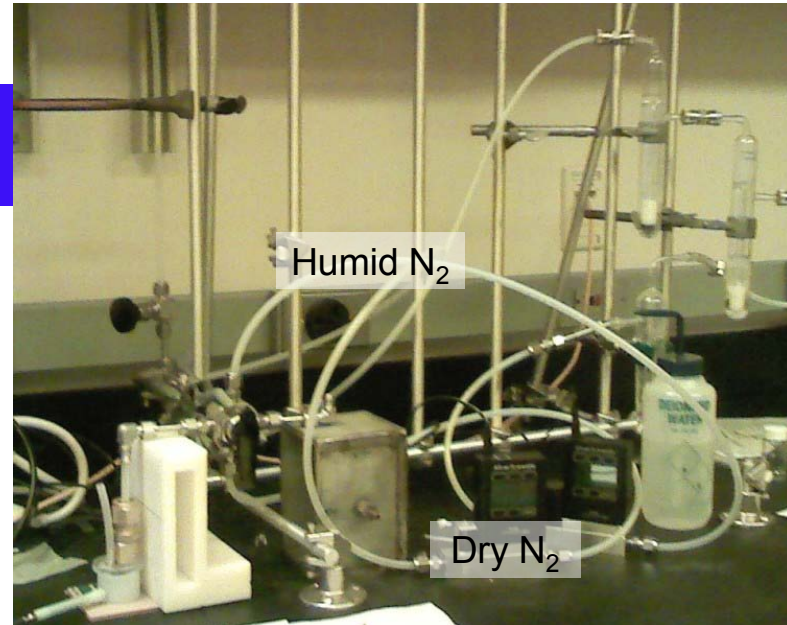
Lag Time: 1300 hours

Transient WVTR: $8 \times 10^{-6} \text{ g}/\text{m}^2/\text{day}$

Steady State: $2 \times 10^{-3} \text{ g}/\text{m}^2/\text{day}$

(Data not obtained from graph on left)

QCM



$$\frac{M_t}{M_\infty} = \frac{2}{L} \left(\frac{Dt}{\pi} \right)^{1/2}$$

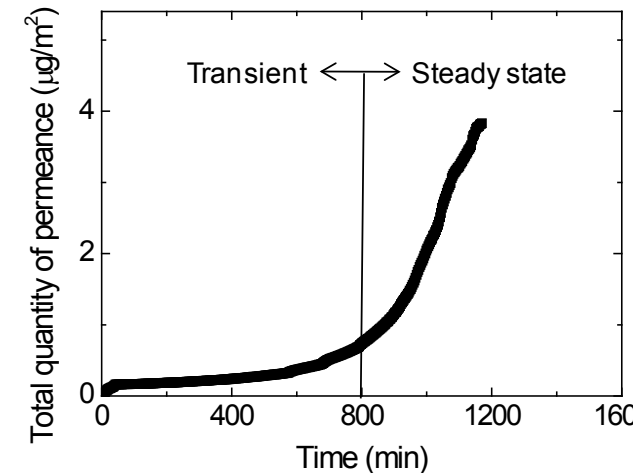
M : Mass uptake
 D : Diffusion coefficient
 L : Thickness of film

Solution to Fick's 2nd law for short times (Valid only $M_t/M_\infty < 0.6$)

Diffusion Coefficient and Solubility

Diffusion Coefficient (cm²/s)

Test #	Before Anneal	After Anneal for 10 min
Average	3.27×10^{-9}	2.06×10^{-9}
STDV	4.29×10^{-9}	1.01×10^{-10}



Solubility coefficient measurement

$$S = \frac{M_{\infty}}{M_0} \frac{\rho_{polymer}}{MW_{penetrant}} \times \frac{22,414}{p}$$

M_{∞} : Equilibrium mass uptake
 M_0 : Mass of the water-free polymer
 $MW_{penetrant}$: molecular weight (g/mole)
 p : Vapor pressure [atm]
 22,241 : Conversion from moles to cm³(STP)

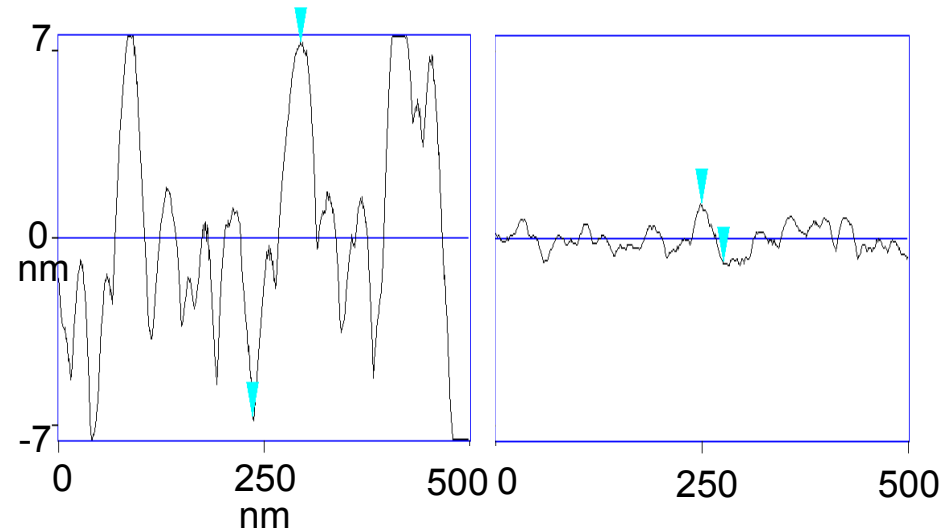
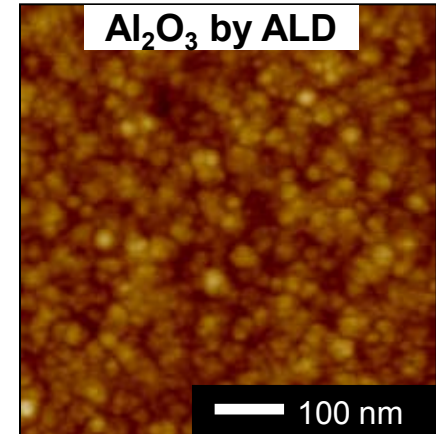
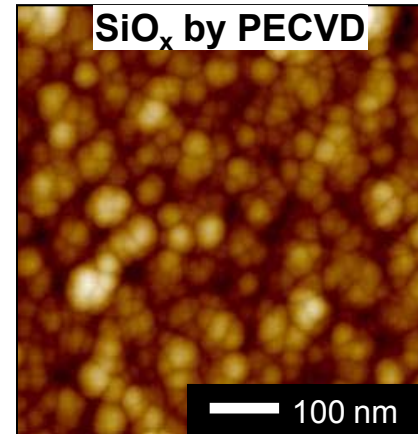
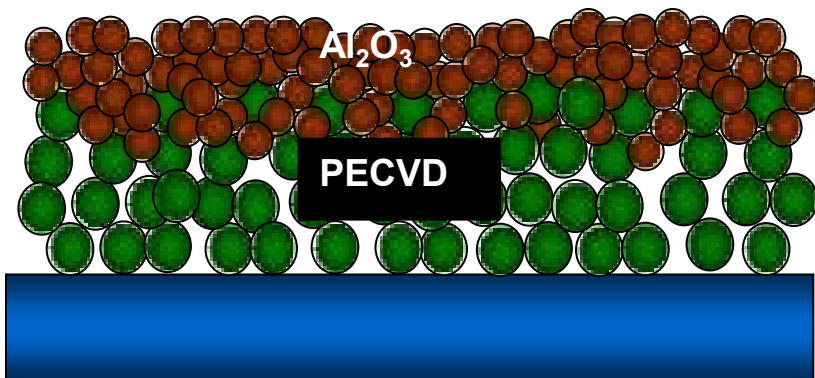
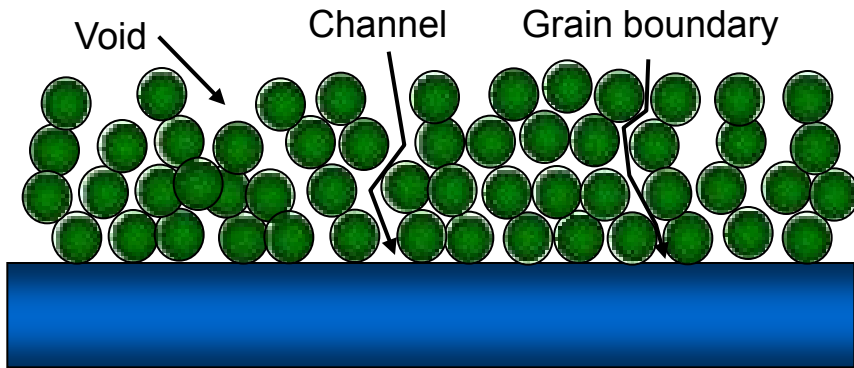
Solubility (g/cm³atm)

Test #	Before Anneal	After Anneal for 10 min
Average	0.286	0.140
STDV	0.054	0.044

Initial water content in sample impacts the lag time and WVTR observed unless measured for a long time.

New Approach: Hybrid Architecture

Combine rapid low temperature deposition by PECVD with high quality atomic layer deposition to simplify barrier architecture.



 SiO_x by PECVD

 Al₂O₃ by ALD

Comparison of Results

Multilayer Film WVTR (g/m²/day)

3 dyads of SiO _x /Parylene	$6 \pm 2 \times 10^{-5}$
3 dyads of SiN _x /Parylene	$7 \pm 2 \times 10^{-6}$

Hybrid Architecture WVTR (g/m²/day)*

SiO _x /Al ₂ O ₃ /Parylene	$2 \pm 1 \times 10^{-5}$
SiN _x /Al ₂ O ₃ /Parylene	$3 \pm 2 \times 10^{-5}$

***Films contain 50 nm of Al₂O₃**

****Al₂O₃ layer 3×10^{-4} g/m²/day**

Reducing ALD thickness: minimal impact!!!

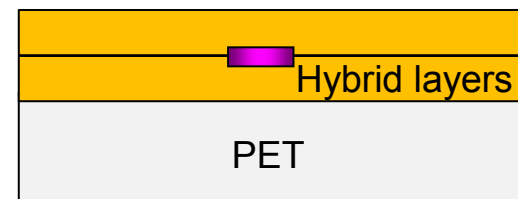
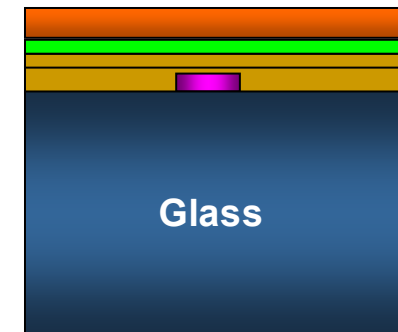
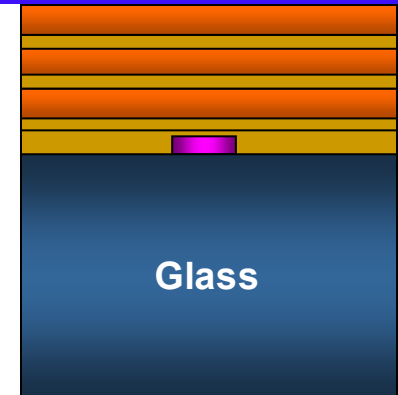
Al₂O₃ thickness : 10 nm

SiO_x/Al₂O₃/Parylene : **$4 \pm 0.5 \times 10^{-5}$ g/m²/day**

Coated PET Substrate

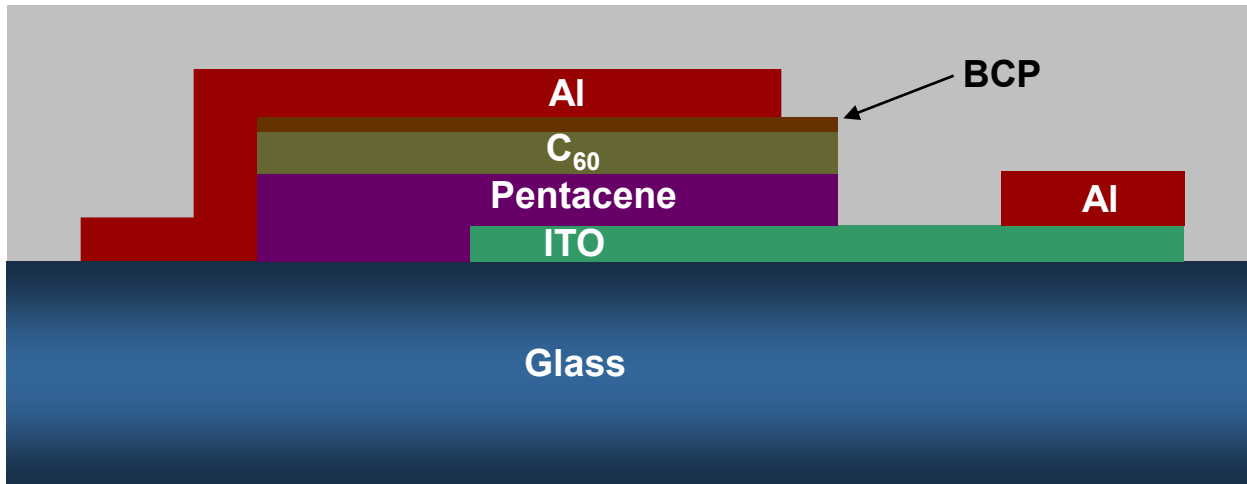
Al₂O₃ thickness : 50 nm

SiO_x/Al₂O₃/Parylene : **$2.5 \pm 1.5 \times 10^{-5}$ g/m²/day**

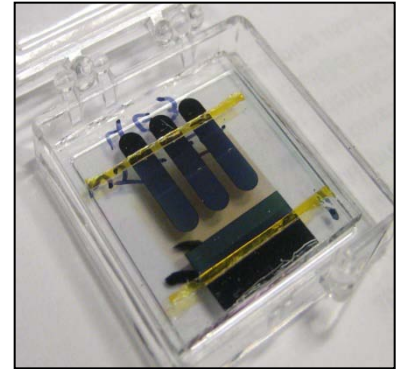


Integration with OPVs

Device fabrication and encapsulation



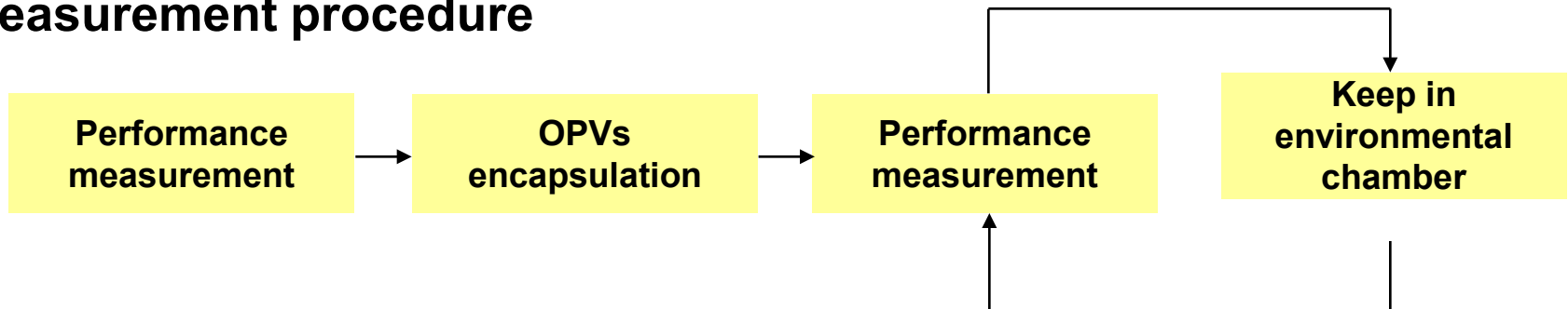
Encapsulation
(1~4 pairs of
SiNx/ Parylene)



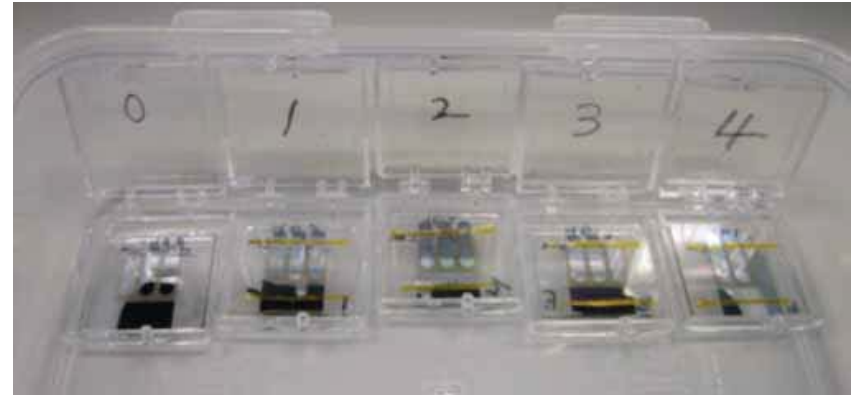
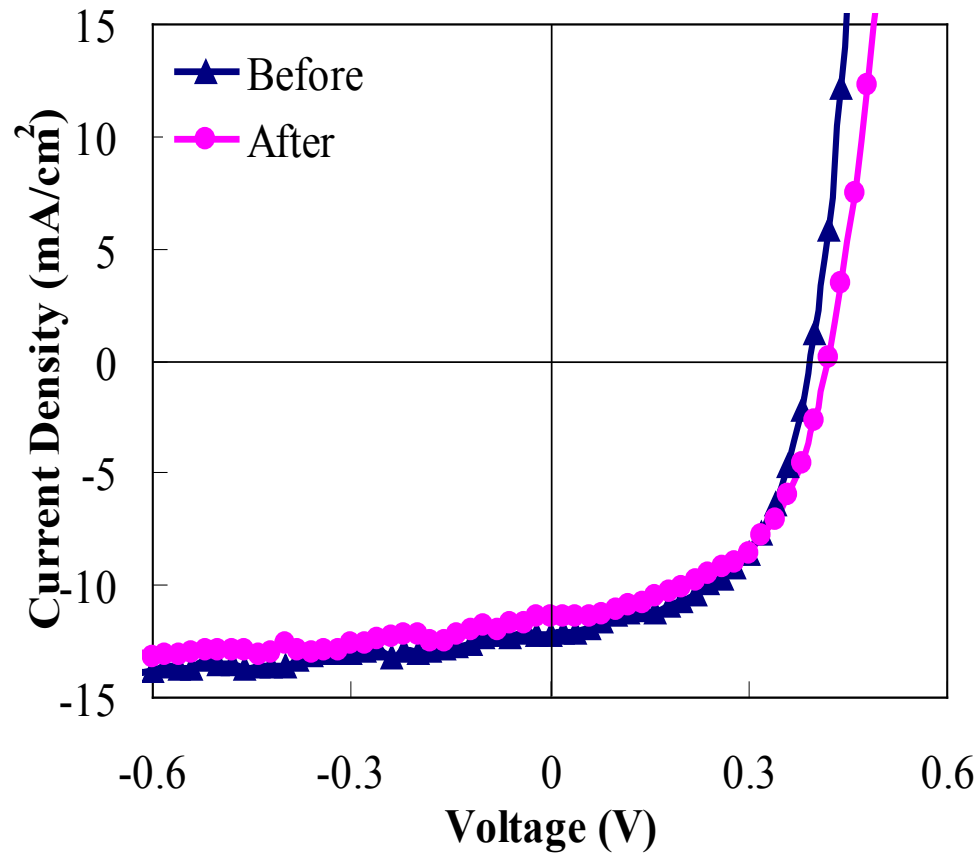
ITO/ Pentacene (50 nm)/ C_{60} (45nm)/ BCP (8 nm)/ Al

S. Yoo, et., Al., Appl. Phy. Lett., 2004, 85, 5427

Measurement procedure



Process Impact on OPVs

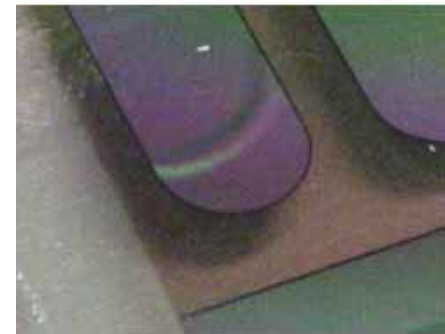
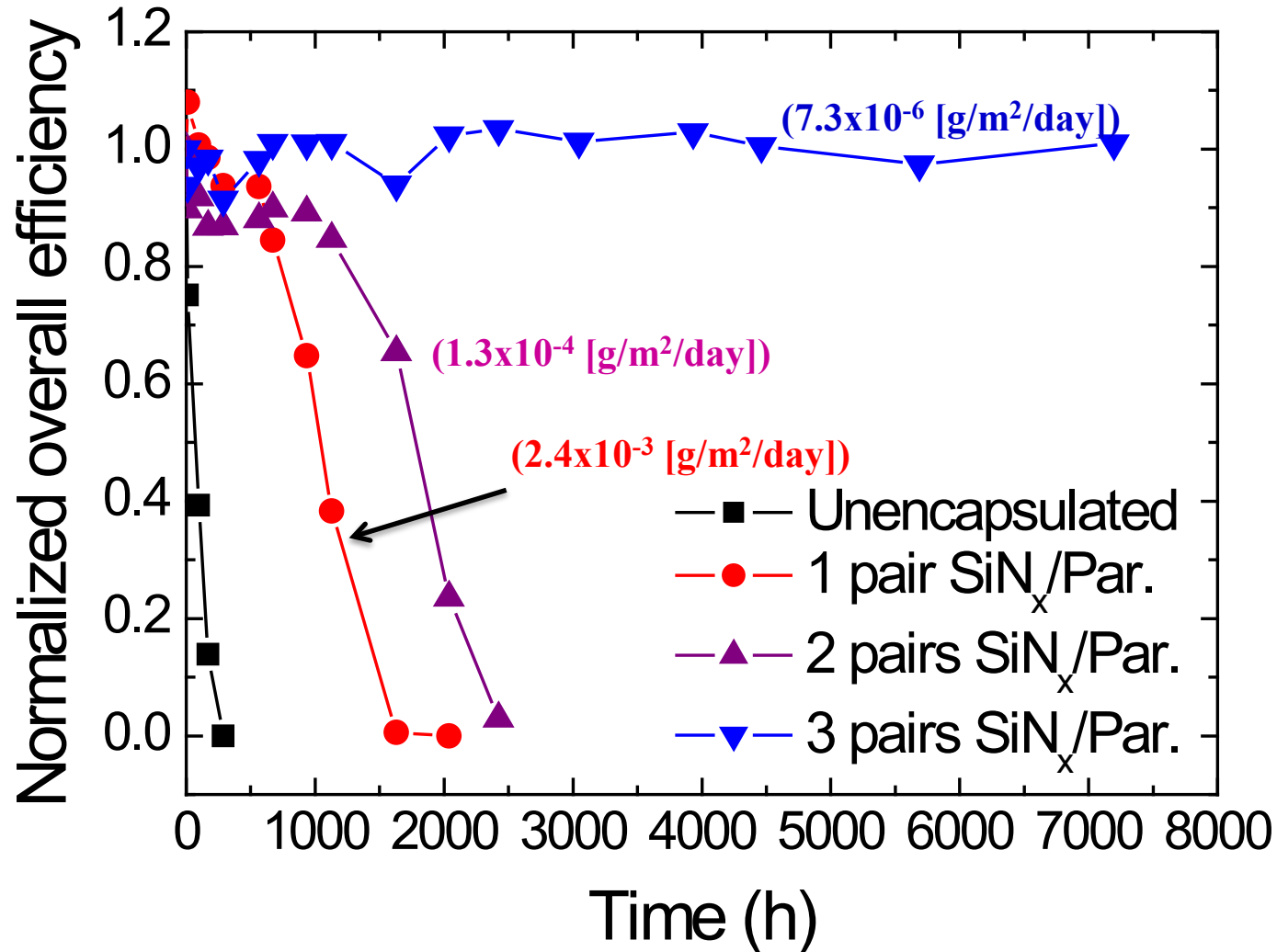


	V_{oc} (V)	J_{sc} (mA/cm ²)	FF	η (%)	Average (%)
Before	0.39	-12.19	0.54	3.3	3.4
After	0.41	-11.40	0.54	3.3	3.3

- Average is based on 12 devices
- Light source : 175W Xenon Lamp

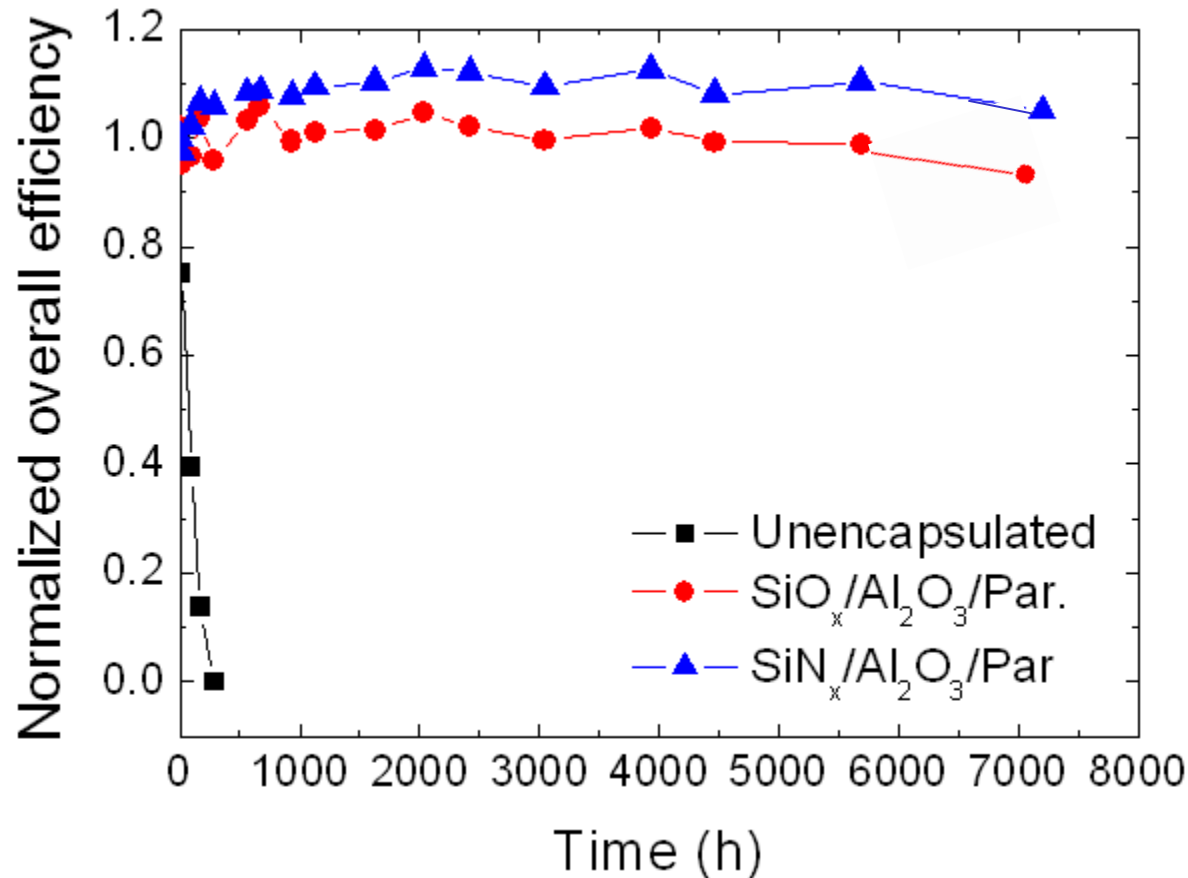
Device Performance Vs. Time

Overall device efficiency (Oriel 91160, AM 1.5G)



Hybrid Encapsulation Architecture

Overall device efficiency (Oriel 91160, AM 1.5G)



Excellent performance for simple architecture. (WVTR~ 10^{-5}).

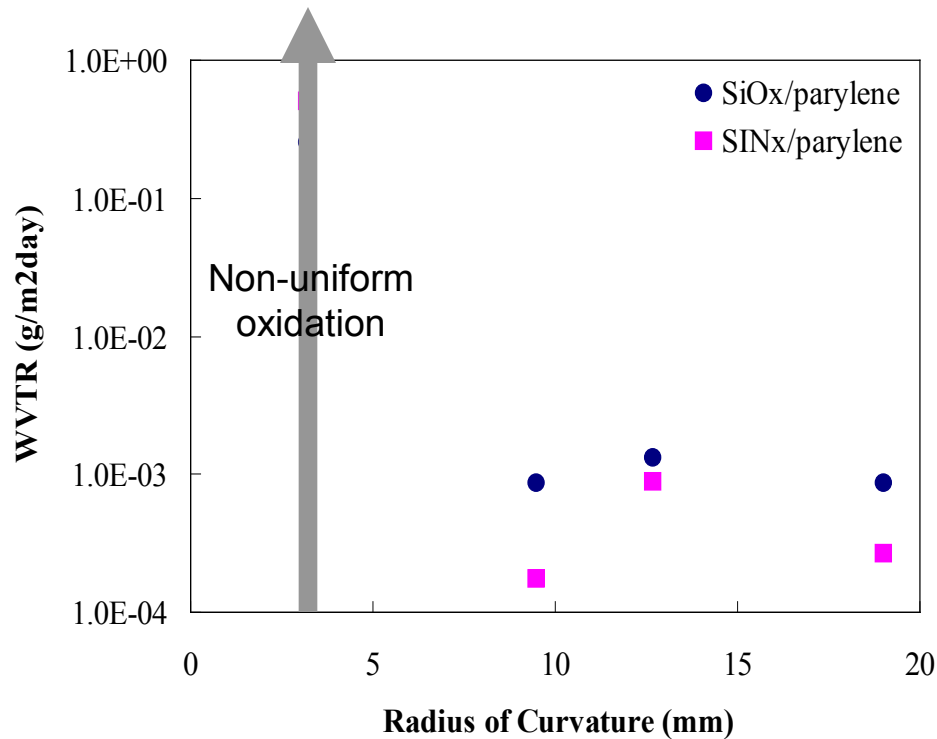
Reduction in deposition time by a factor of 5.

Delamination and buckling were eliminated

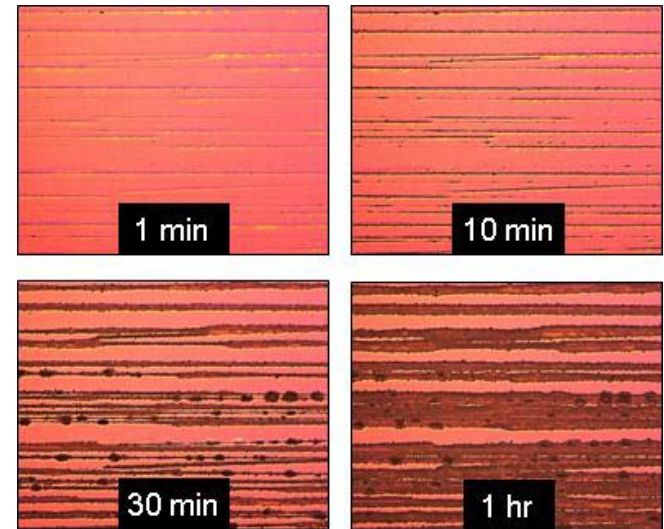
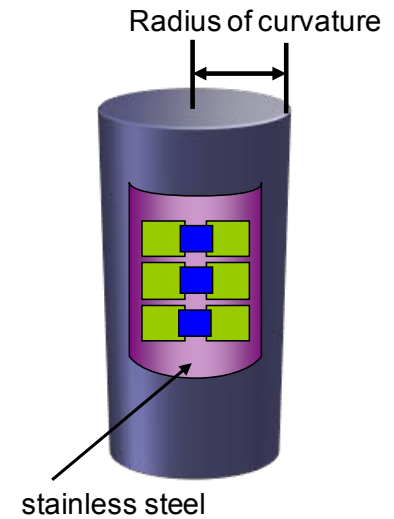
Additional work on reducing processing time by an order of magnitude are underway.

Barrier Performance under Bending

Results



WVTR as the function of the radius of curvature

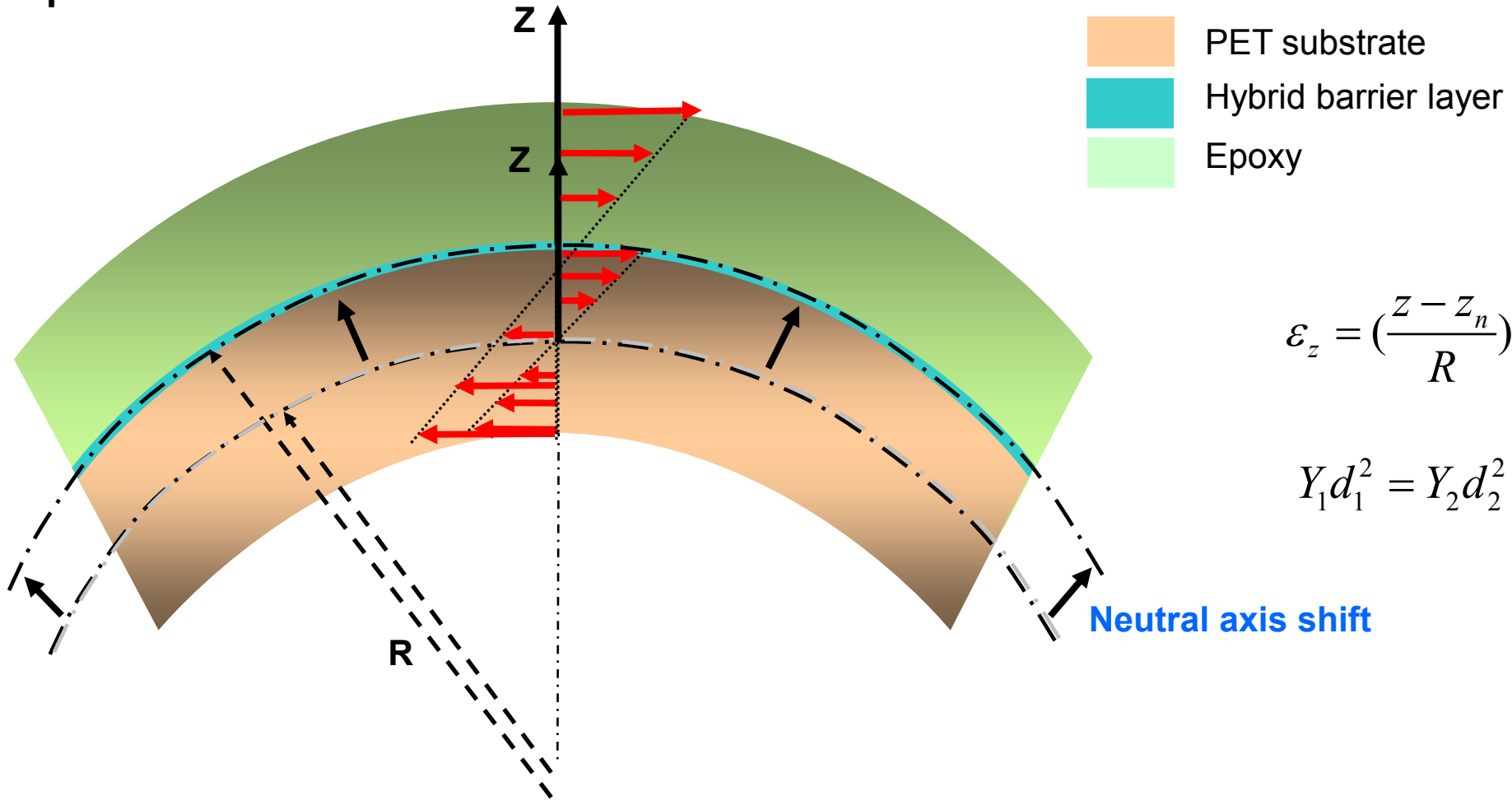


Oxidation through cracks in inorganic layer

Improving Flexibility

Flexibility is limited by the failure strain of the inorganic layer (0.5-2%).

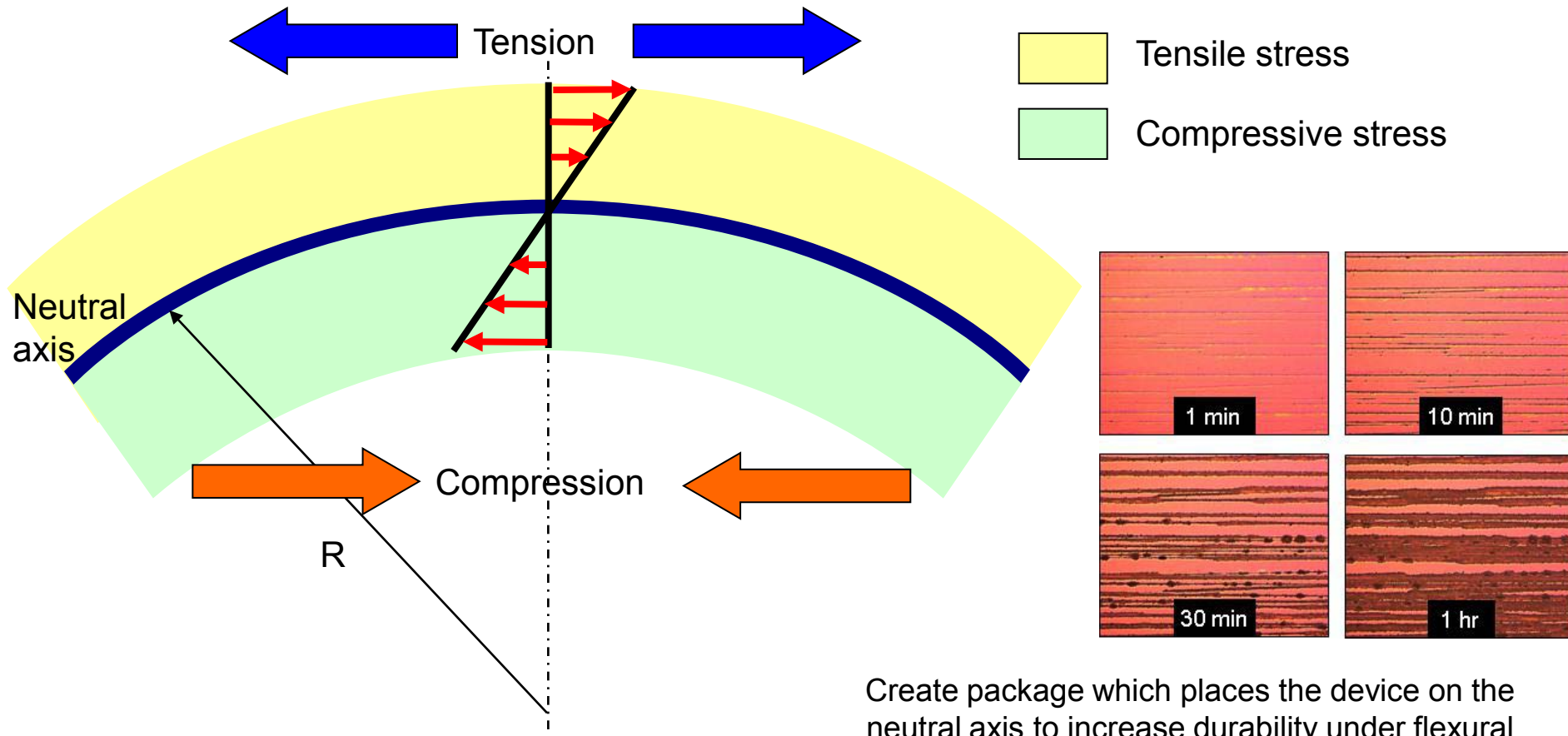
Must improve failure strain without reducing WVTR or reduce strain on the encapsulation.



Improving Flexibility

Flexibility is limited by the failure strain of the inorganic layer (0.5-2%).

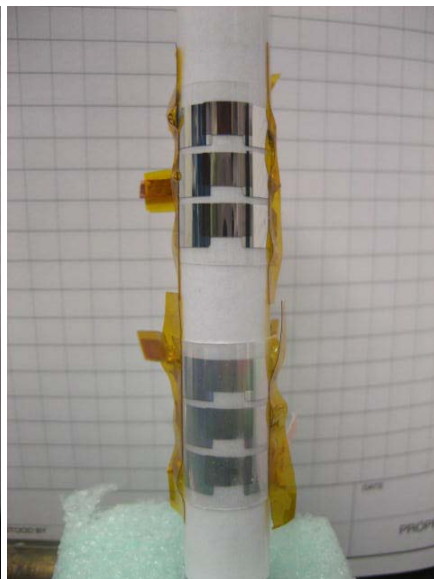
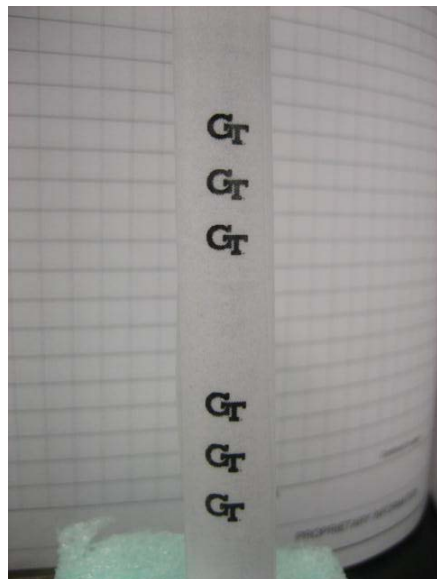
Must improve failure strain without reducing WVTR or reduce strain on the encapsulation.



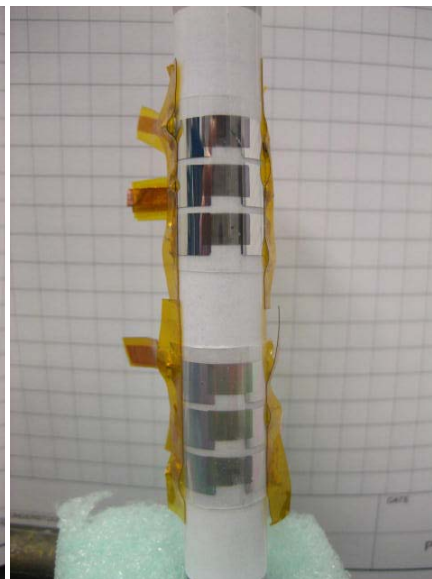
Create package which places the device on the neutral axis to increase durability under flexural deformation.

Improving Flexibility

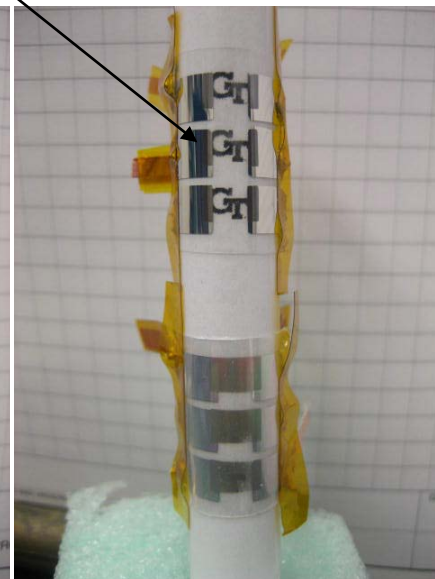
Preliminary results



Bending



10 min



6 -100 hr

GT logo was used for visual qualitative comparison and not WVTR measurement.

Summary

■ **Multilayer Encapsulation**

- Defect structure and solubility of polymer layer control laminate performance
- Reporting WVTR and lag time may be necessary for understanding barrier performance.

■ **Hybrid Encapsulation**

- Simplified Architecture provides ultralow barrier performance
- Promising opportunities exist for further reductions in processing time.

■ **Device Integration**

- Successful direct encapsulation of OPVs by hybrid thin films

■ **Future Efforts**

- Development of Edge Seals important for the packaging toolbox.
- Accelerated testing under harsh environments including light soaking!

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Q & A