

# Impact of ethanol and butanol as oxygenates on SIDI engine efficiency and emissions using steady-state and transient test procedures

Thomas Wallner, Neeraj Shidore, Andrew Ickes Argonne National Laboratory

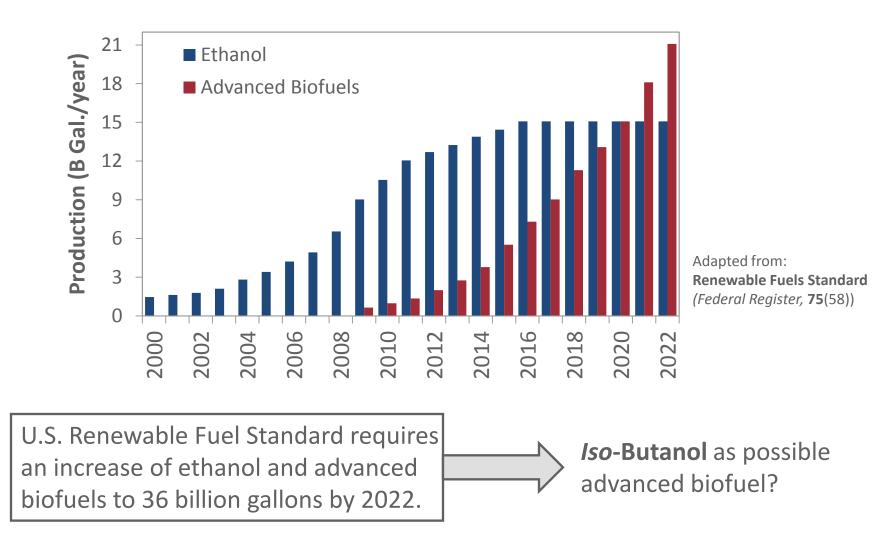
**16<sup>th</sup> Directions in Engine-Efficiency and Emissions Research (DEER) Conference** Detroit, Michigan September 27-30, 2010



## Acknowledgements

- Co-Authors
  - Thomas Wallner
  - Neeraj Shidore
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  - Connie Bezanson
  - Lee Slezak
  - Kevin Stork
  - Gupreet Singh

## Background & Motivation



### **Objectives**

Assess the potential of blending gasoline with several alcohol fuels for use in a gasoline direct injection (DI) spark ignition (SI) engine.

Evaluate the effect of ethanol and butanol addition on regulated and non-regulated emissions compared to gasoline baseline.



Utilize Engine Hardware-In-Loop capability to characterize emissions trends over a simulated test cycle.

## **Fuel Specifications**

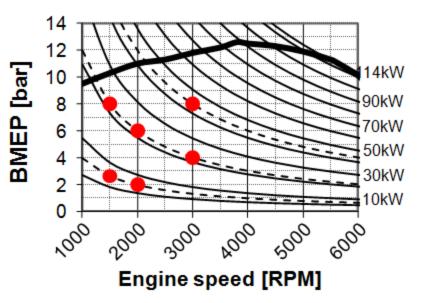
		Gasoline	Ethanol	<i>iso</i> -Butanol
Chemical formula		C <sub>4</sub> - C <sub>12</sub>	C <sub>2</sub> H <sub>5</sub> OH	C <sub>4</sub> H <sub>9</sub> OH
Composition (C, H, O)	Mass-%	86, 14, 0	52, 13, 35	65, 13.5, 21.5
Lower heating value	MJ/kg	42.7	26.8	33.1
Density	kg/m <sup>3</sup>	715 - 765	790	802
Octane number ((R+M)/2)	-	90	100	103
Stoichiometric air/fuel ratio	-	14.7	9.0	11.2
Latent heat of vaporization	kJ/kg	380 – 500	919	686
Ethanol BlendsFuel Oxygen (%-mass)Butanol Blends				
E10 3.7 E50 18	ass conto	iso-But16 iso-But83	<u>Blend lower heating values</u> Gasoline: 42 MJ/kg E10/B16: 41 MJ/kg	
Constant oxygen mass content			E50/B83: 35 MJ/kg	

## **Experimental Methods - Steady State Testing**

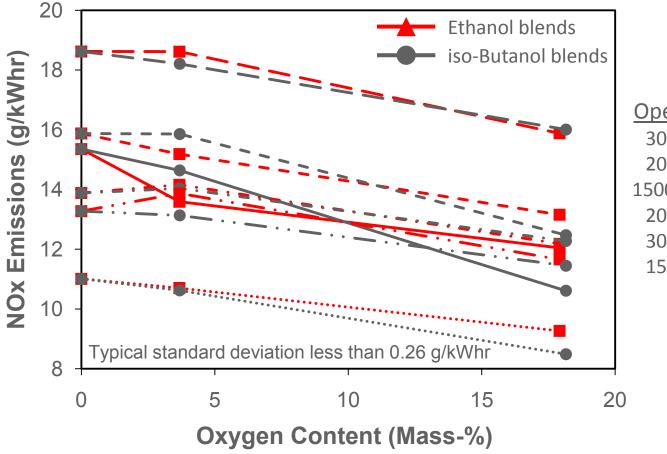


- **Typical vehicle operating points** 
  - Constant speed
  - Constant load
  - Constant power
- Engine-out (no TWC) emissions

- Opel 2.2 | Ecotec Direct (GM L850)
  - SIDI 4-Cylinder Engine, no VVA
  - ECU calibrated for gasoline
- Emissions measurement
  - Horiba MEXA Model 7100D-EGR
  - AVL Sesam FTIR



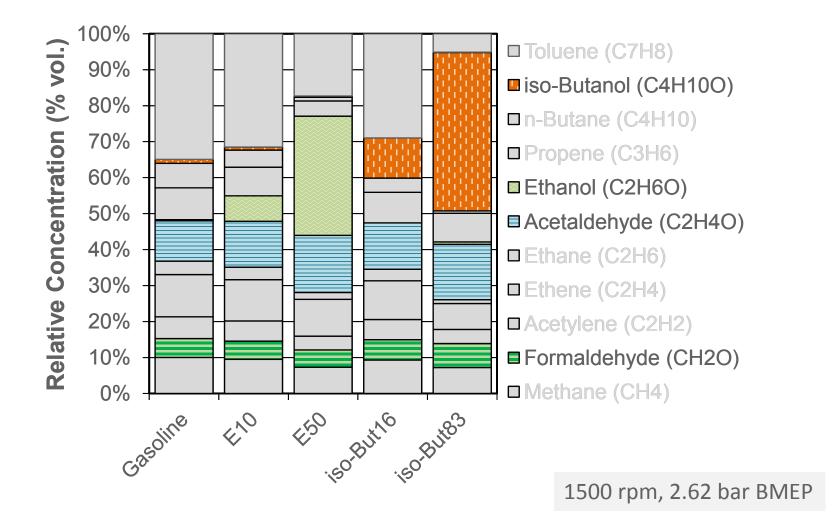
#### **Regulated Emissions**



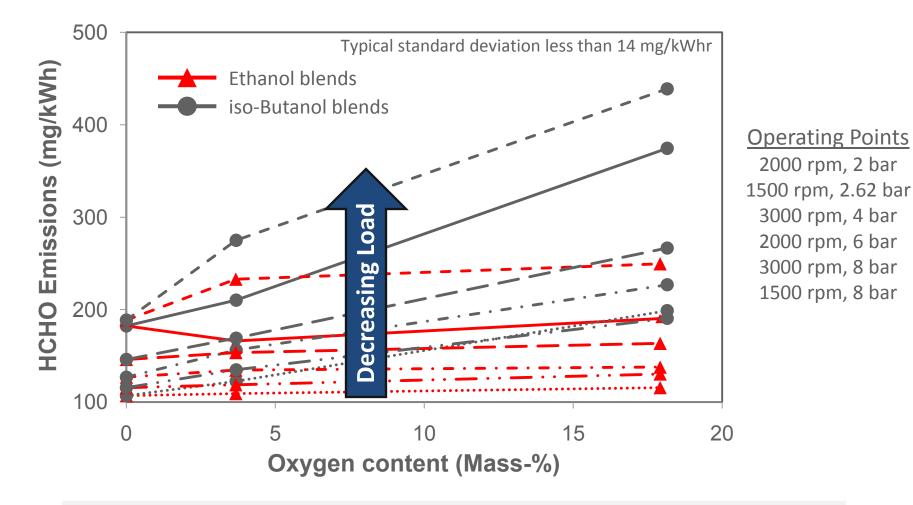
<u>Operating Points</u> 3000 rpm, 4 bar 2000 rpm, 2 bar 1500 rpm, 2.62 bar 2000 rpm, 6 bar 3000 rpm, 8 bar 1500 rpm, 8 bar

NOx decreases with increasing fuel alcohol and oxygen content

## Speciated Hydrocarbons (from FTIR)

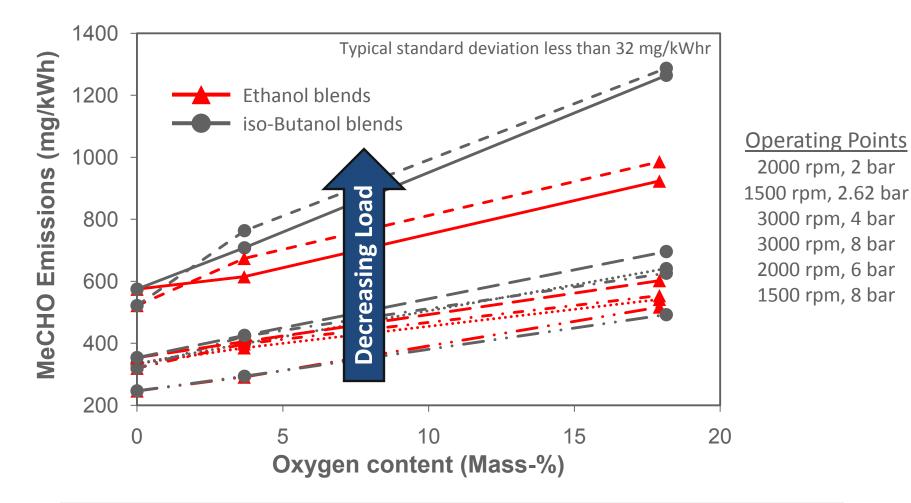


## Formaldehyde Emissions



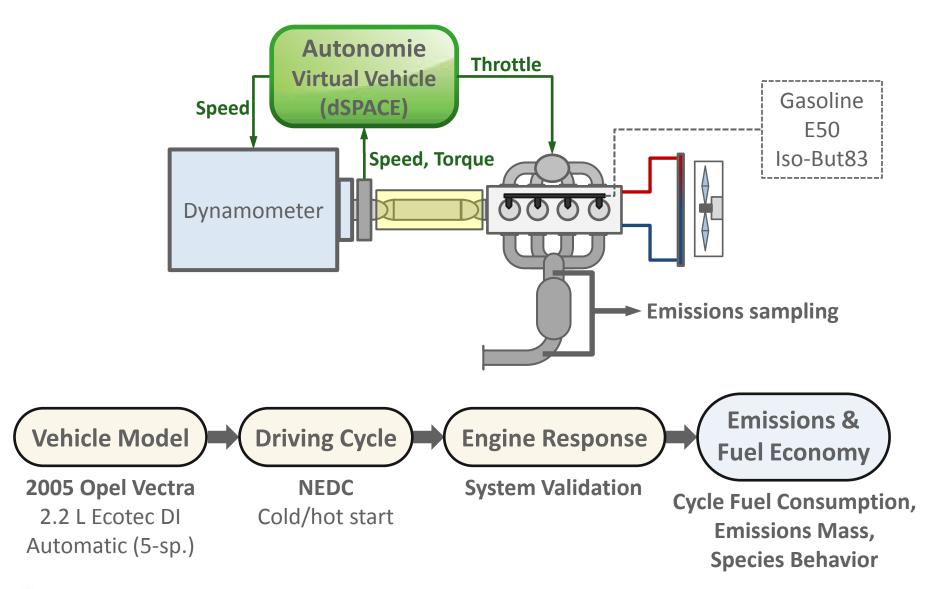
Increased formaldehye with iso-butanol blends, but not with ethanol blends

## Acetaldehyde Emissions

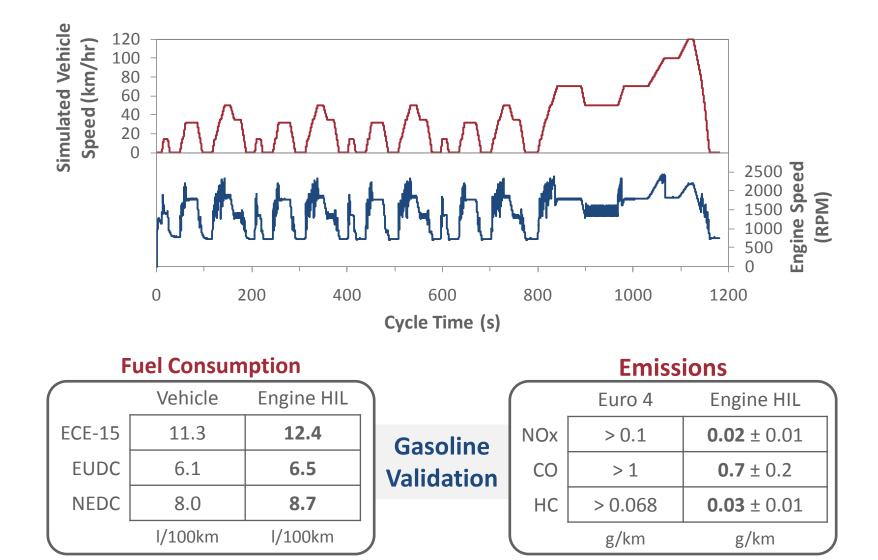


Increased acetaldehye with both ethanol and *iso*-butanol blends

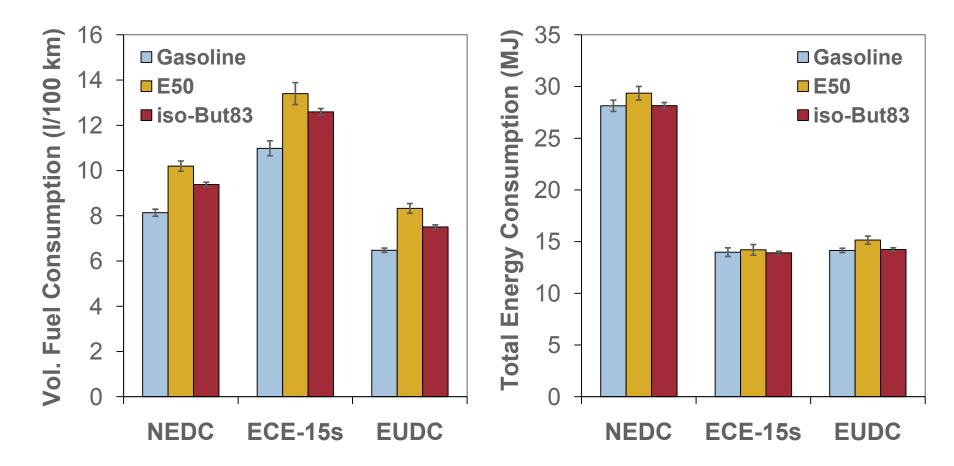
### Engine Hardware-In-Loop Concept



### **Experimental Methods: Engine HIL Testing**

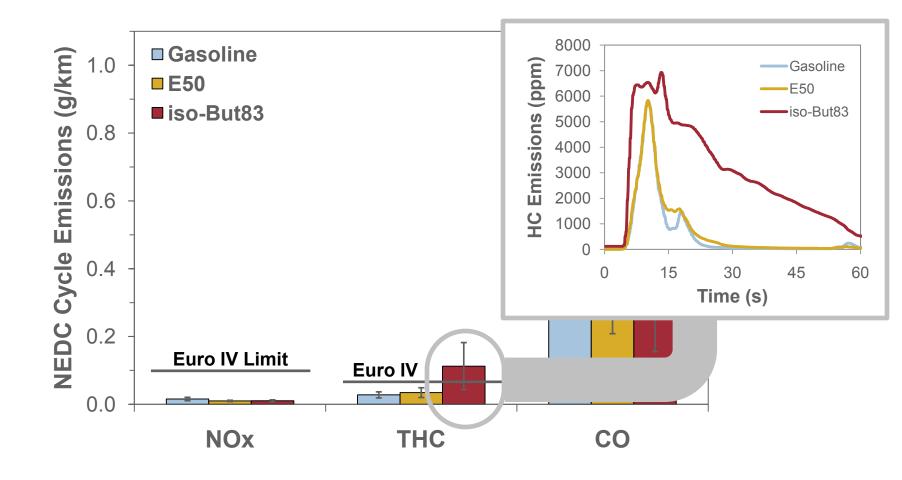


### **Cycle Fuel and Energy Consumption**



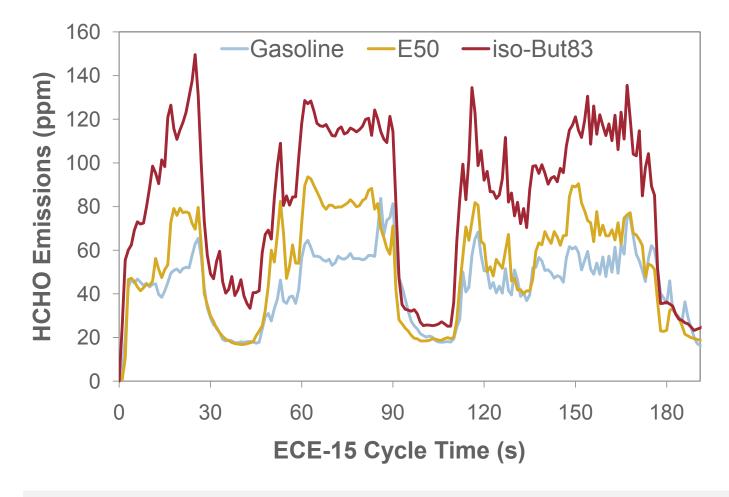
Hot-start cycles, integrated fuel flow

## **Total Cycle Emissions Mass**



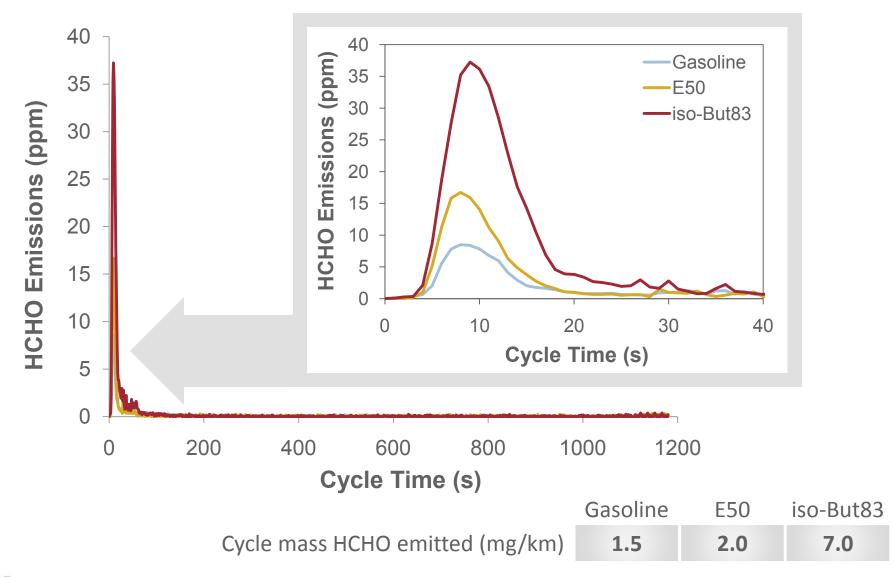
NEDC, cold-start, post-TWC emissions

#### Formaldehyde Emissions: Hot-Start, Pre-TWC

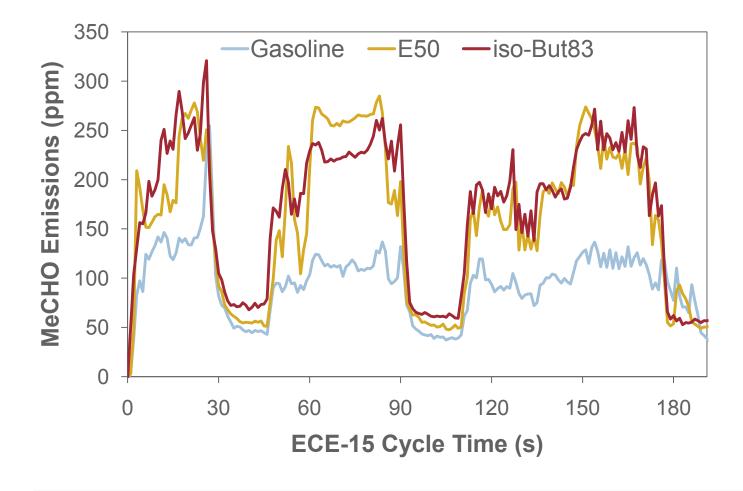


1<sup>st</sup> urban cycle of NEDC, hot-start, pre-TWC emissions

#### Formaldehyde Emissions: Cold-Start, Post-TWC

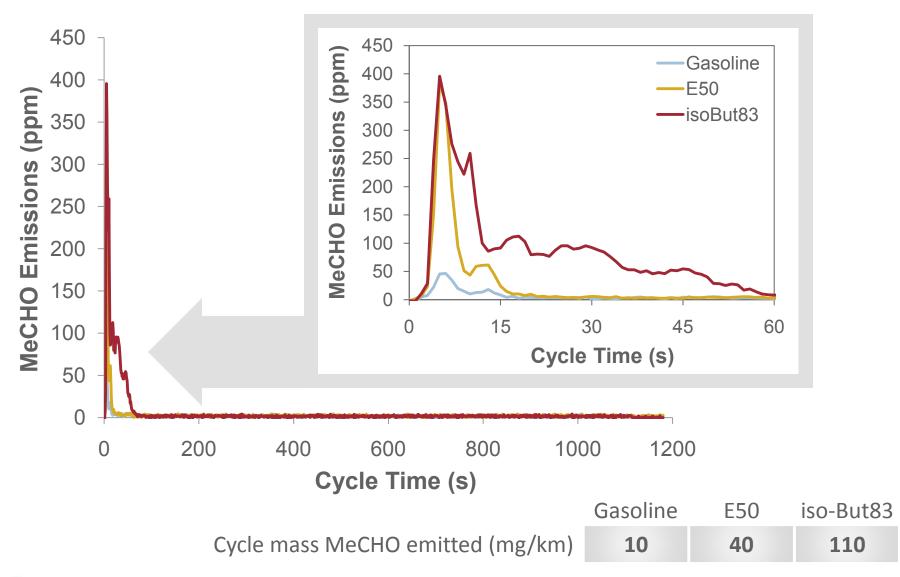


#### Acetaldehyde Emissions: Hot-Start, Pre-TWC



1<sup>st</sup> urban cycle of NEDC, hot-start, pre-TWC emissions

#### Acetaldehyde Emissions: Cold-Start, Post-TWC



## **Conclusions & Future Opportunities**

- Both ethanol and *iso*-butanol blends reduced cycle mass emissions of NOx and CO, and yield comparable cycle energy consumption.
- Blends of gasoline and *iso*-butanol increase both acetaldehyde and formaldehyde emissions, while ethanol-gasoline blends increase acetaldehyde emissions, but not significantly formaldehyde.
- Aldehyde emissions are eliminated in an active (warm) three-way catalyst: cycle aldehyde emissions stem from initial cold-start phase.
- Improved cold-start engine operation with high-alcohol fuels (including iso-butanol blends) is critical for meeting emissions targets
- Future exploration opportunities include particulate matter characterization from alcohol fuels utilizing engine HIL.



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