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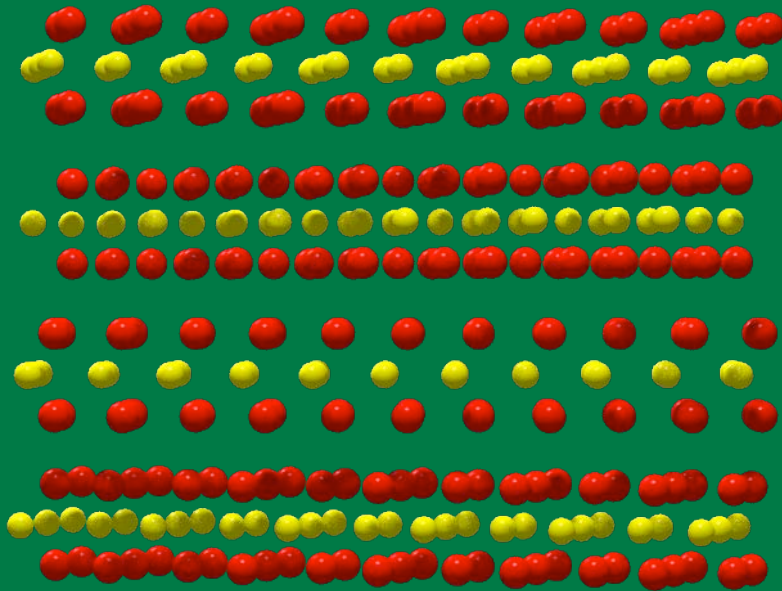
New Thermoelectric Materials with Ultra-low Thermal Conductivity

Mary Smeller, Qiyin Lin, Colby Heideman, Paul
Zschack, Michael Anderson, Andrew A. Herzing, Ian
Anderson, Raimar Rostek, Matt Beekman and
David C. Johnson

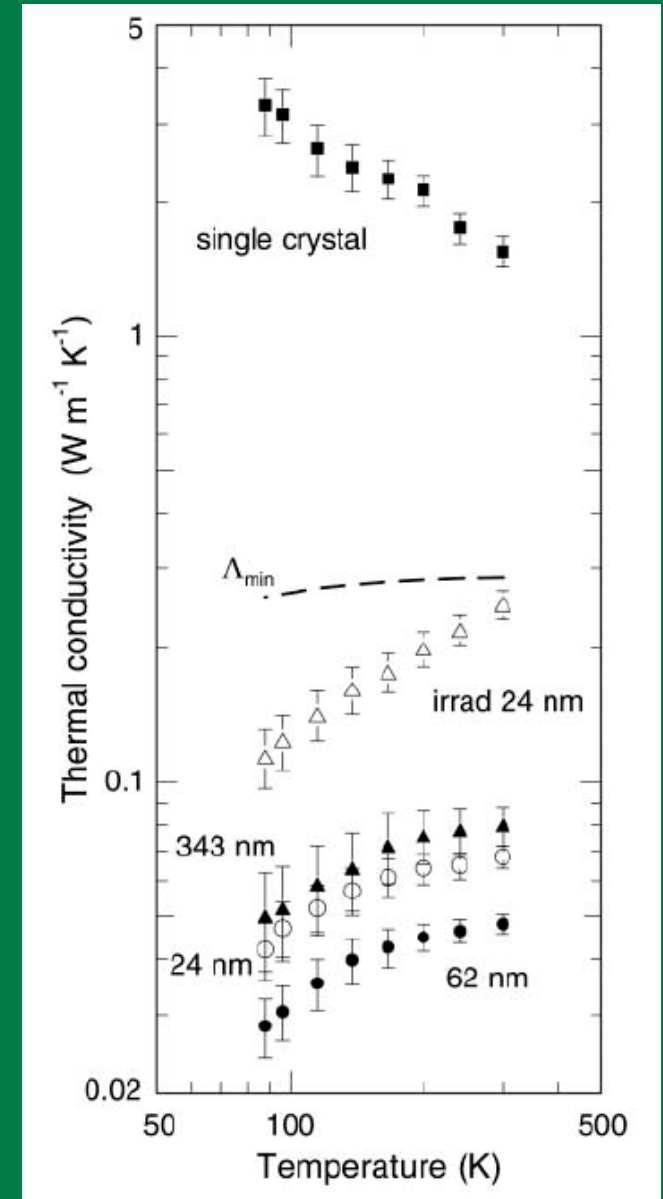
Chemistry Department
Materials Science Institute
University of Oregon

Ultralow Thermal Conductivity in WSe₂

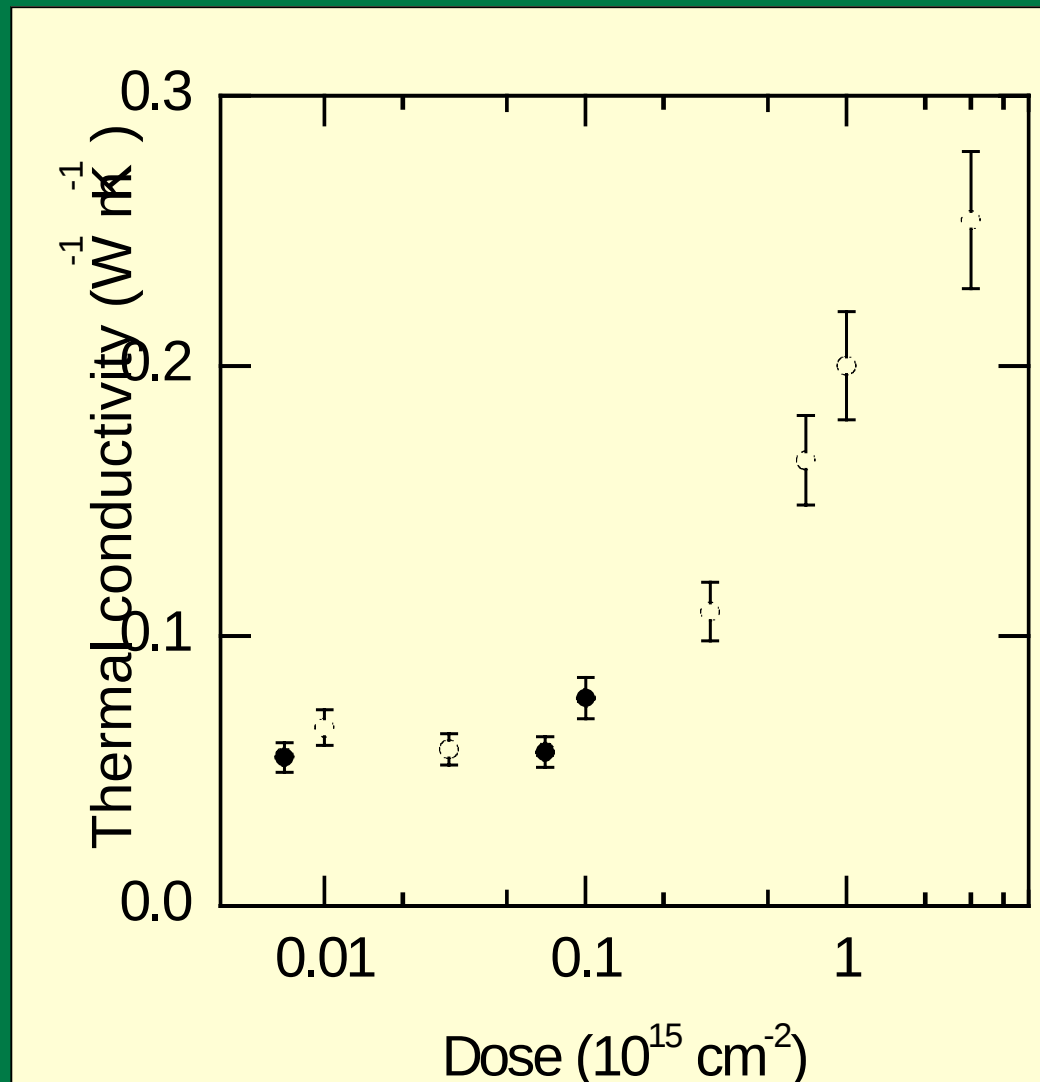
- Crystalline within sheets
- Disordered between sheets



C. Chiritescu, D. G. Cahill, N. Nguyen, D. C. Johnson, A. Bodapati, P. Keblinski, P. Zschack, *Science* **2007**, 315, 351.



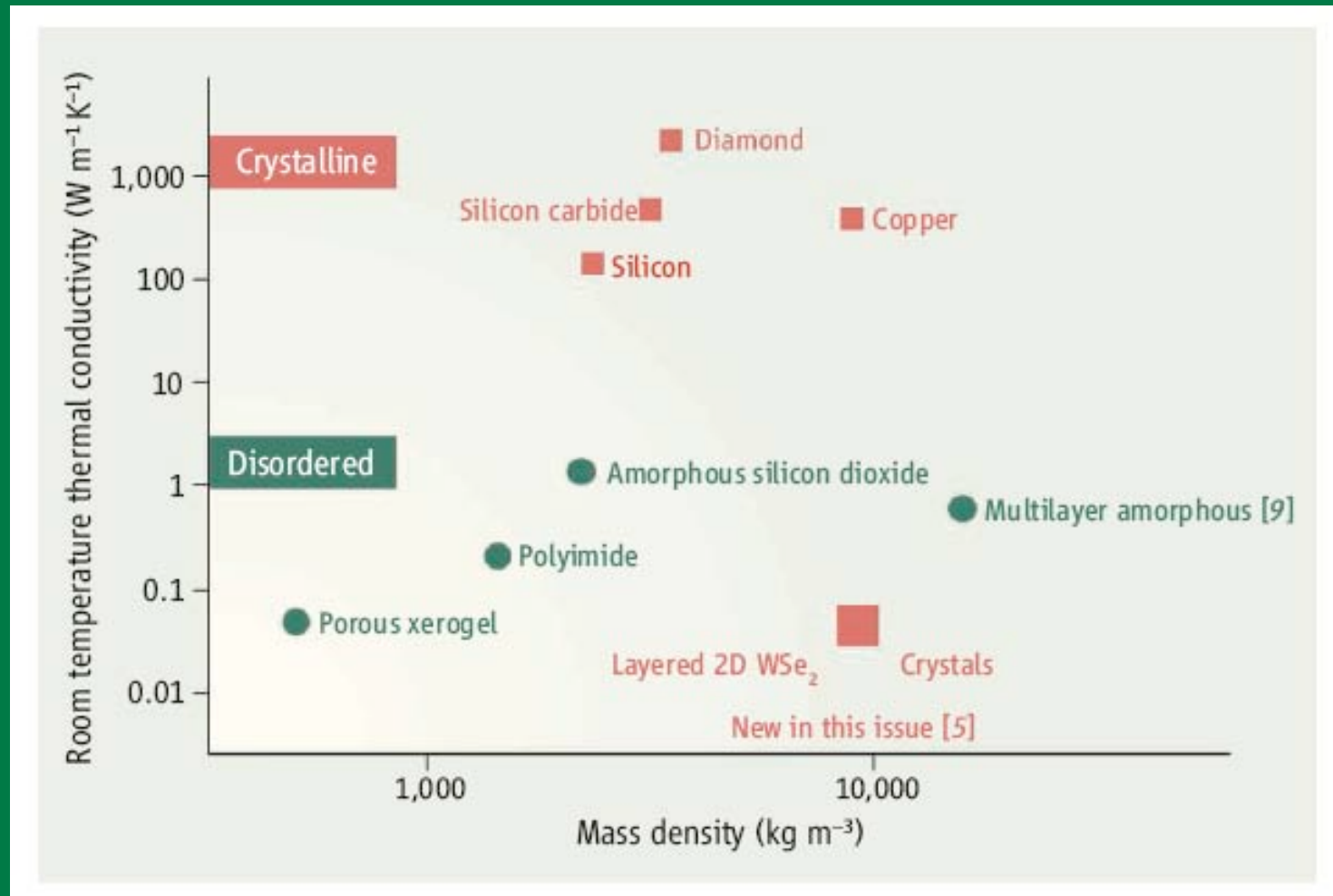
Balance of Order and Disorder Important



Ion damage destroys order in a-b plane and along c.

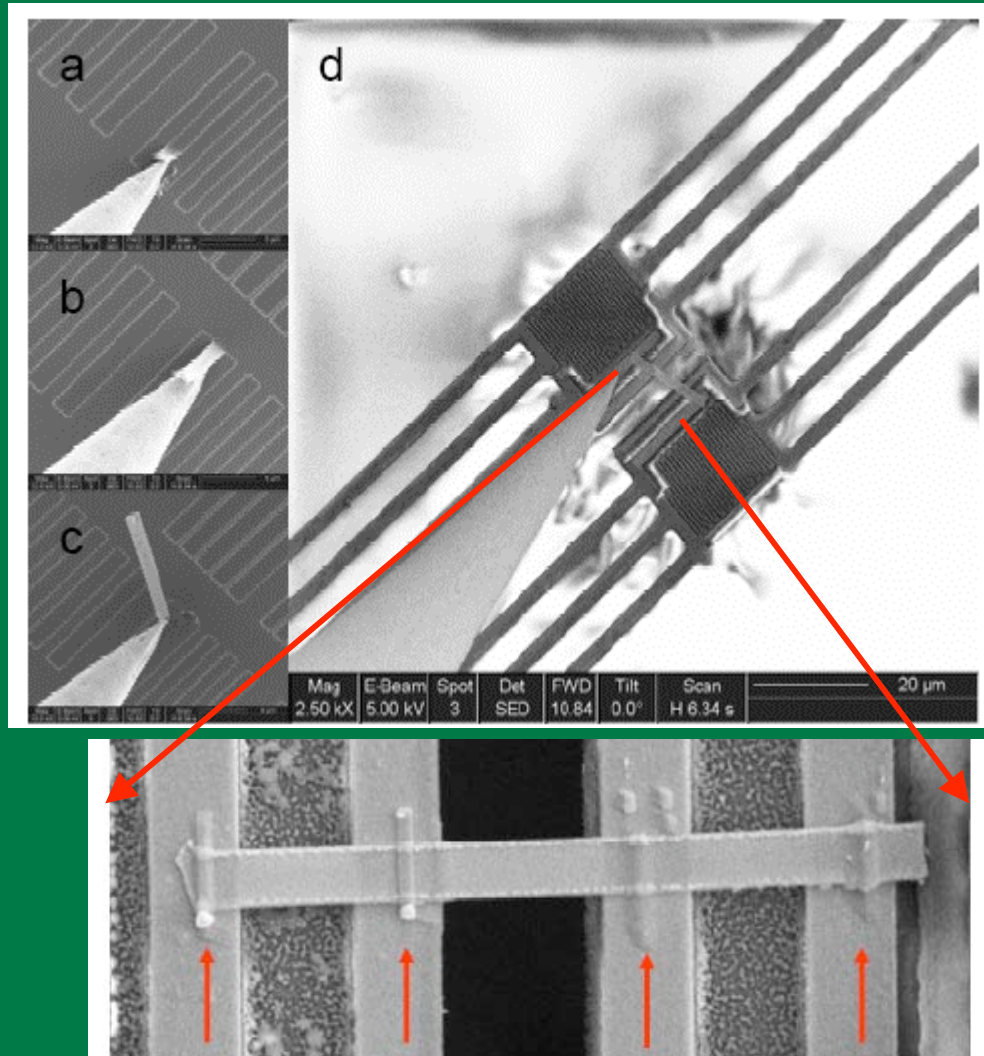
Ion damage increases the thermal conductivity!

Thermal Conductivity of Solids



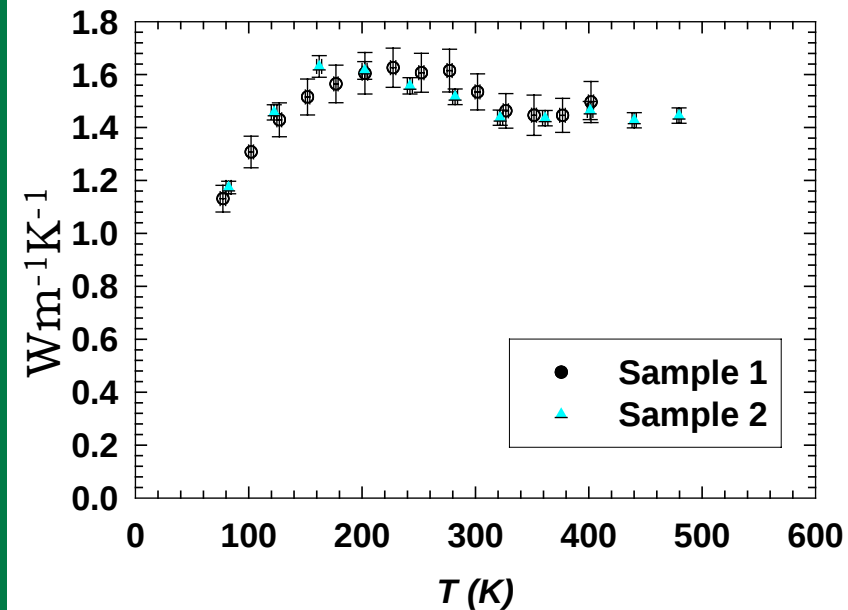
K. E. Goodson, *Science* **2007**, 315, 342.

Device Assembly for In-plane Thermal Transport

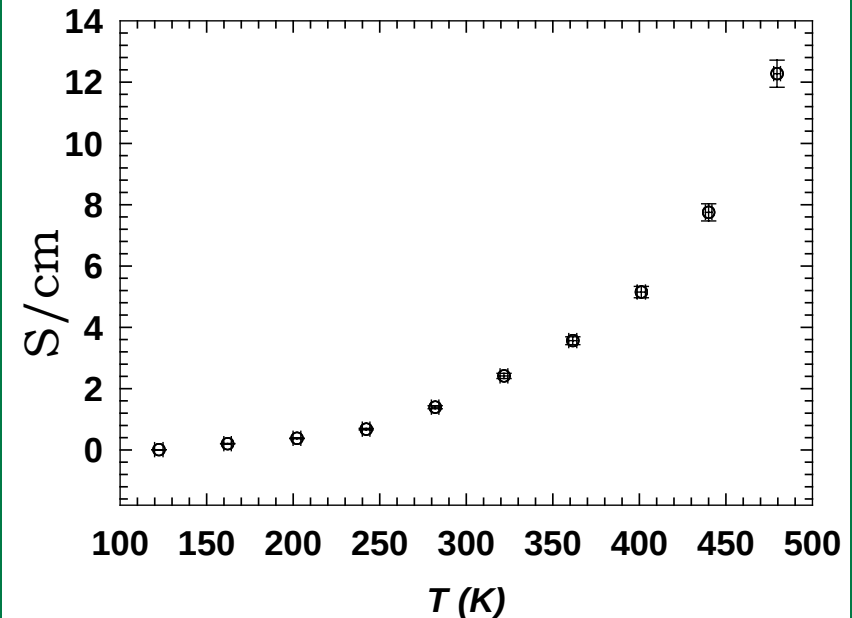


Courtesy of Dr. Li Shi, UT at Austin

WSe₂ In-plane Properties



Thermal conductivity



Electrical conductivity

Need a way to tune electrical properties

Misfit Layered Compounds

- Interleaved layers of transition metal dichalcogenides and rock salt structures
- Lattice mismatch in the a-b plane $[(AX)_{1+y}]_n(MX_2)_m$
- $n, m=1$ forms by direct reaction of the elements at $\sim 1000^\circ\text{C}$ for ~ 30 compounds
- Compounds can be metallic, semiconducting or insulating (magnetic, superconducting, etc.)
- ~ 10 different dichalcogenides and ~ 30 different rock salt constituents - lots of potential flexibility

Prior Results

- $(\text{YbS})_{1.24}\text{NbS}_2$
 $S = 60\mu\text{V/K}$
 $\bar{\sigma} = 1.4 \times 10^{-5} \Omega \cdot \text{m}$
 $K = 0.8 \text{ W m}^{-1} \text{ K}^{-1}$
 $ZT = 0.1$
- $(\text{LnS})_m(\text{NbS}_2)_n$ and $(\text{SnS})_m(\text{TiS}_2)_n$ $n, m = 1, 2$
power factor, $6 \sim 12 \times 10^{-4} \text{ W/m} \cdot \text{K}^{-2}$ at 300K
If the K value is $\sim 1 \text{ W m}^{-1} \text{ K}^{-1}$,
 $ZT \sim .1-.3$

Miyazaki, Ogawa, Kajitani, Jap. J. Appl. Physics, 43 (2004) L1202

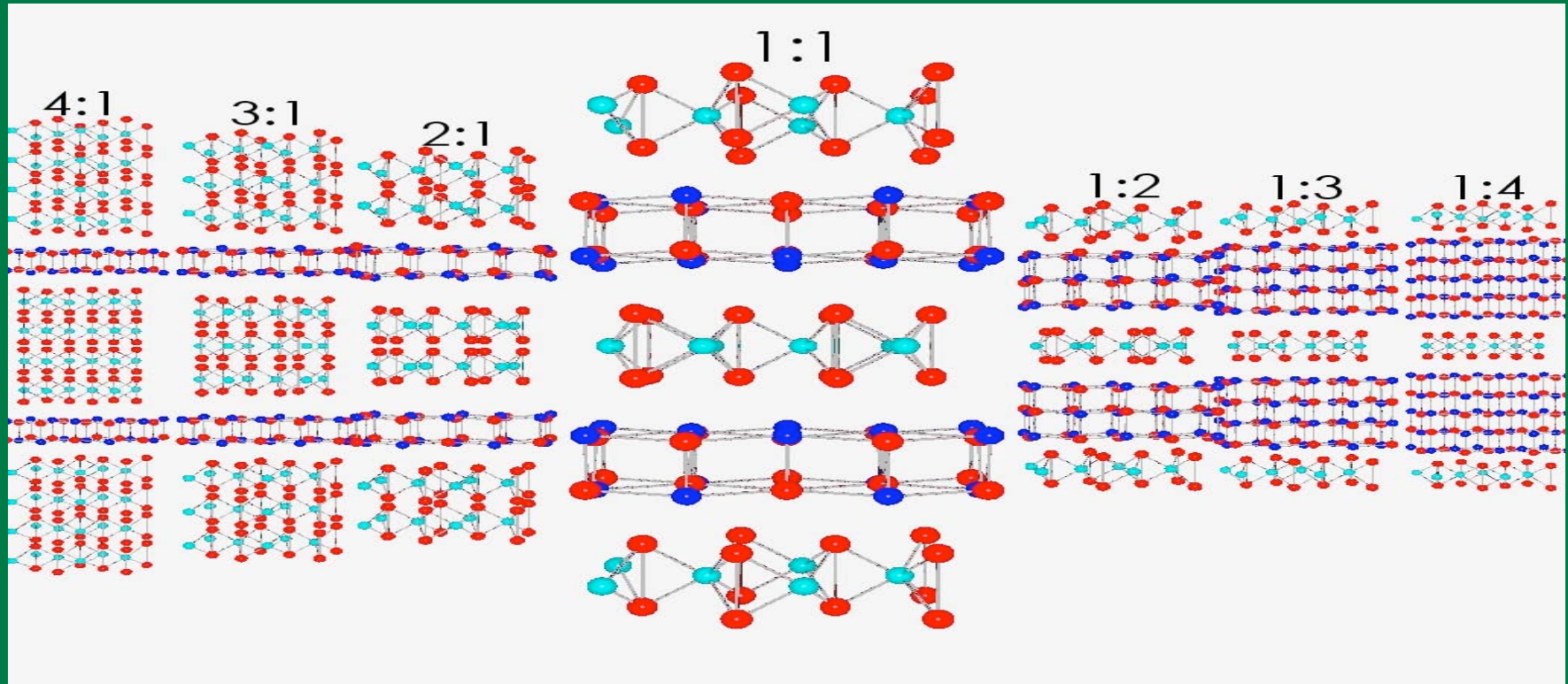
- $(\text{SnS})_1(\text{TiS}_2)_2$ $K_{\text{lattice}} \sim 0.8 \text{ W m}^{-1} \text{ K}^{-1}$,
 $ZT \sim .4$ at 400°C

C. Wan, Y. Wang, N. Wang and K. Koumoto - ICT 2009 Freiburg

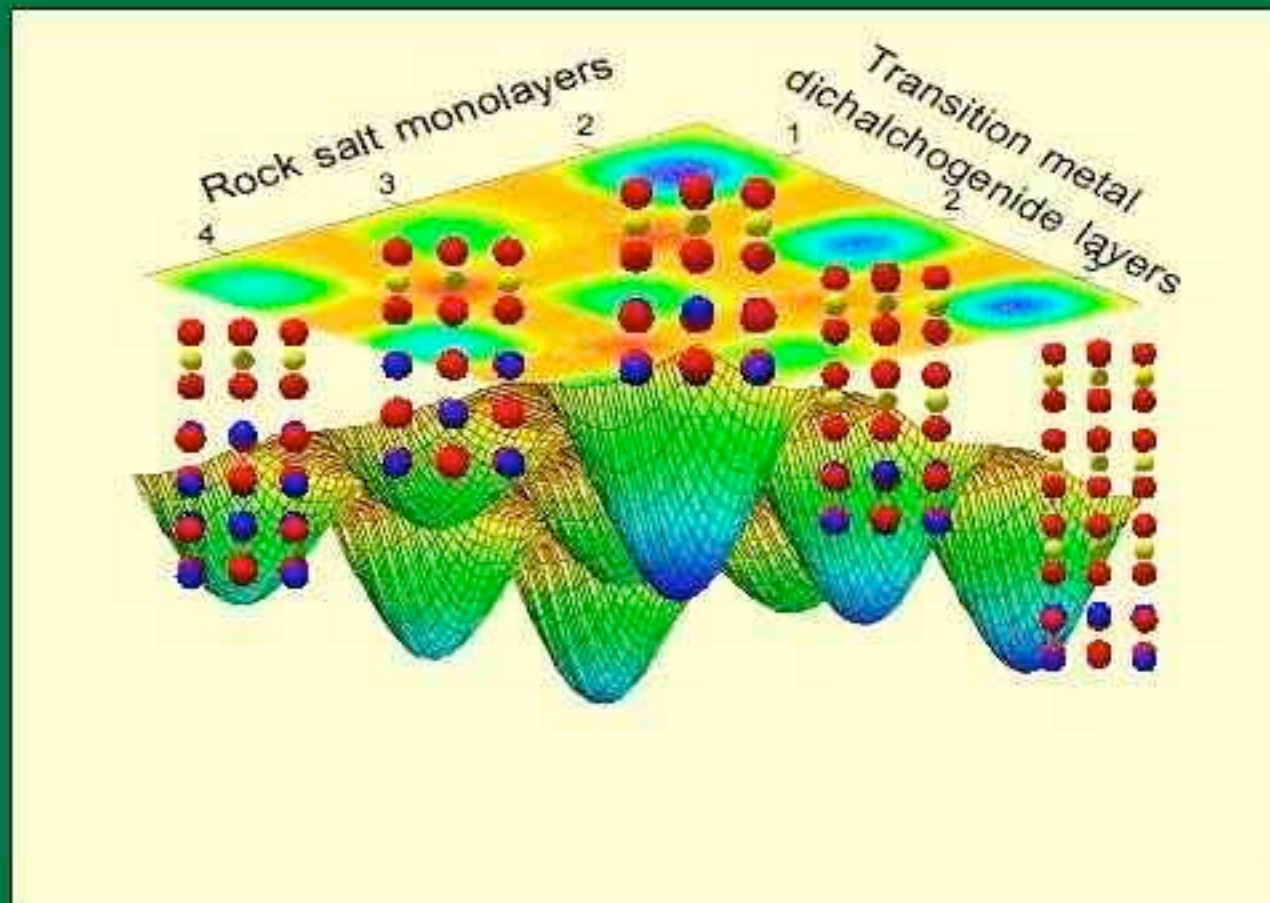
No report of controlled doping

Only one or two compounds in each family made

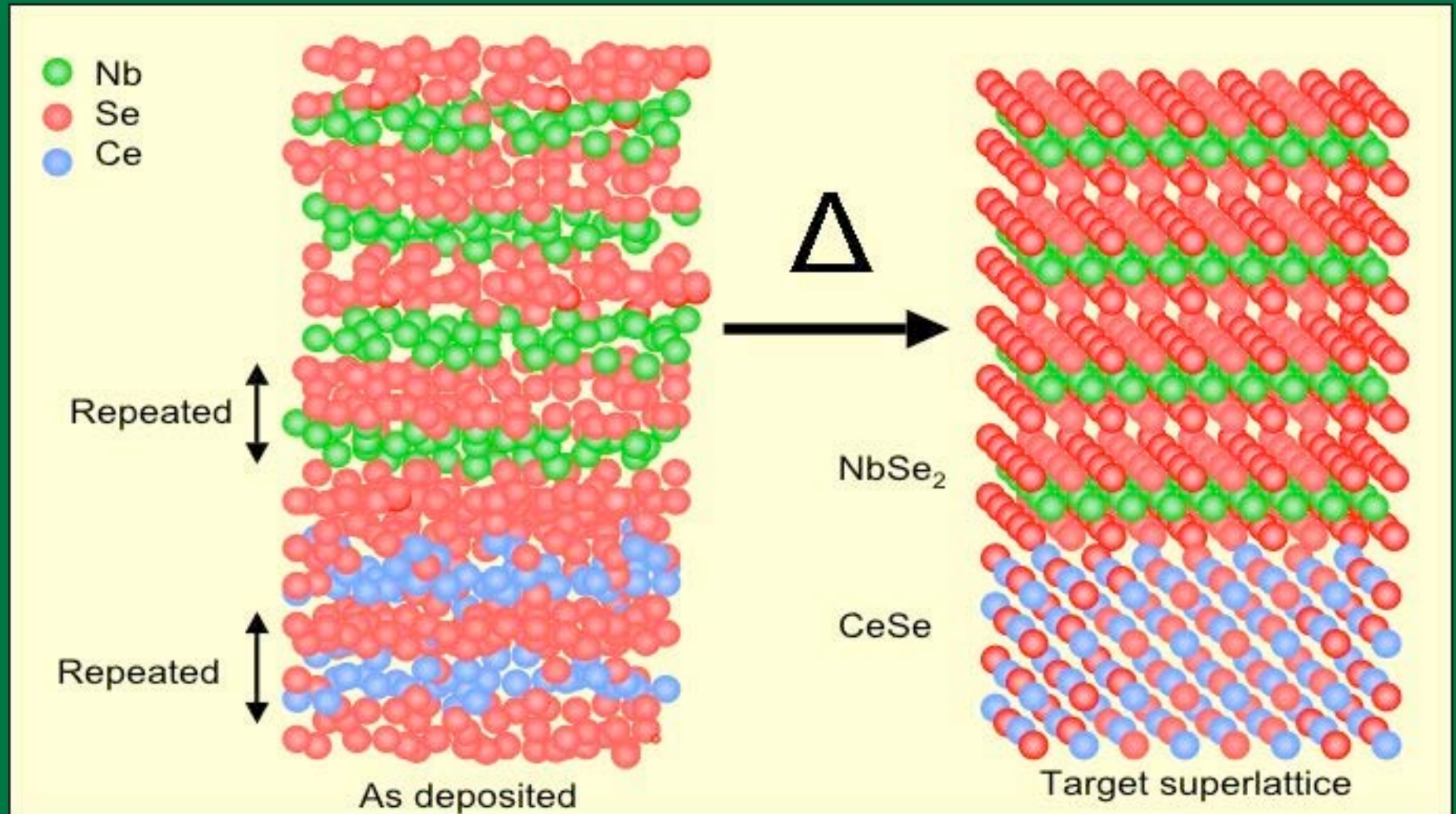
Controlled Nanostructure



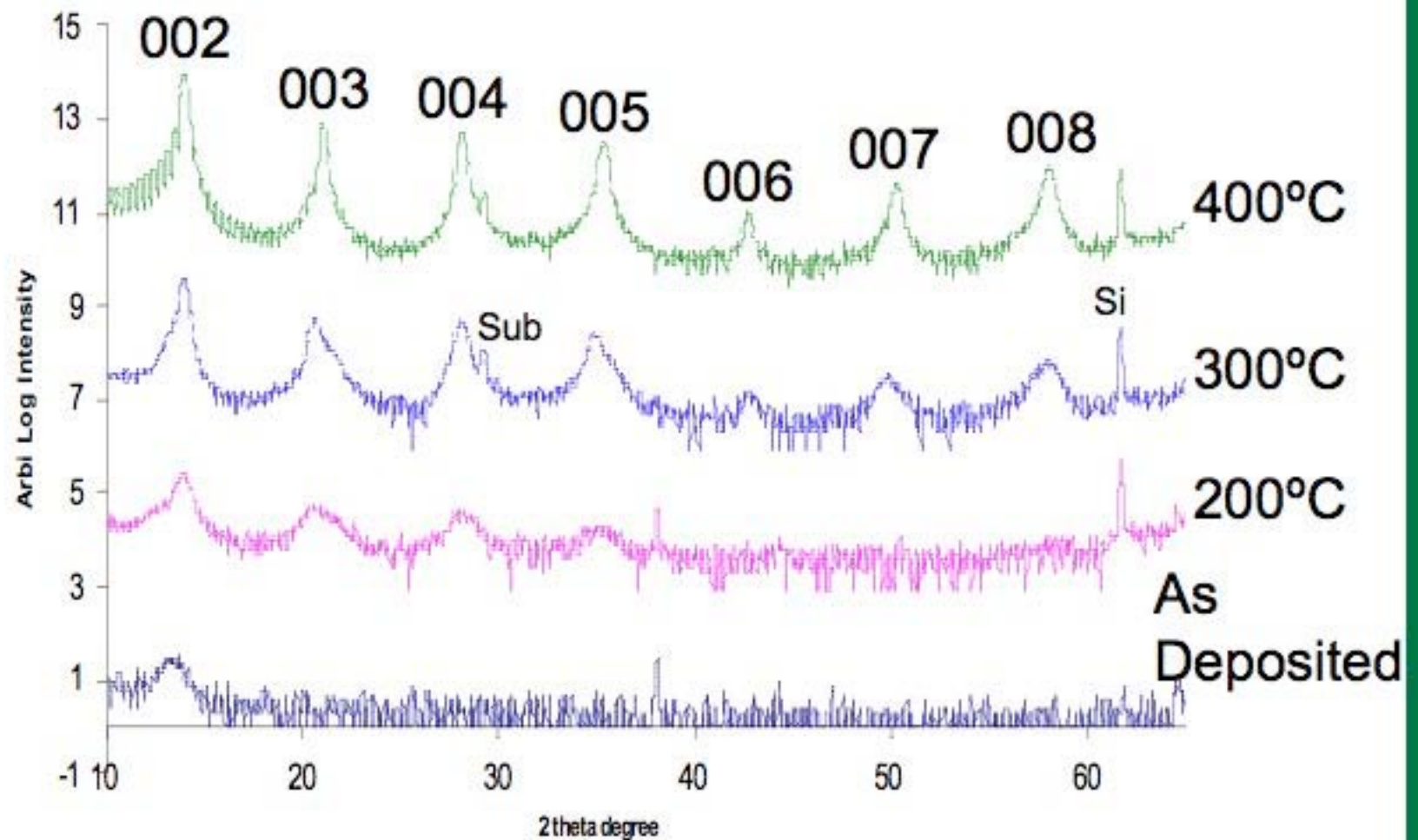
Assume local free energy minima
at each value of n and m



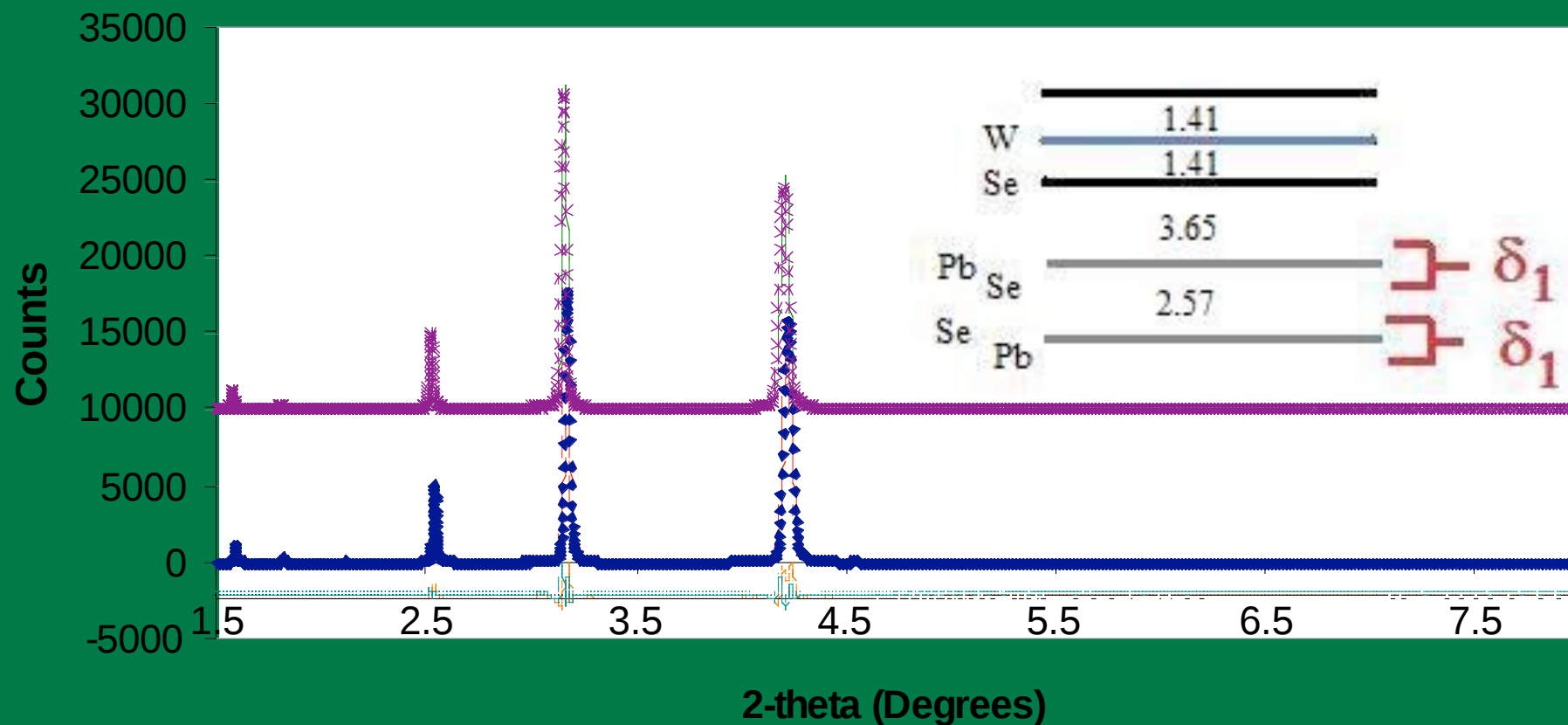
Two diffusion lengths trap nanostructure



$[(\text{PbSe})_{0.99}]_1[\text{WSe}_2]_1$ Superlattice forms with annealing

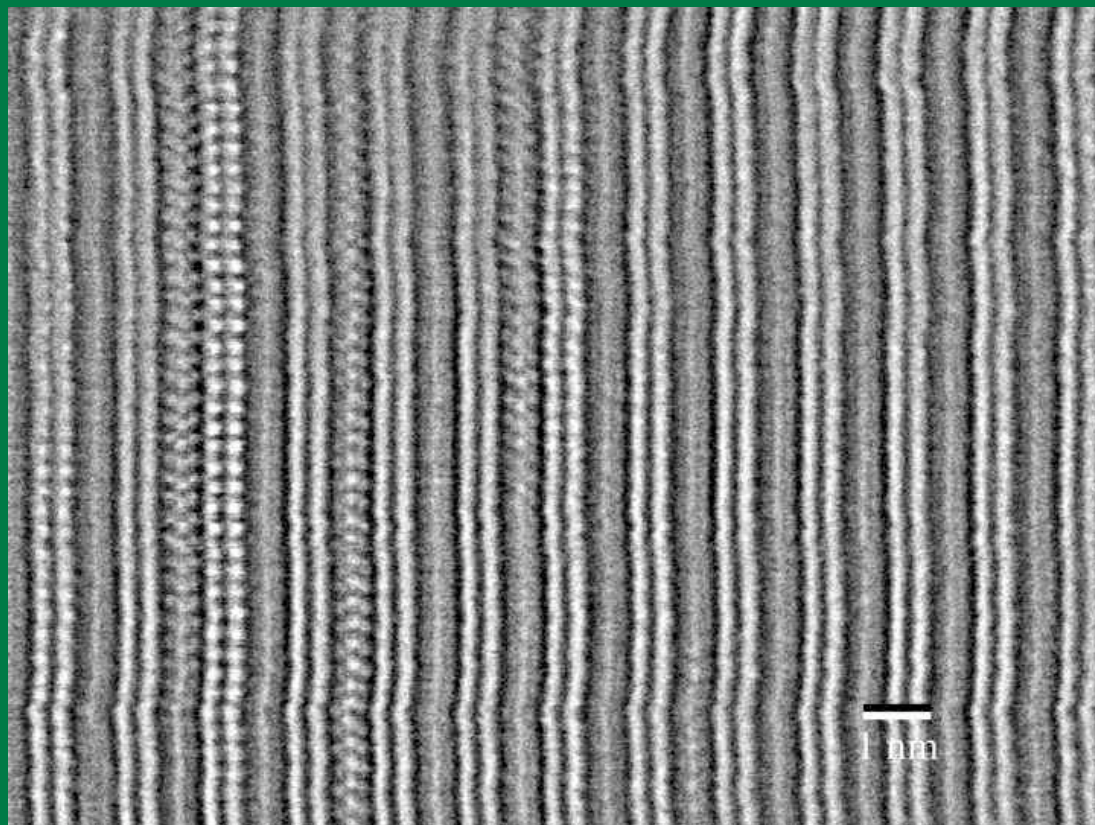


$(\text{PbSe})_1(\text{WSe}_2)_1$ Refinement



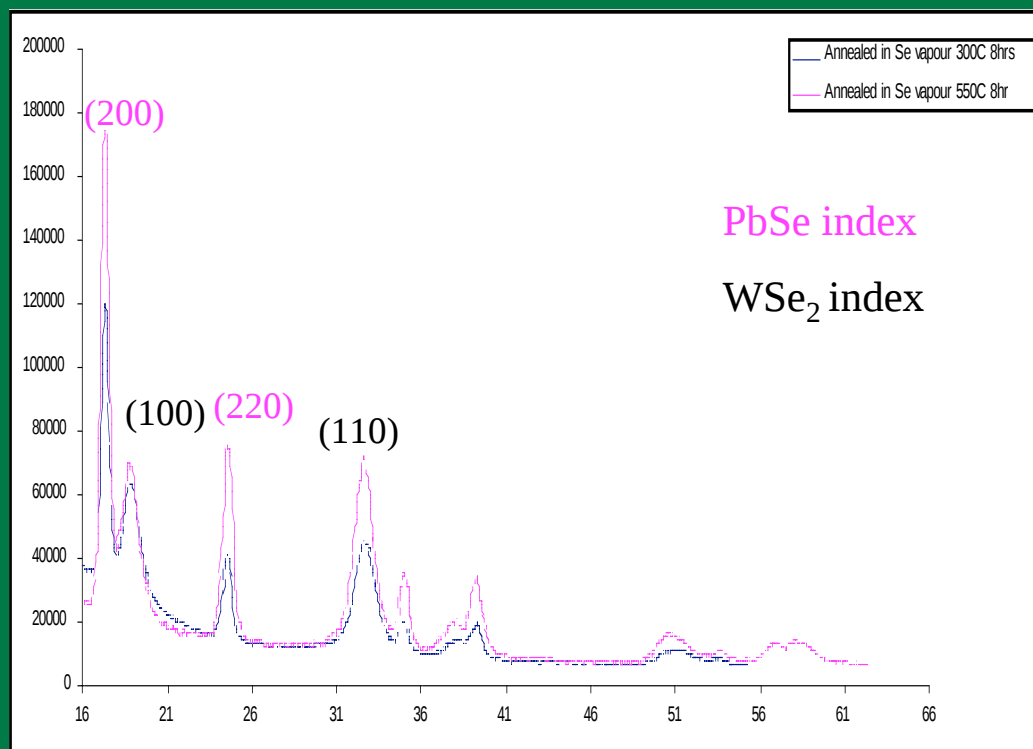
◆ 0.9819 Experimental	— 0.9819 Le Bail	— 0.9819 Difference	— Z-LINE
✱ 0.9855 Experimental	— 0.9855 Le Bail	— 0.9855 Difference	

Distortion from 001 rock salt plane



$(\text{CeS})_{1.15}\text{TaS}_2$	0.3623
$(\text{LaS})_{1.13}\text{TaS}_2$	0.3469
$(\text{BiS})_{1.07}\text{TaS}_2$	0.4164
$(\text{PbS})_{1.18}\text{TiS}_2$	0.4407
$(\text{SnS})_{1.20}\text{TiS}_2$	0.5540
$(\text{PbS})_{1.14}\text{NbS}_2$	0.5212
$(\text{LaS})_{1.14}\text{NbS}_2$	0.6058
$(\text{PbS})_{1.13}\text{TaS}_2$	0.4
$(\text{SnSe})_{1.16}\text{NbSe}_2$	0.2456
$(\text{PbS})_{1.14}(\text{NbS}_2)_2$	0.2068
$(\text{PbSe})_{0.99}\text{WSe}_2$	0.228
$(\text{PbSe})_{0.99}\text{MoSe}_2$	0.228
$(\text{SnSe})_{0.99}\text{MoSe}_2$	0.364

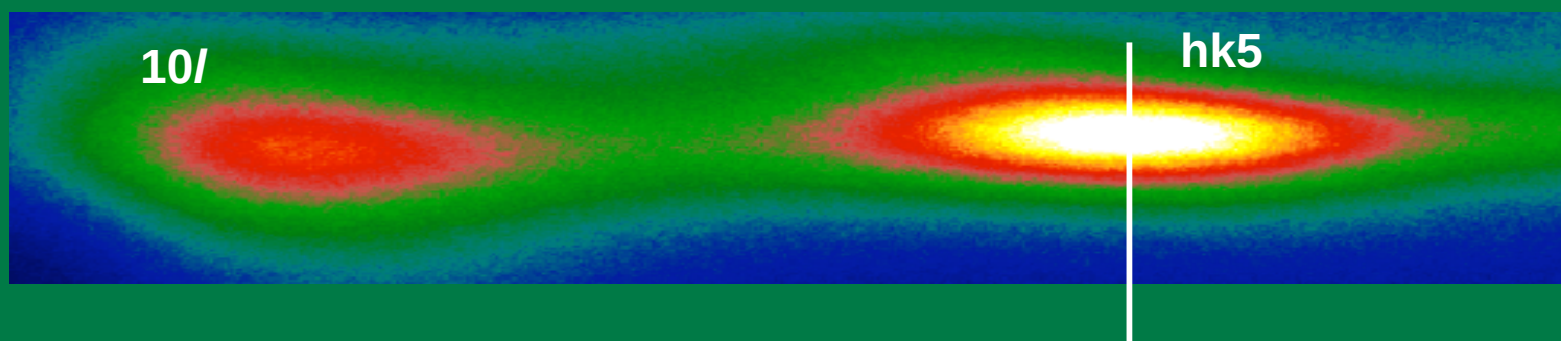
In-plane Structure $(\text{PbSe})_1(\text{WSe}_2)_1$



Long range order
within a-b planes

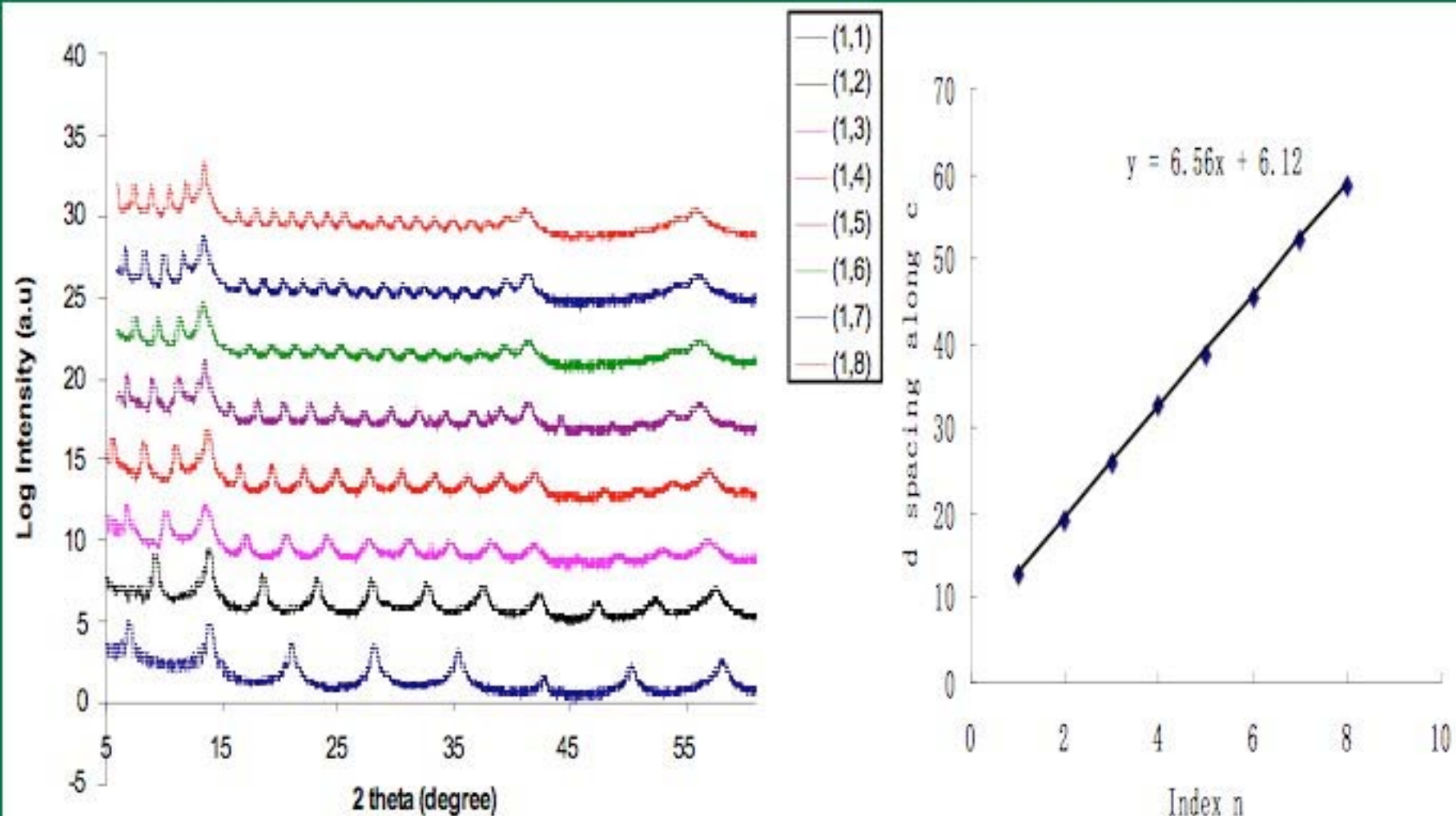
Grain sizes larger
in PbSe than WSe_2

Short range order
between a-b planes

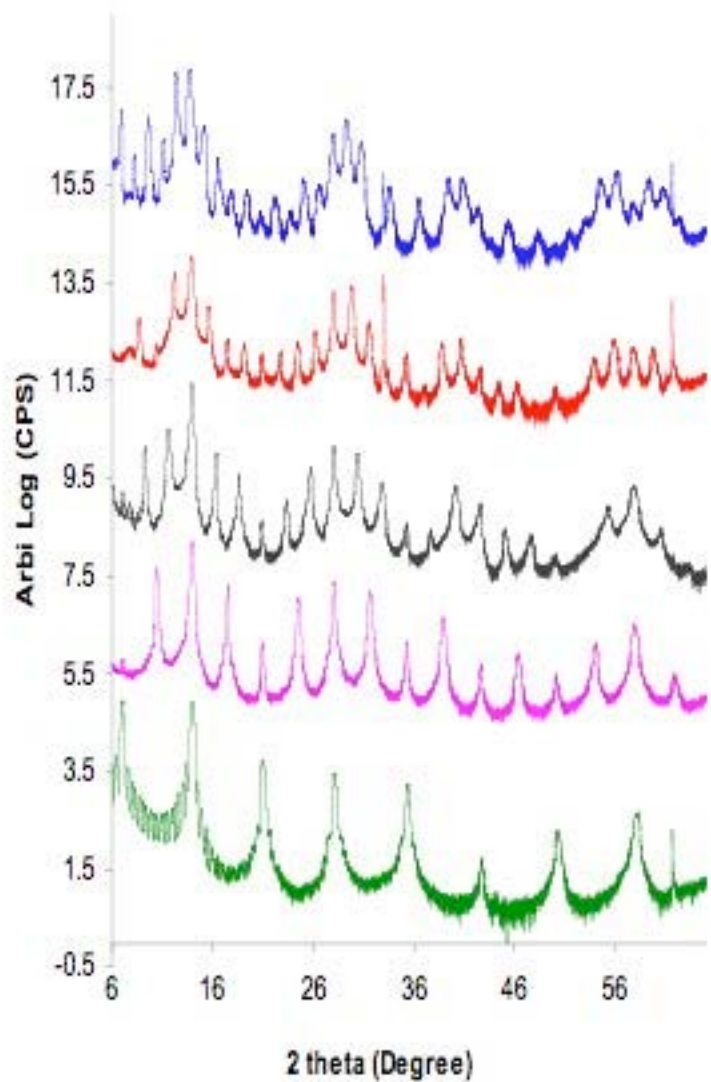


Unique Structure of Misfit compounds

- Thermal expansion of components differ
Planes slide over one another
Robust to thermal cycling
- Abrupt interfaces between components with minimal interdiffusion
Thermodynamically stable metal - misfit junctions
low contact resistances possible because of zero or minimal interfacial phase growth.
- Electron transfer between components a method to tune through resonance states



$[(\text{PbSe})_{0.99}]_n(\text{WSe}_2)_n$ Superlattices



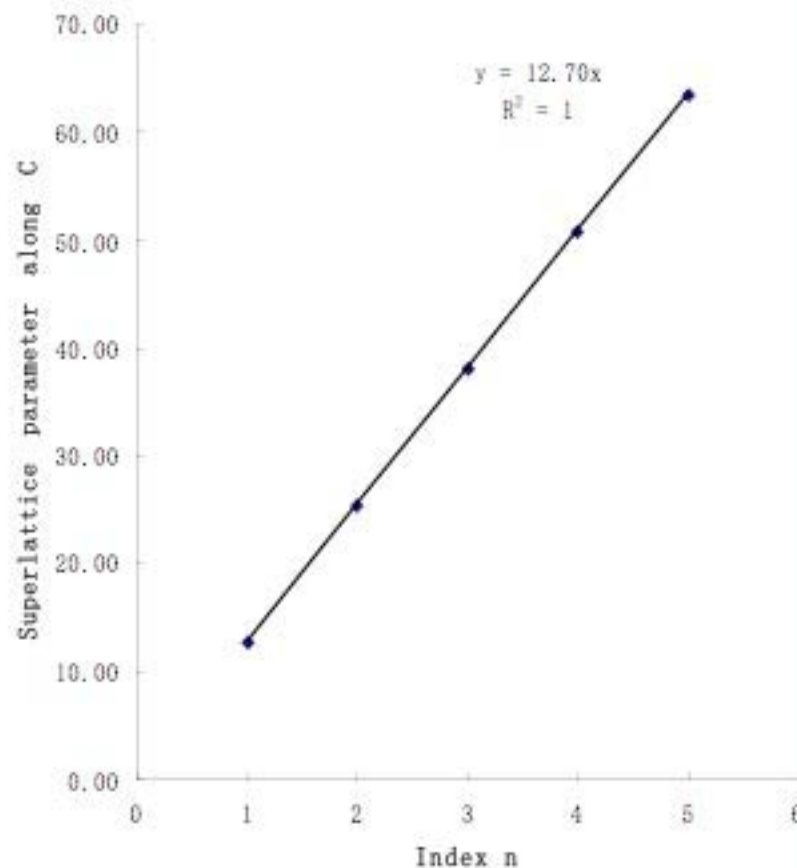
(5,5)

(4,4)

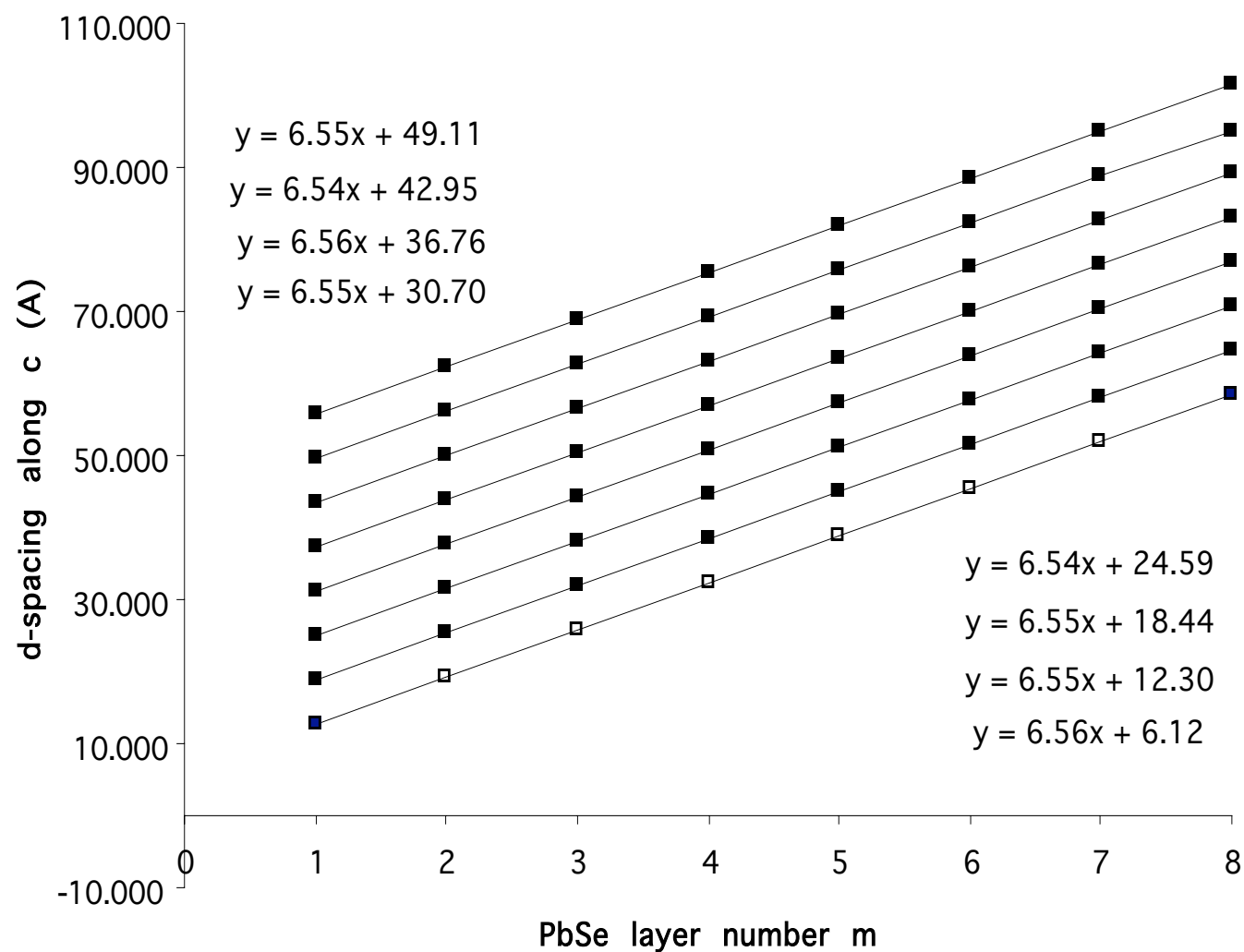
(3,3)

(2,2)

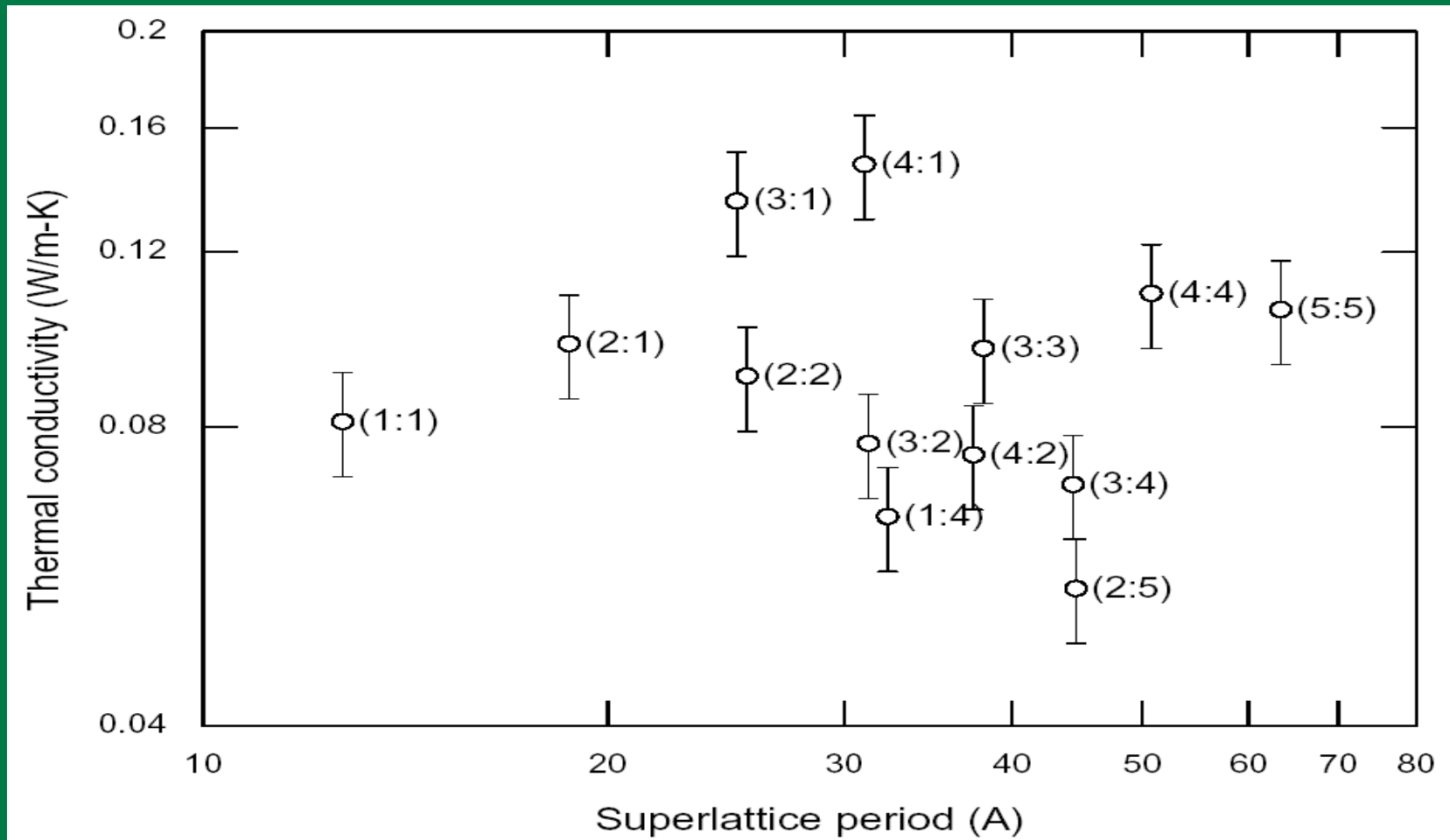
(1,1)



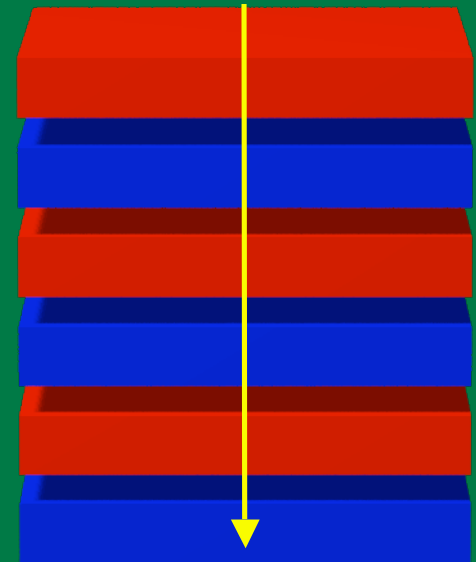
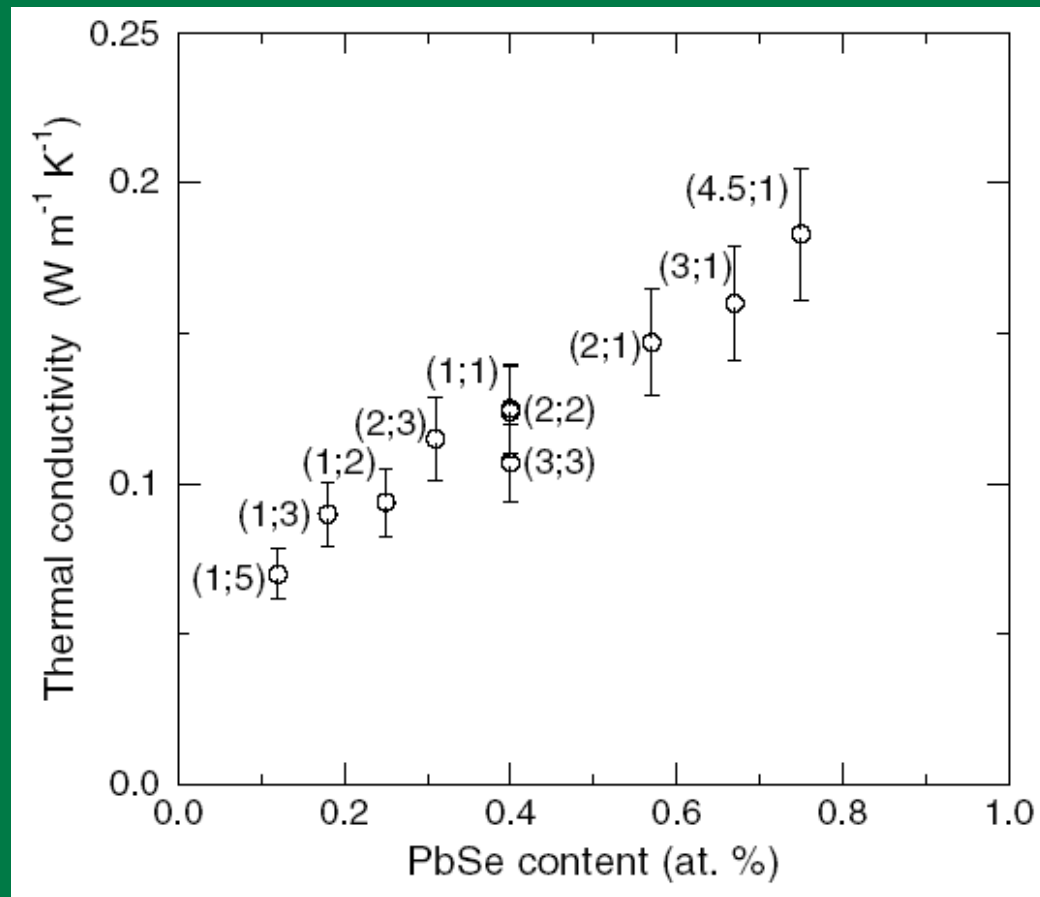
d-space of first 64 $[(\text{PbSe})_{0.99}]_m(\text{WSe}_2)_n$



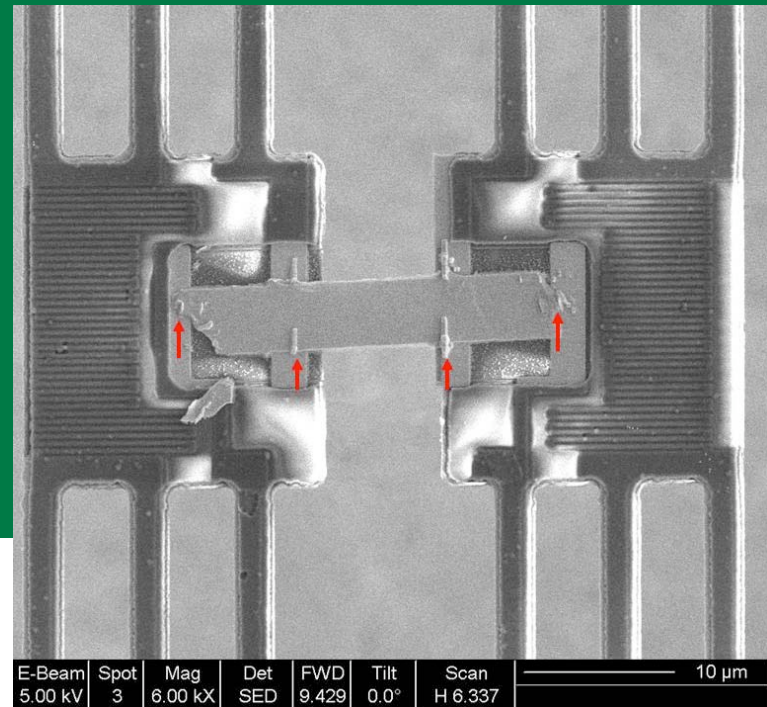
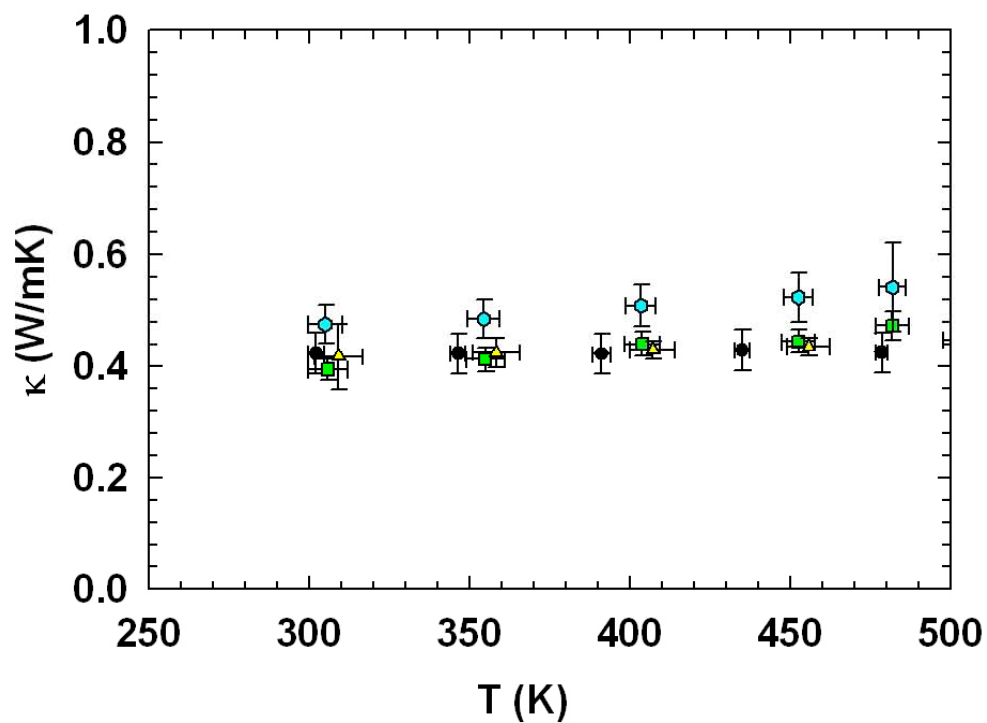
Thermal Conductivity $[(\text{PbSe})_{0.99}]_n(\text{WSe}_2)_m$



Cross Plane Thermal Conductivity (PbSe)_n(MoSe₂)_m

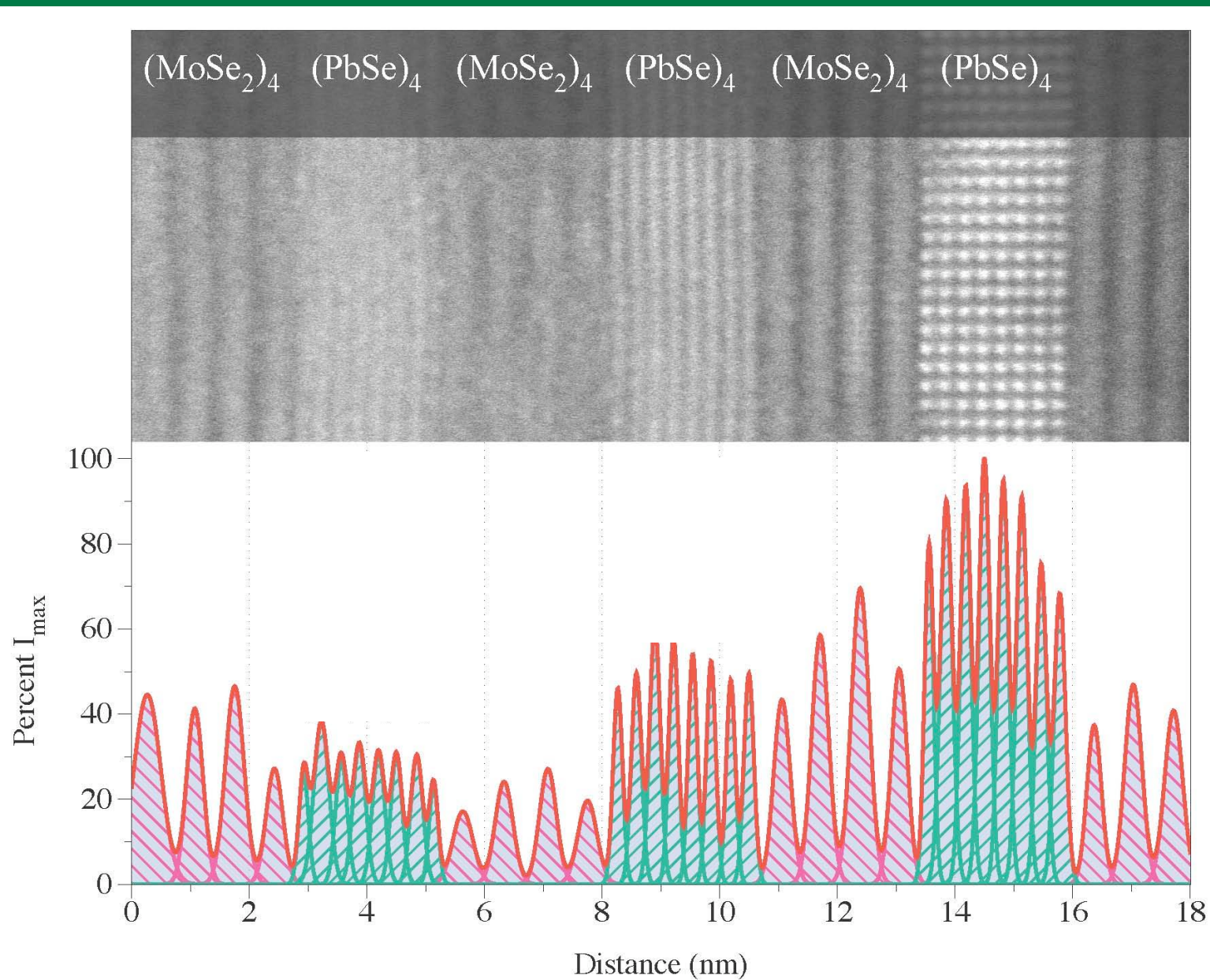


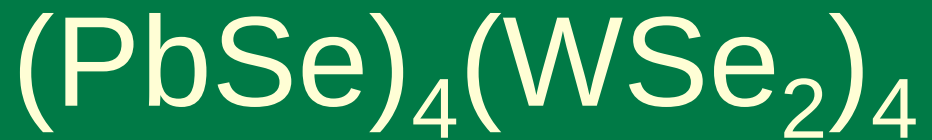
In-plane Thermal Conductivity



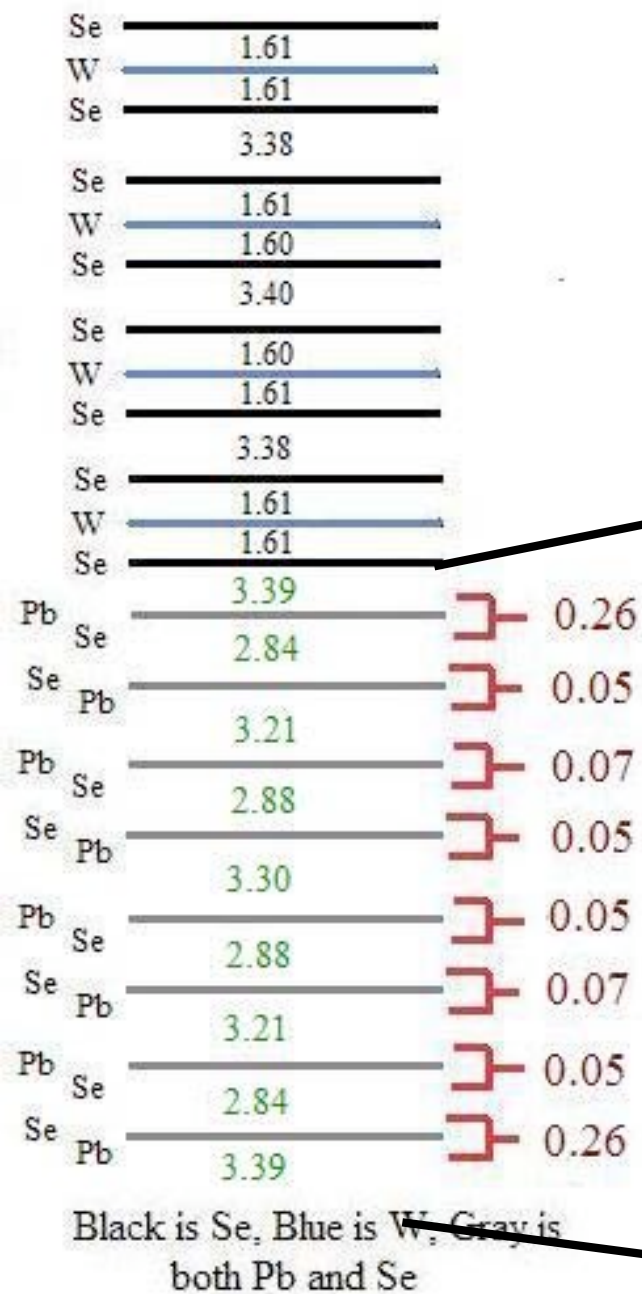
- $(\text{PbSe})_2(\text{WSe}_2)_2$ before annealing
- $(\text{PbSe})_3(\text{WSe}_2)_3$ after annealing
- ▲ $(\text{PbSe})_4(\text{WSe}_2)_4$ after annealing
- ◆ $(\text{PbSe})_2(\text{WSe}_2)_2$ after annealing

Z-contrast image of $(\text{PbSe})_4(\text{MoSe}_2)_4$

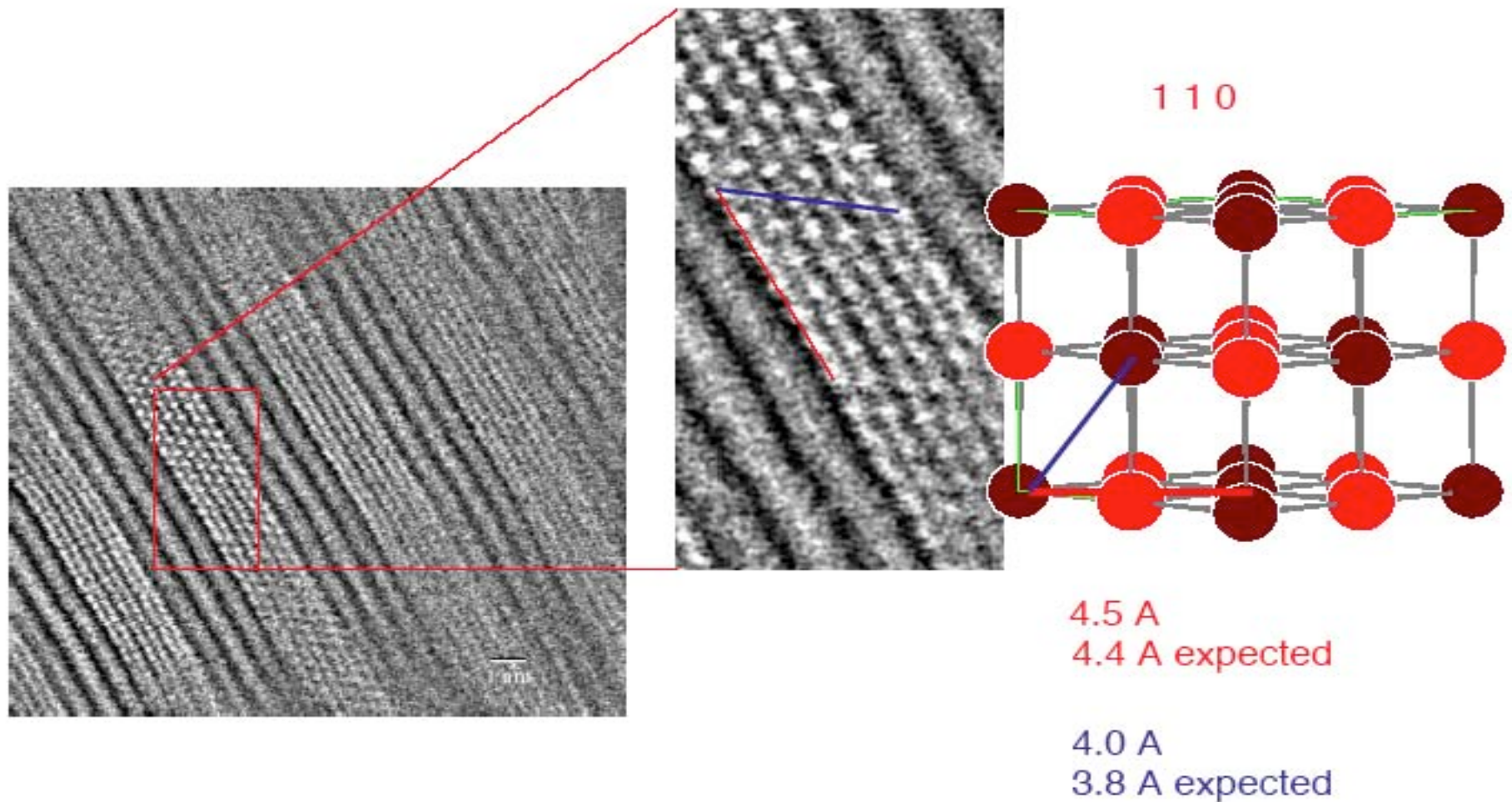




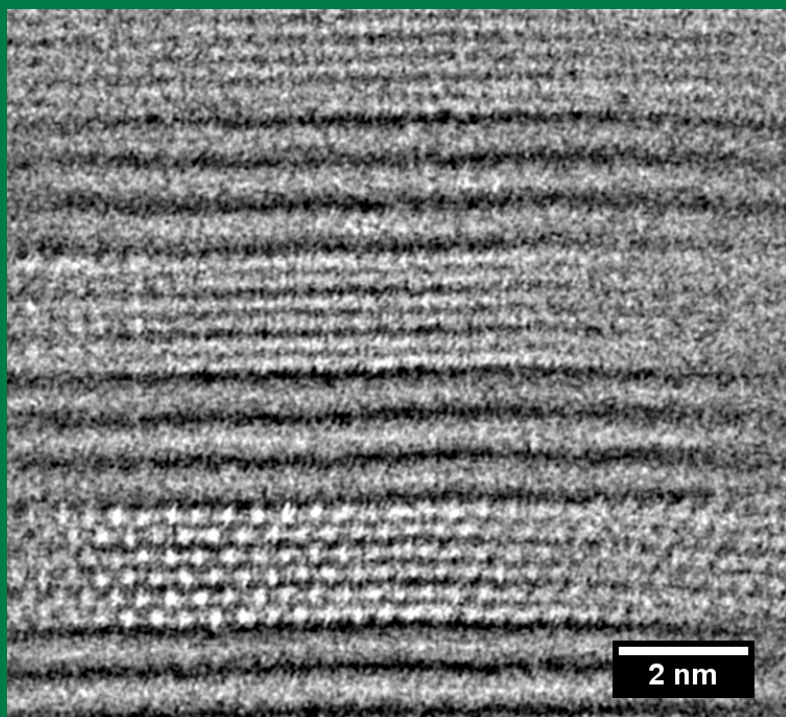
distortion in PbSe layers changes with position



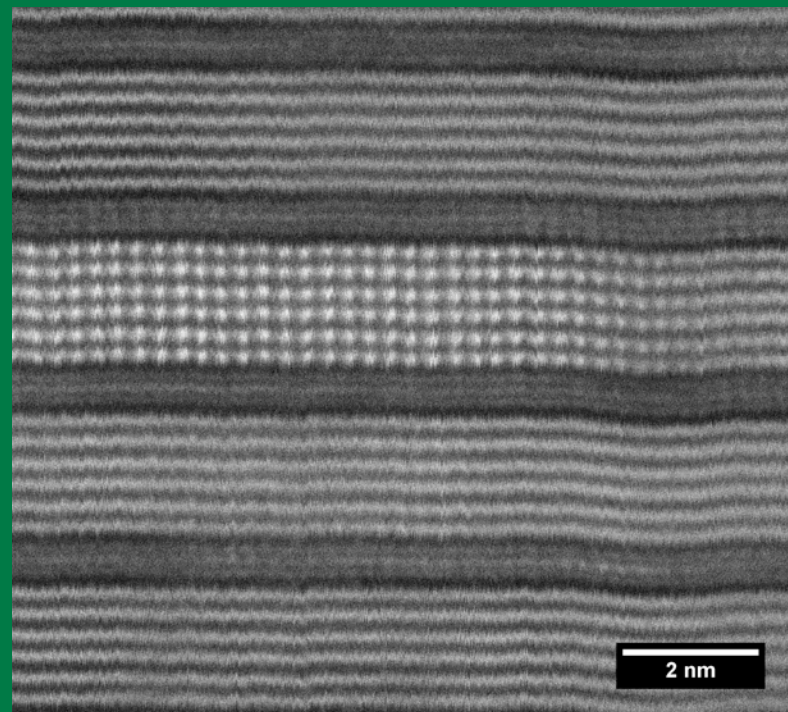
Structure of Misfit Layered Compounds



Distortion Change with m/n ratio in $(\text{PbSe})_n(\text{MoSe}_2)_m$ compounds



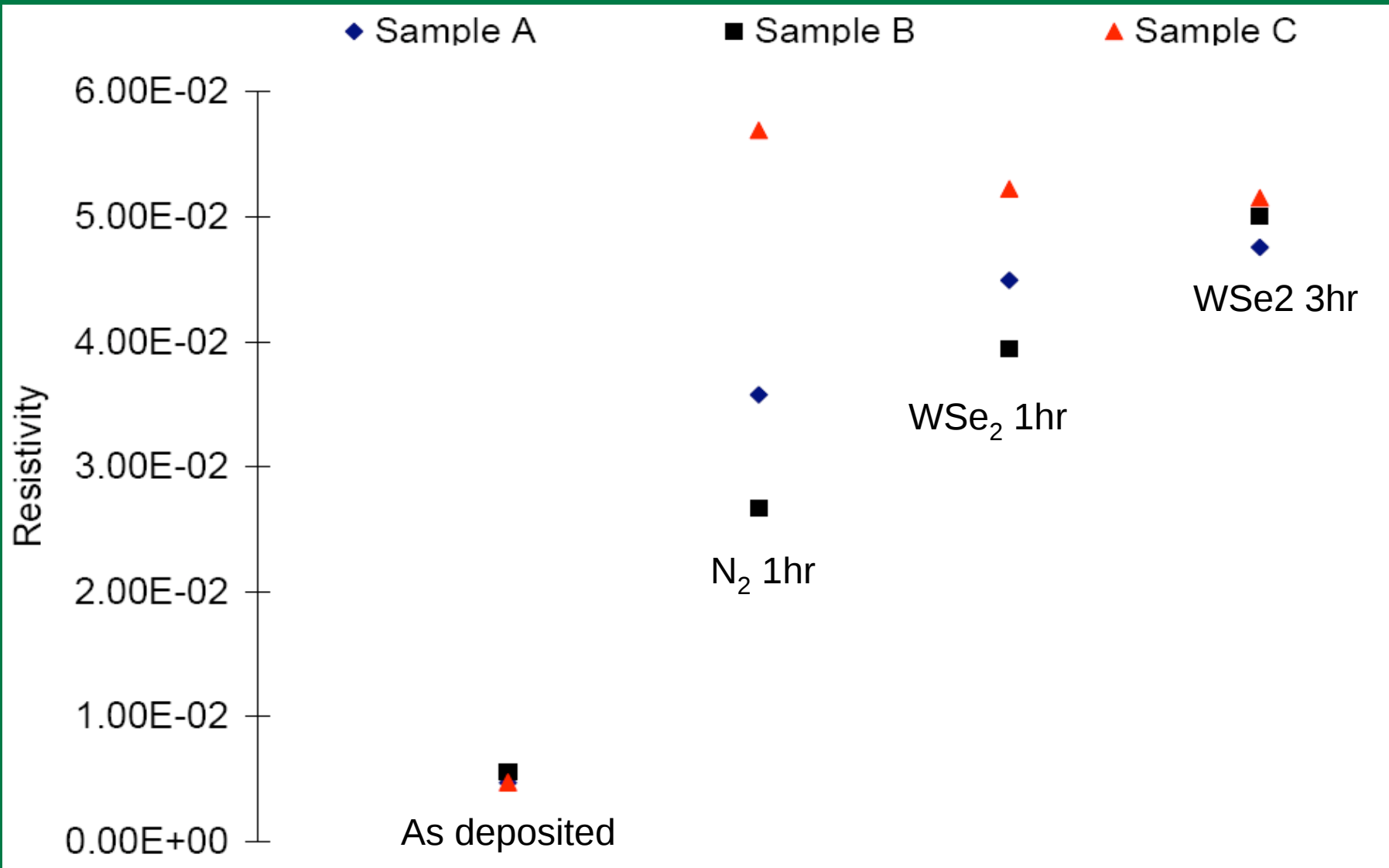
PbSe planes alternate
between 0.28 and 0.35
nm separation in 3:3



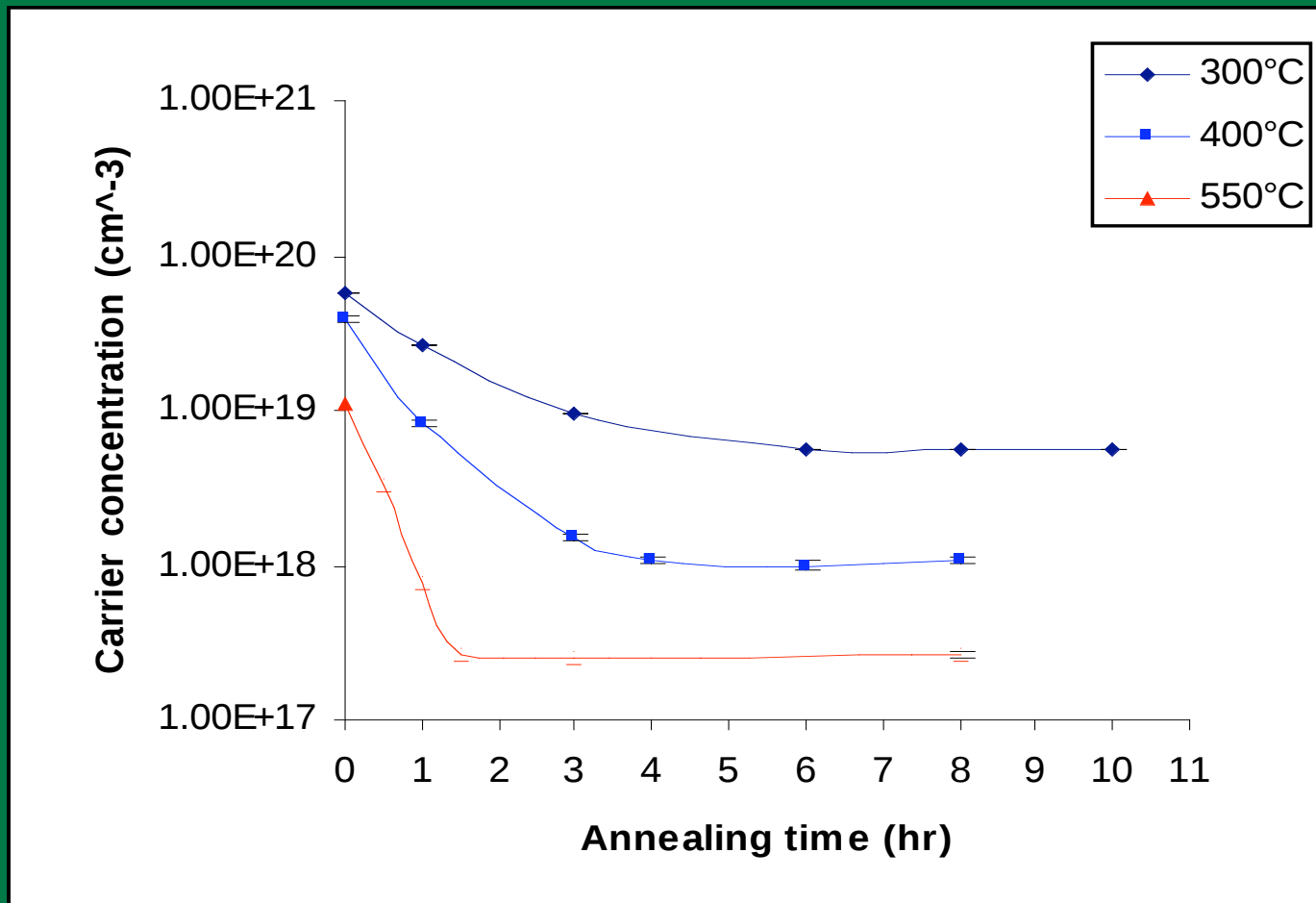
PbSe planes have a 3.1
nm separation in 3:1
compound

Annealing study [$(\text{PbSe})_{0.99}$]₁ (WSe_2) ₁

WSe_2 source of Se vapor pressure

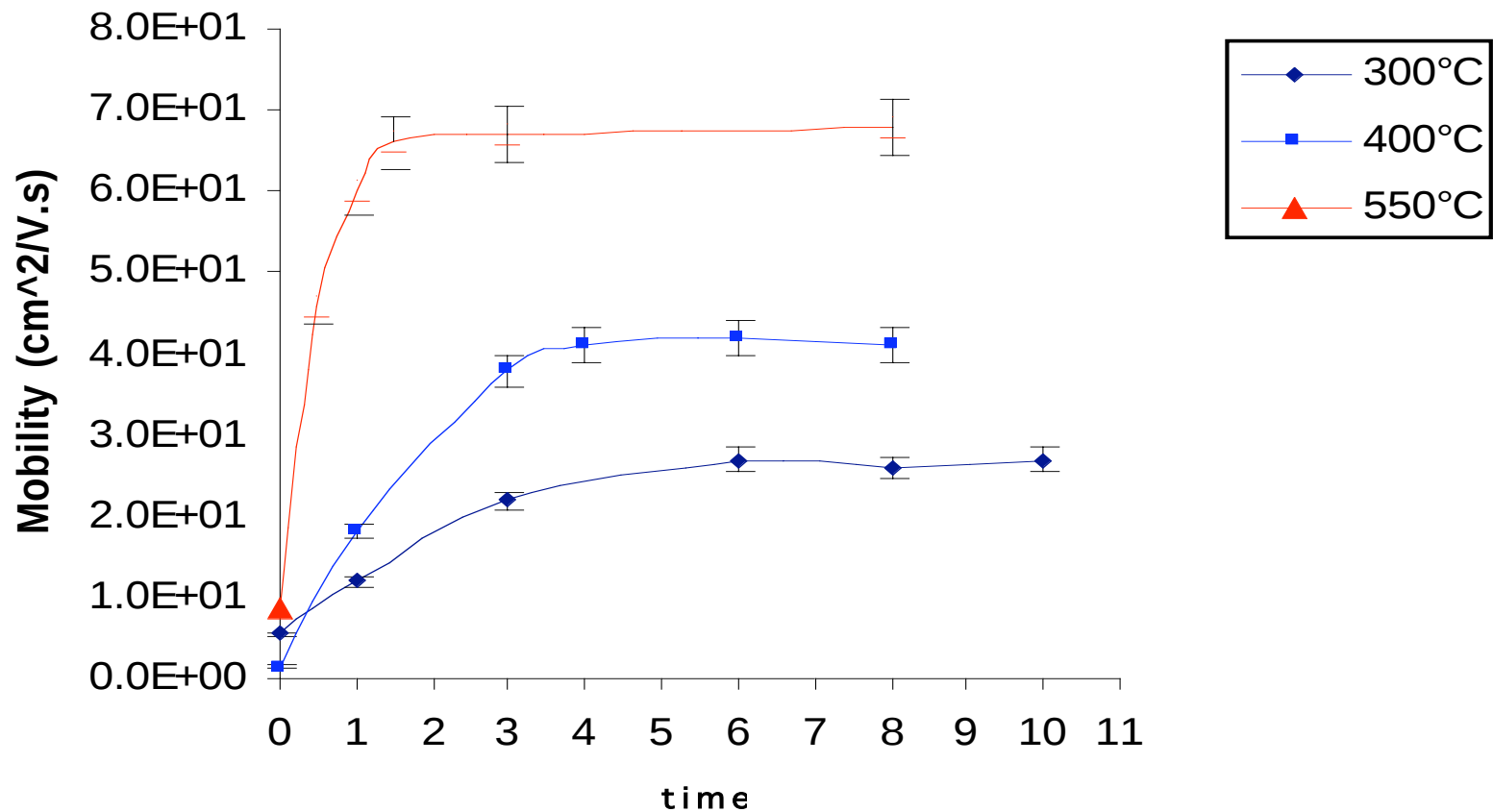


Effect of Annealing Temperature $[(\text{PbSe})_{0.99}]_1(\text{WSe}_2)_1$

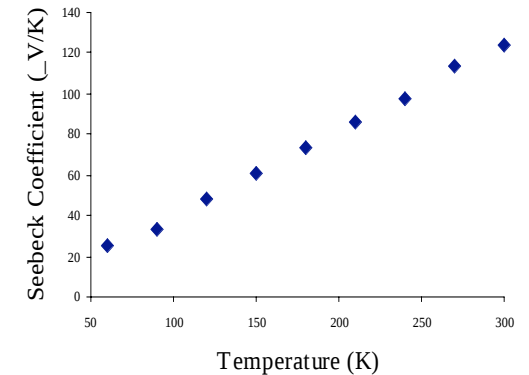
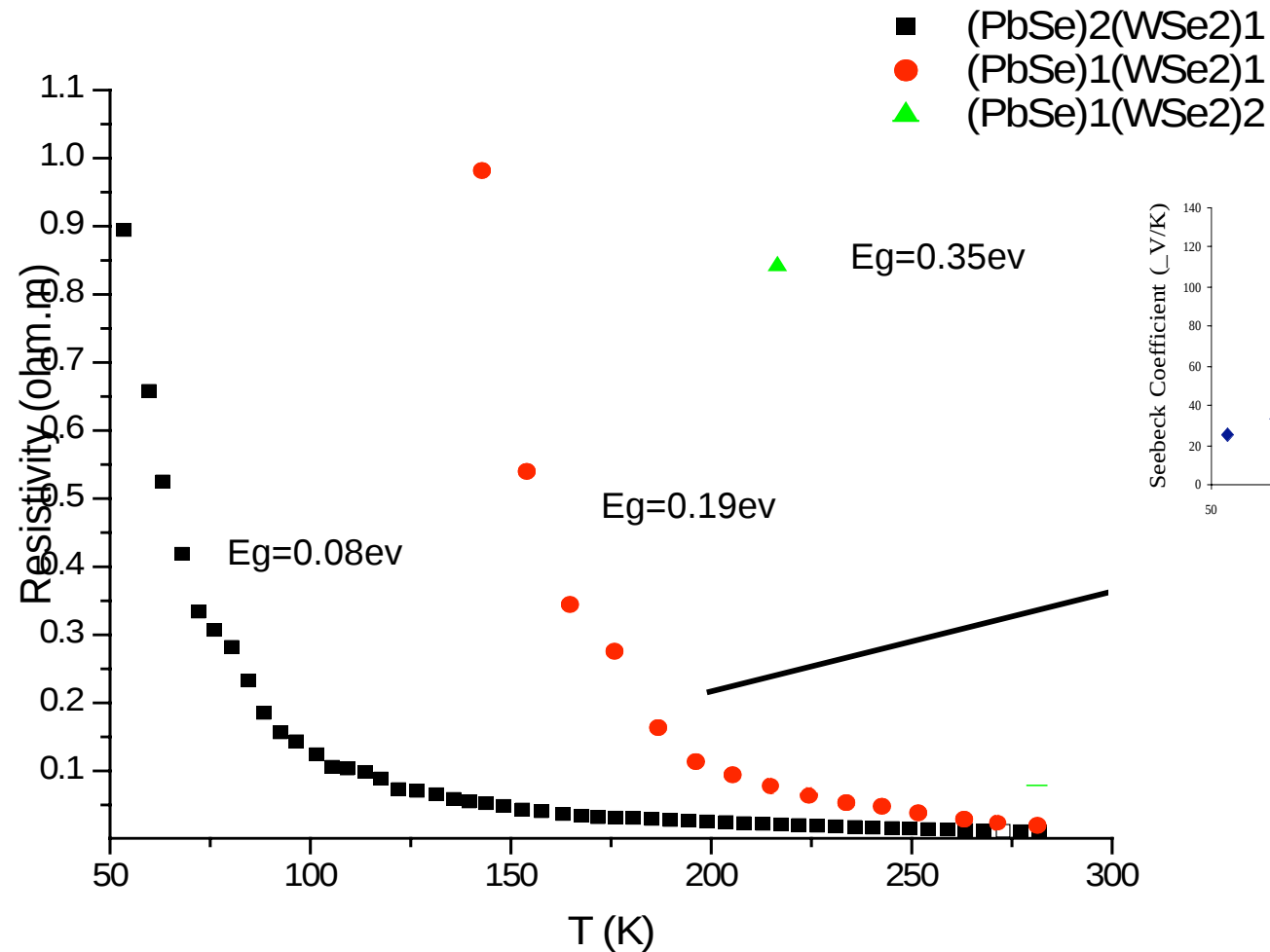


Annealing Eliminates Defects

$[(\text{PbSe})_{0.99}]_1(\text{WSe}_2)_1$



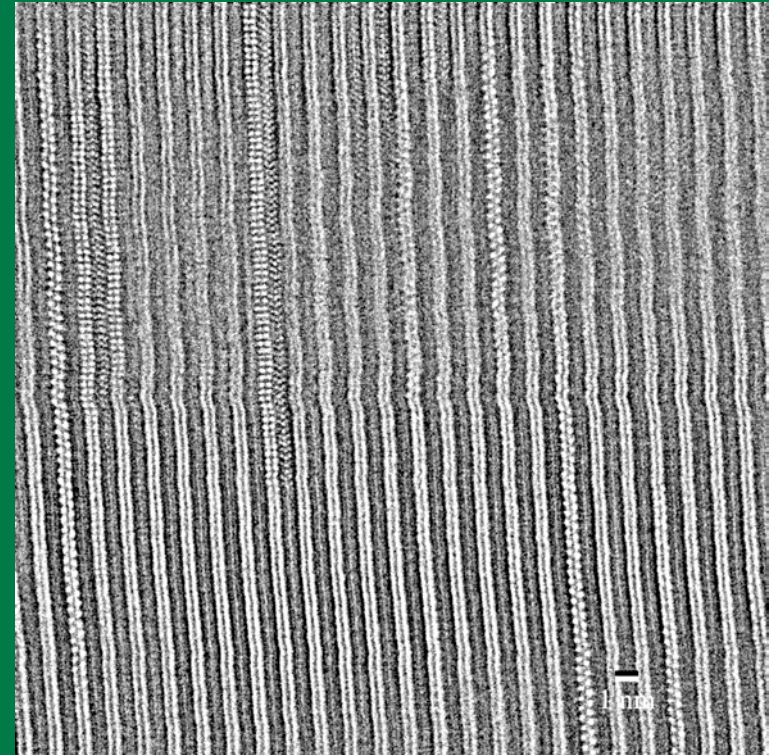
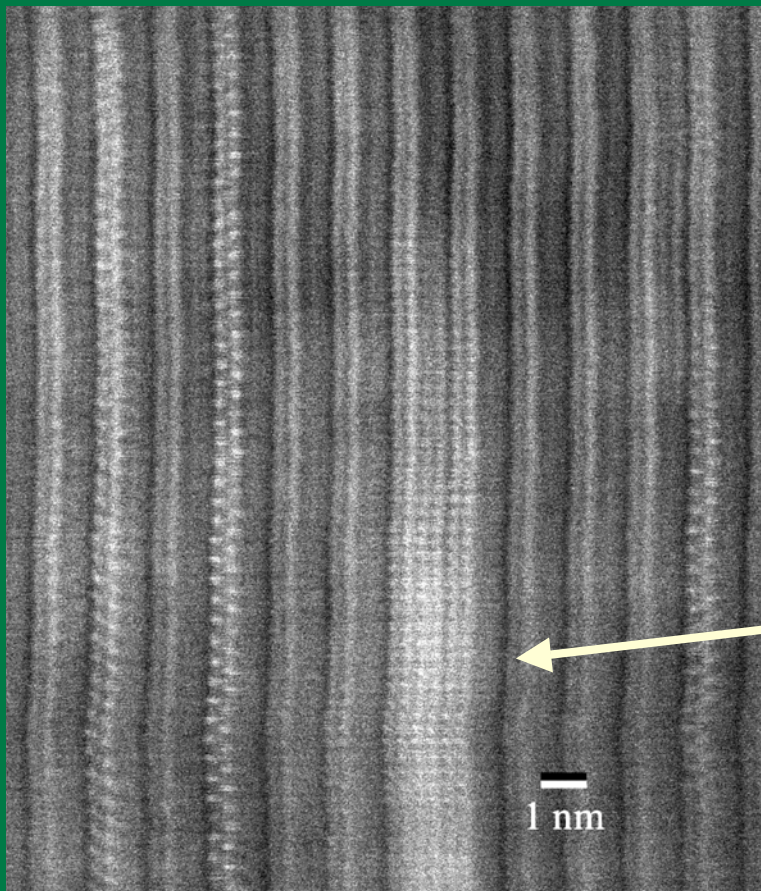
Activation Energy for Conduction varies with n/m ratio



Temperature dependence of Seebeck: heavy doped semiconductors.

Intergrowth Defects in $(\text{PbSe})_1(\text{MoSe}_2)_1$

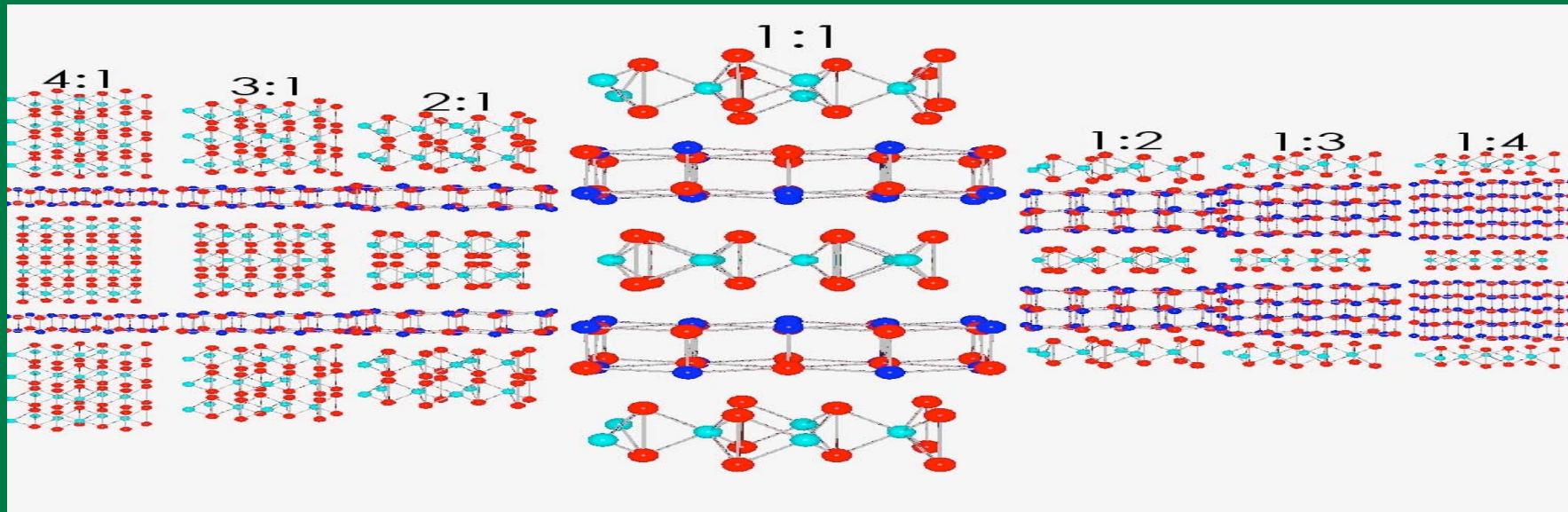
Most of samples have regular misfit structure



Intergrowth defects ($\text{PbSe} \rightarrow \text{MoSe}_2$) can be controlled by initial modulated reactant structure

Summary

- Nanostructure effects structure of the constituent layers.
- Structure effects properties
- A turbostratic disorder observed for all compounds investigated leading to a reduced thermal conductivity.
- Carrier concentration controlled by annealing in Se vapor
- New “nano chemistry” using multi component ABC and ABAC nanostructure.



Acknowledgements

Graduate Students

Myungkeun Noh
Fred Harris
Mary Smeller
Clay Mortensen
Ngoc Nguyen
Colby Heideman
Qiyin Lin

Collaborators

Paul Zschack - APS
David Cahill
Catalin Chirtescu
Li Shi
Anastassios Mavrokefalos
Harold Bottner
Raimar Rostek
Ian Anderson
Mike Anderson

Funding ONR

Murdock Foundation
ARL Air Force
Department of Education
State of Oregon - ETIC and ONAMI

Asahi Kasei
NSF
NIST