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Screen Electrode Materials and Cell Chemistries

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Vehicle Technologies Program



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Overview

Timeline

- Start – Oct. 2008
- Finish – Sep. 2014
- < 8% Complete

Budget

- Total project funding in FY2009
 - DOE: \$350K

Barriers

- An overwhelming number of materials are being marketed by vendors for Lithium-ion batteries.
 - How to select and screen these materials within the effort allocated to this project?
- No commercially available high energy material to meet the 40 mile PHEV application established by the FreedomCAR and Fuels Partnership.
- The impact of formulation and fabrication on performance of electrode materials with a broad variation of chemical and physical properties.

Objectives of This Study

- To identify and evaluate low-cost cell chemistries that can simultaneously meet the life, performance, and abuse tolerance goals for Plug-in HEV application.
- To enhance the understanding of advanced cell components on the electrochemical performance and safety of lithium-ion batteries.

Approach

- Focus of the investigations will be shifted to high energy materials
 - Avoid materials based on rare elements, expensive precursors, or elaborate processing
 - Search battery material suppliers for new materials
- Material screening process will include
 - Test protocol development for lab scale cells
 - Evaluate materials for PHEV use by testing their
 - *Rate capability,*
 - *HPPC impedance,*
 - *Cycle life, and*
 - *Thermal properties (DSC)*
 - Use laboratory scale cells – coin cell, fixture cell, and pouch cell.
- Recommend promising materials for further thermal abuse evaluation and consider for use in longer-term aging studies.

Technical Accomplishments

- Test Protocol Development for material screening
- LiFePO_4 : Mitsui Engineering Shipbuilding (MES), Japan
 - Carbon coated nano size particle
 - Electrochemical enhanced by engineering
- $\text{Li}_{1+w}[\text{Ni}_x\text{Co}_y\text{Mn}_z]_{1-w}\text{O}_2$ (NCM): Toda Kogyo, Japan
 - High tap density
 - Surface fluorination
- LiMn_2O_4 : Tronox, USA - *Domestic supplier*
 - Doping
 - Fluorination
- $\text{LiNi}_x\text{Co}_y\text{Mn}_z\text{O}_2$: SoBright, China
- Other materials tested, but not shown here, include
 - Graphite
 - Separator

USABC Requirements of Energy Storage Systems for PHEV

Characteristics at EOL (End of Life)		High Power/Energy Ratio	High Energy/Power Ratio
Reference Equivalent Electric Range	miles	10	40
Peak Pulse Discharge Power (10 sec)	kW	45	38
Peak Regen Pulse Power (10 sec)	kW	30	25
Available Energy for CD (Charge Depleting) Mode, 10 kW Rate	kWh	3.4	11.6
Available Energy for CS (Charge Sustaining) Mode	kWh	0.5	0.3
Maximum System Weight	kg	60	120
Maximum System Volume	Liter	40	80

- Test procedure and method have been defined in “Battery Test Manual for Plug-in Hybrid Electric Vehicles” by INL 2008.
- The energy requirement is a challenge for PHEV success.

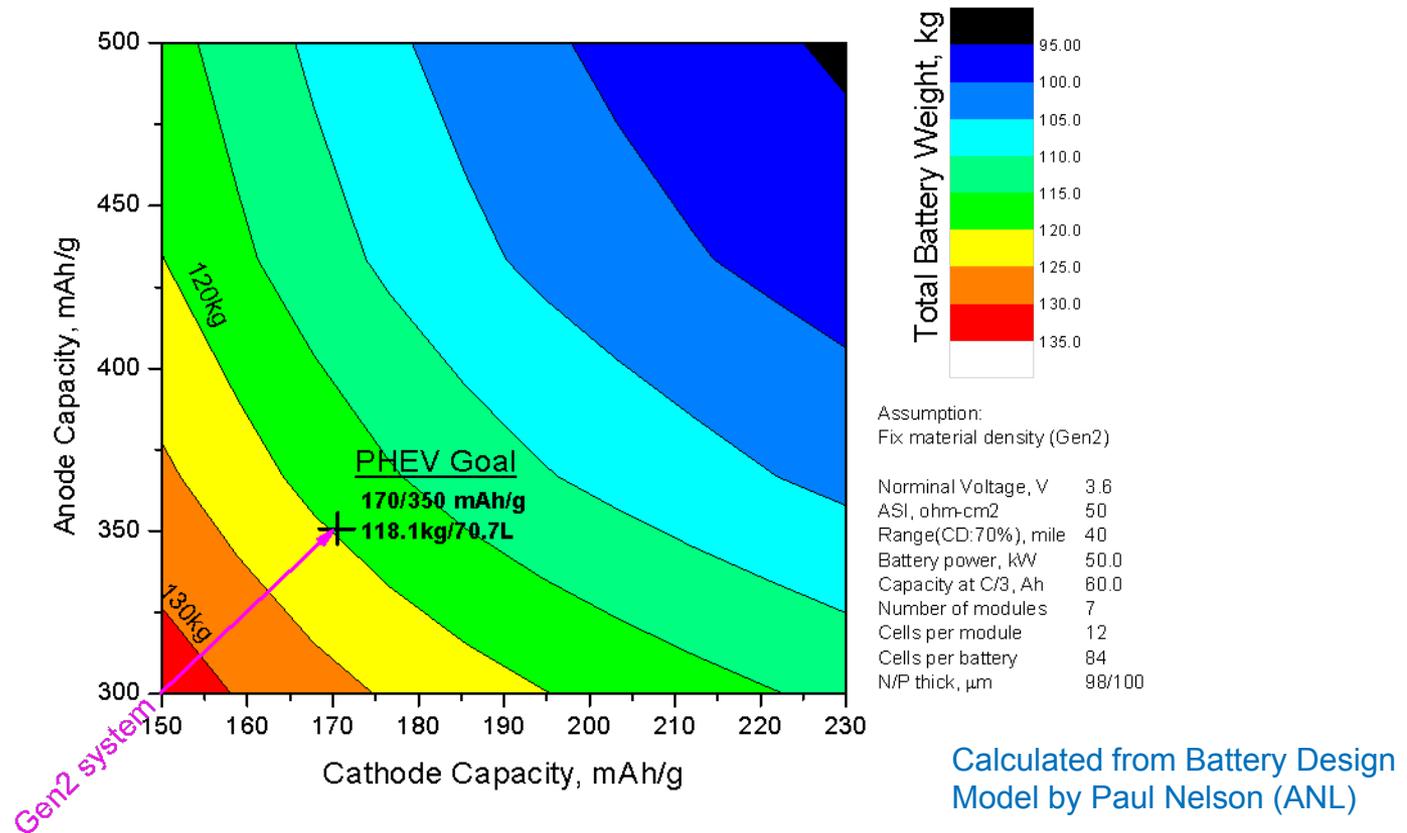
C-rate calculation for PHEV

Available Energy*	Peak rate	CD rate
50%	1.7C	0.3C
60%	2.0C	0.4C
70%	2.3C	0.5C

* The calculation includes 30% energy margin.

- The discharge pulse rate is equal to 2C using 60% energy in the battery.
- The charge depleting (CD) rate is equal to C/3 when 50% of battery energy will be used.

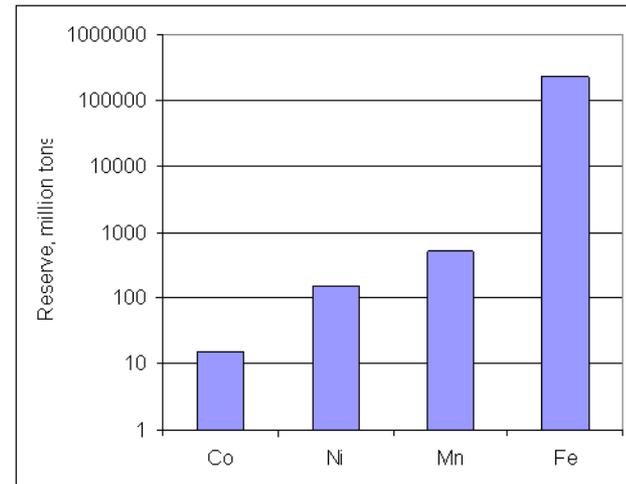
Capacity Requirement for Electrode Materials to Reach PHEV Weight Goal



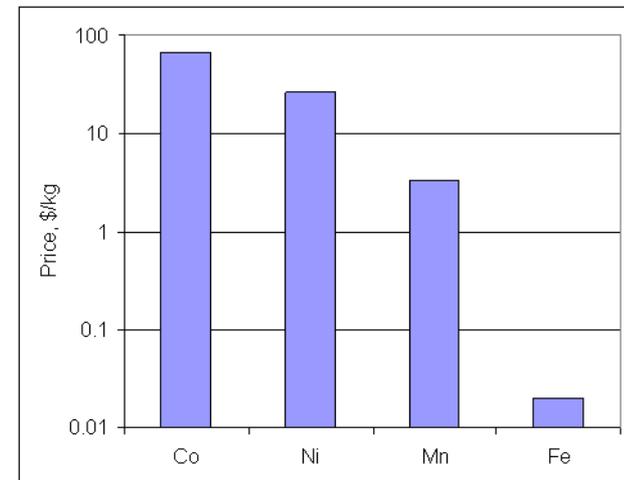
- The specific capacities of anode and cathode materials should be above 350 and 170 mAh/g, respectively, to meet the PHEV goal with weight and volume requirements.

Battery Cost Analysis

- The unit cost of 20 kWh battery is more than \$4500, according to Battery Design Model by Paul Nelson (ANL).
- Material cost makes the major contribution to the battery cost.
 - Active materials, positive and negative, take half of material cost.
 - Electrolyte and separator are the next major contributors.
- Manganese and Iron base materials has the potential to reduce the cost.
- New high energy density anode will significantly reduce the cost.
- New electrolyte and separator should be investigated.



<http://www.usgs.gov/>

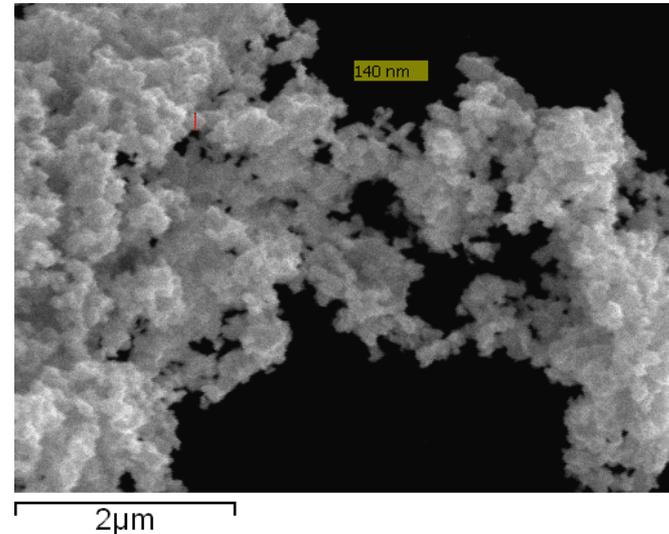


<http://www.metalprices.com/>

LiFePO₄: Mitsui Engineering Shipbuilding (MES)

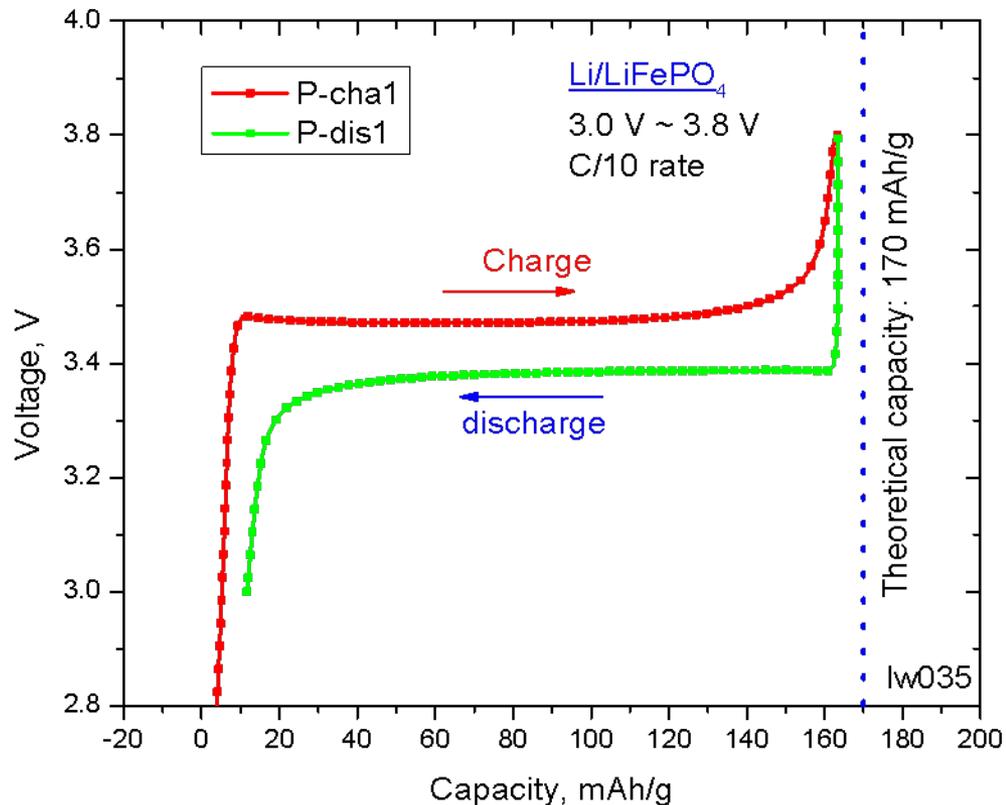
Advantage of MES LiFePO₄

- Low cost
- Thermal stability
- High specific capacity
- High power capability
 - Carbon coating
 - Nano size Particle



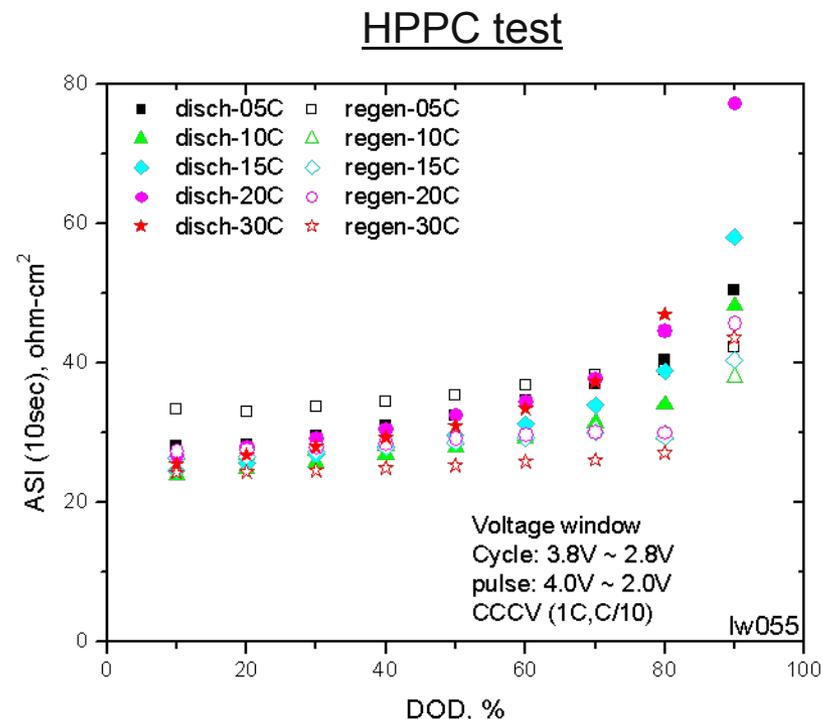
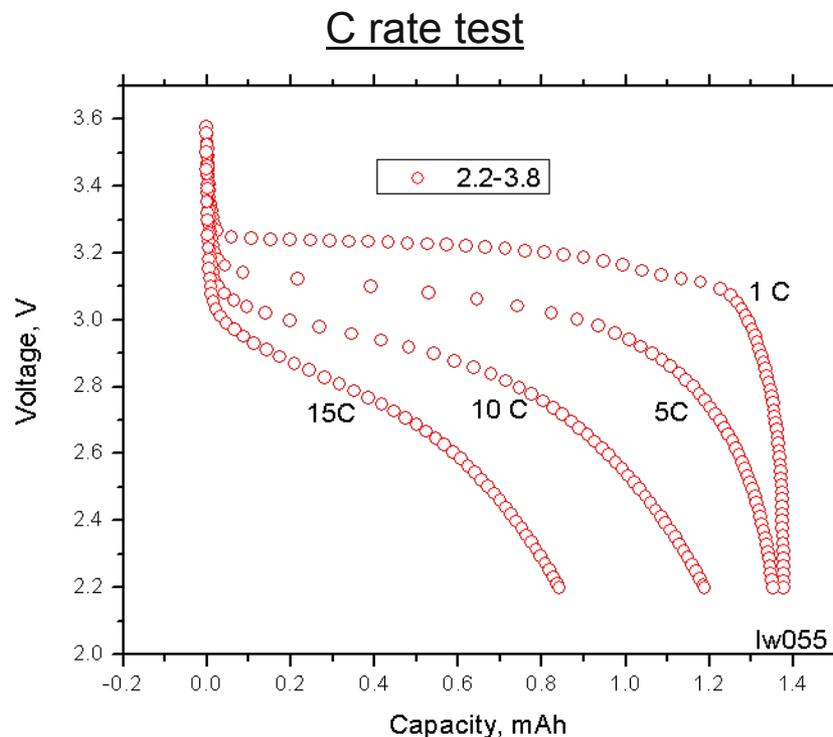
Parameter	Current target value	Measured value	Measurement method
Carbon content	5.3 ~ 6.1 wt.% (target: 5.6 wt%)	6.0wt%	High-frequency furnace methode
Specific surface area	18~26 m ² /g	23.9m ² /g	BET method
Bulk density	0.18~0.28 g/cm ³	0.21g/cm ³	-
Tap density	0.40~0.52 g/cm ³	0.44g/cm ³	-
Particle size (D50)	1.1~1.5µm	1.4µm	Laser scattering method

High Specific Capacity of MES LiFePO_4



- Specific capacity of LiFePO_4 is determined to be 160 mAh/g with about 5% irreversible capacity loss during the 1st cycle between 3.8 V and 3.0 V at C/10 rate.

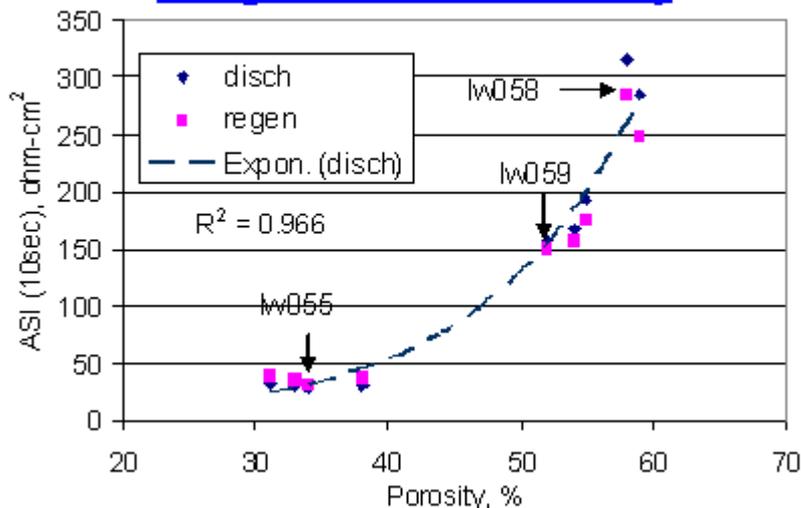
High Rate Performance of MES LiFePO_4



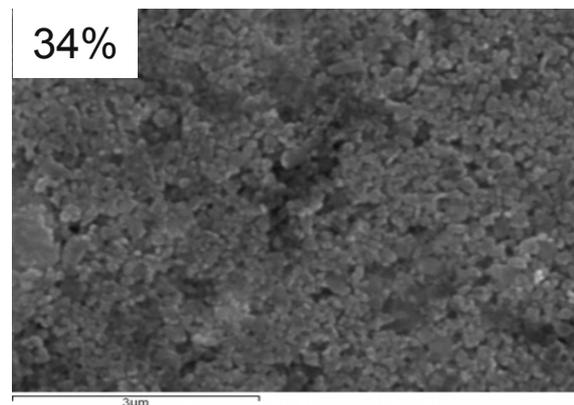
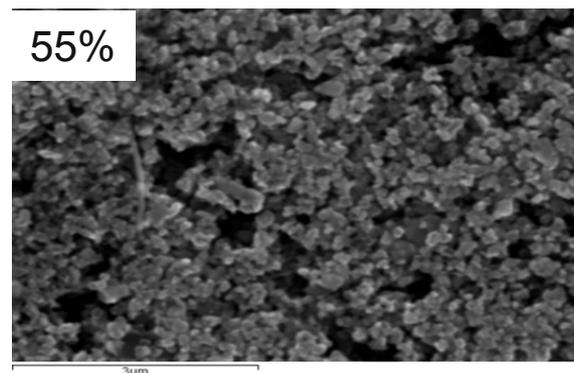
- The Graphite/ LiFePO_4 cell can deliver more than 80% capacity at 10C rate.
- ASI results of HPPC is comparable to Graphite/ $\text{LiNi}_{0.8}\text{Co}_{0.15}\text{Al}_{0.05}\text{O}_2$ lithium ion battery.

Engineering Effect on MES LiFePO_4 Performance

ASI @ 10% DOD vs. Porosity



- The ASI increases with increasing electrode porosity.
- According to SEM image, much better contact between the particles are obtained after calendaring.
- The carbon/carbon matrix is likely to be the major cause for the high resistance of the cell with high porosity.



SEM image of electrode

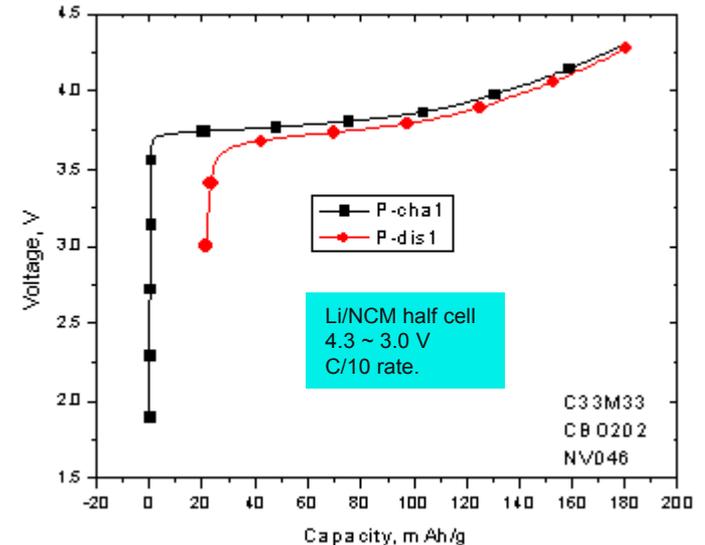
Toda $Li_{1+w}[Ni_xCo_yMn_z]_{1-w}O_2$ (NCM):

Serial of lithium rich NCM materials with various Ni, Co, and Mn ratios

Advantages of Toda materials:

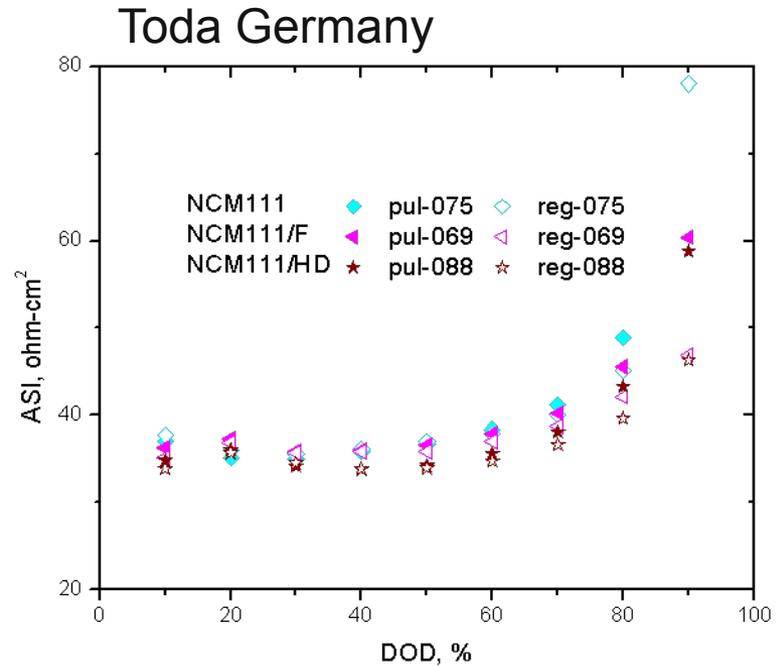
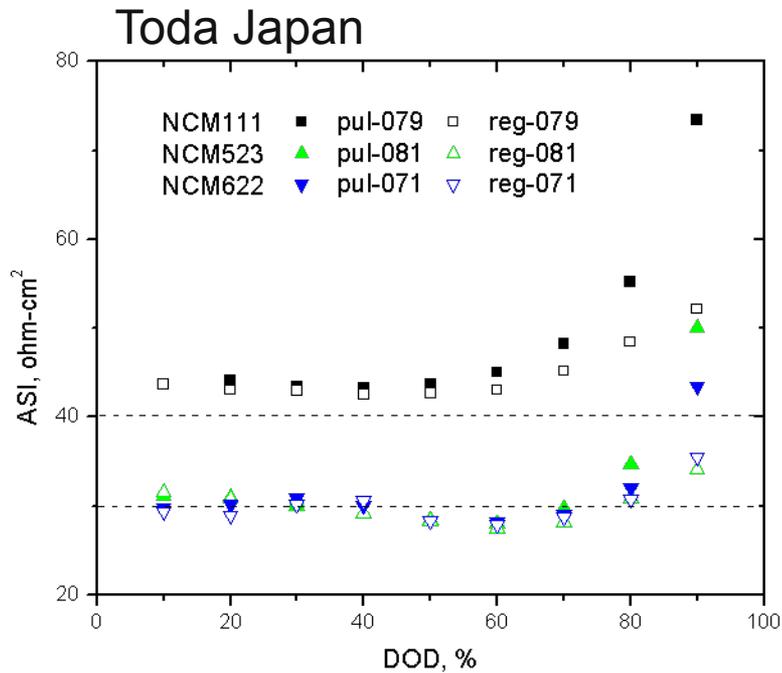
- ❖ High energy density
 - High specific capacity,
 - High tap density,
 - More capacity with nickel rich material
 - High electrode loading density
- ❖ Better stability with surface fluorination

Voltage Profile of Toda NCM



Sample		BET	D50	Tap density	Capacity, mAh		coulombic	ASI
producer	ID	m2/g	μm	g/cm3	charg-1	disch-1	%	ohm-cm2
Japan	NCM111	0.39	6	2.43	169.72	148.65	87.59	43.8
	NCM523	0.23	10.2	2.37	191.26	162.53	84.98	28.5
	NCM622	0.22	10.8	2.61	201.18	166.76	82.89	28.5
Germany	NCM111	0.59	8	2.4	177.88	156.66	88.07	34.8
	NCM111/F	0.36	8.2	2.3	181.70	159.88	87.99	35.8
	NCM111/HD	0.26	8.5	2.8	170.00	147.33	86.66	36.2

5C HPPC Impedance Results of $Li_{1+w}[Ni_xCo_yMn_z]_{1-w}O_2$

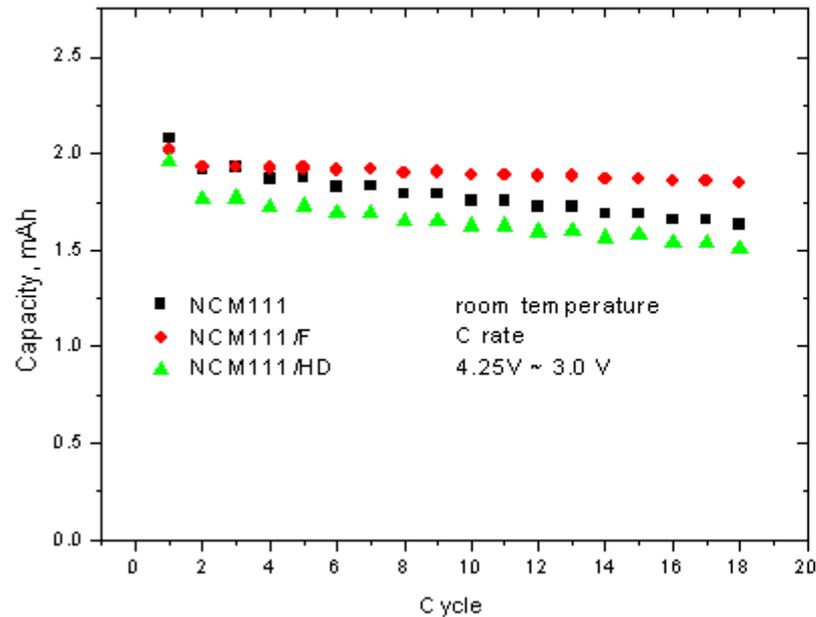


- NCM523 and NCM622 have comparable ASI (30 ohm-cm²) to NCA cell (Gen2), which is lower than that of NCM111 electrode.

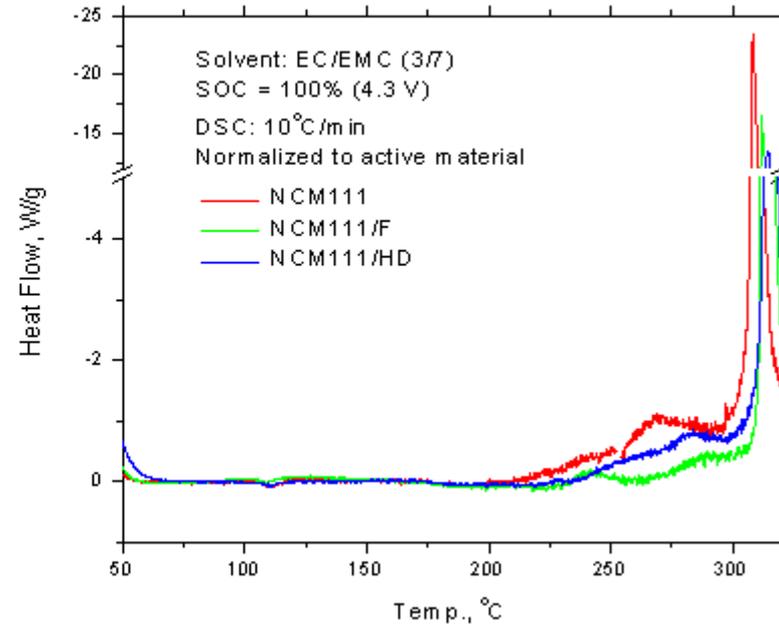
- The pulse ASI of fluorinated NCM and high tap density NCM are similar to each other.

Fluorination Effect on Toda $Li_{1+w}[Ni_xCo_yMn_z]_{1-w}O_2$

Cycle - Toda Germany



DSC - Toda Germany

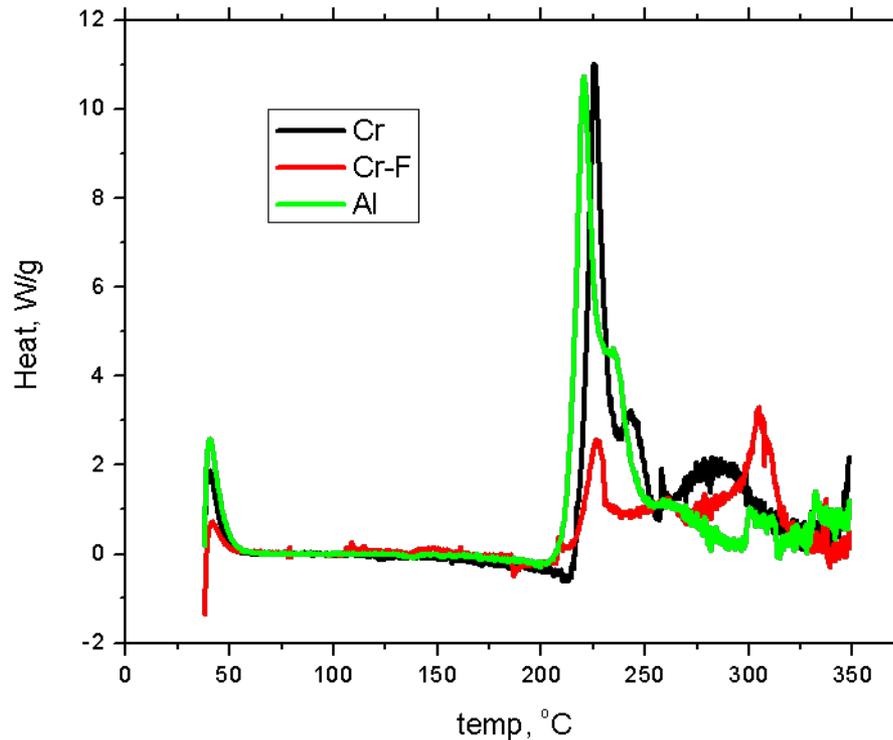


Advantages of fluorinated NCM

- Improve the cycle life under 1C rate continuous charge and discharge at room temperature.
- Reduce the heat generation at fully charged state with electrolyte.

Fluorination Effect on Tronox LiMn_2O_4

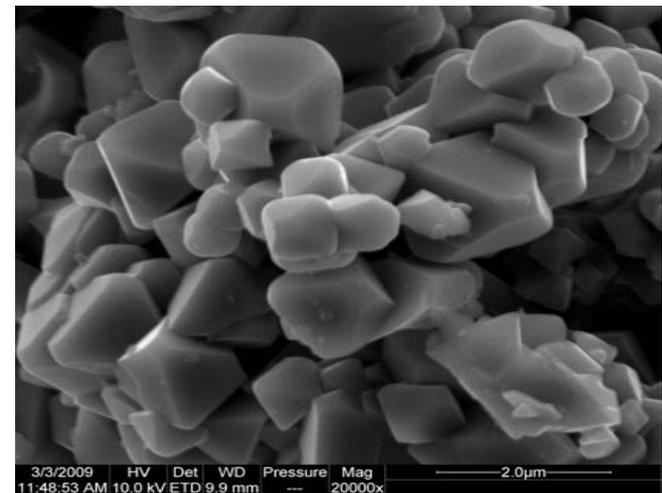
DSC of LiMn_2O_4



- The thermal stability of LiMn_2O_4 can also be improved by fluorination.

- Tronox is domestic manganese spinel supplier.
- The Tronox spinel with following characteristics
 - Aluminum doped
 - Chromium doped
 - Surface fluorinated

SEM of LiMn_2O_4

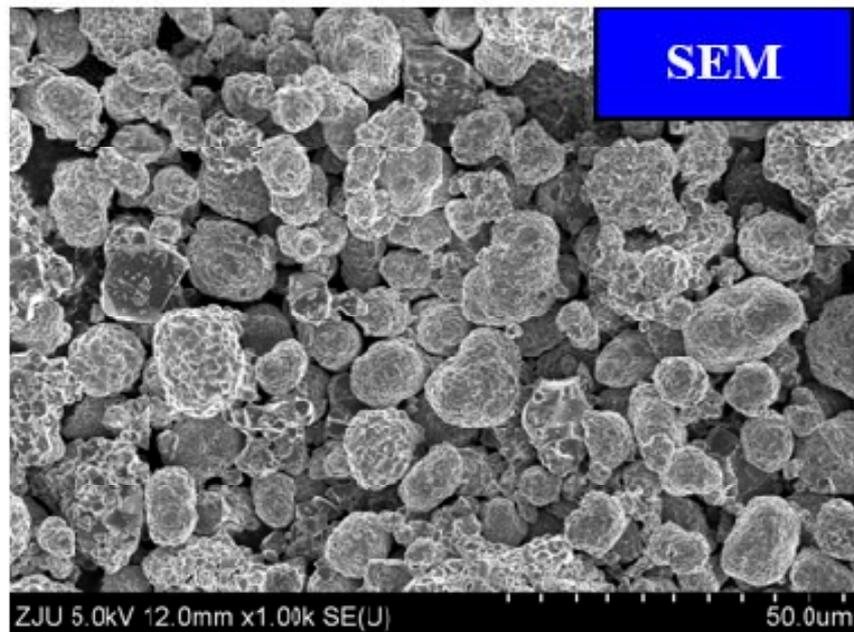


SoBright NCM Material ($\text{LiNi}_x\text{Co}_y\text{Mn}_z\text{O}_2$)

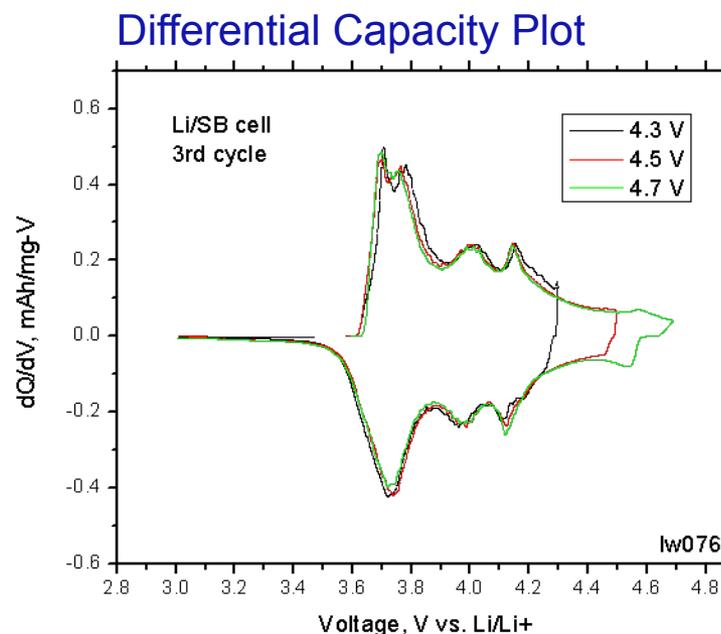
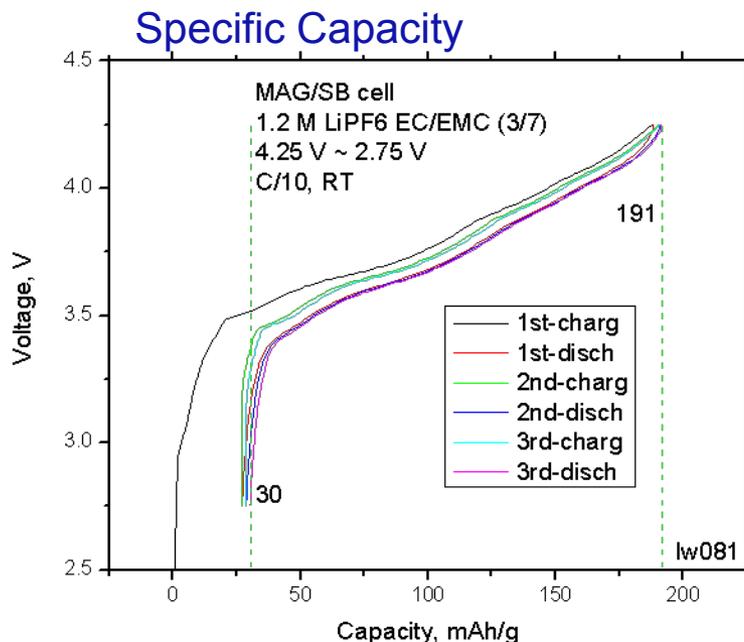
- The NCM might provide
 - Higher energy density
 - Higher power

Physical Properties

Item		Typical value
Ni+Co+Mn (wt%)		58.6
PSD (μm)	D10	6.08
	D50	11.13
	D90	22.73
Tap Density (g/cm^3)		2.38
SSA (m^2/g)		0.25
Morphology		Spherical particle
pH value		11.08

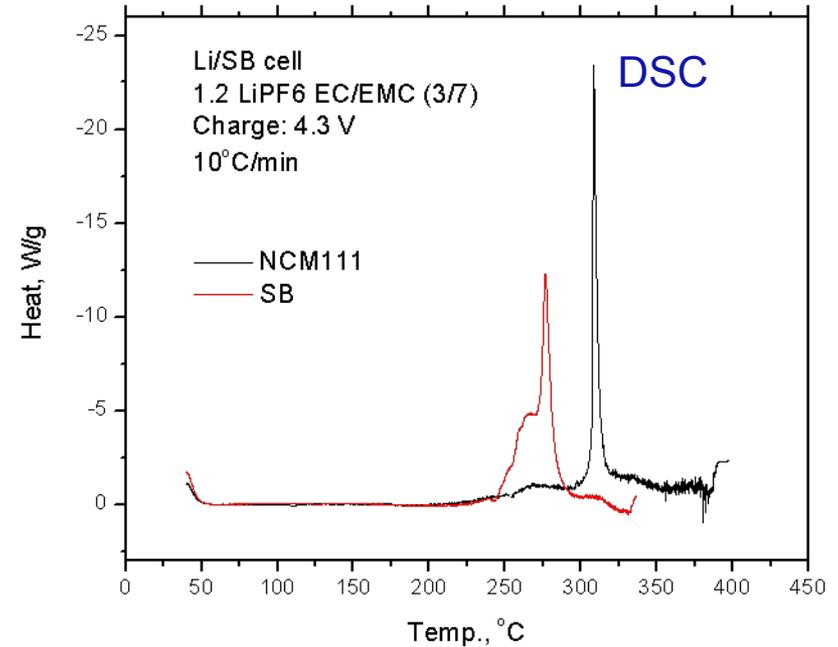
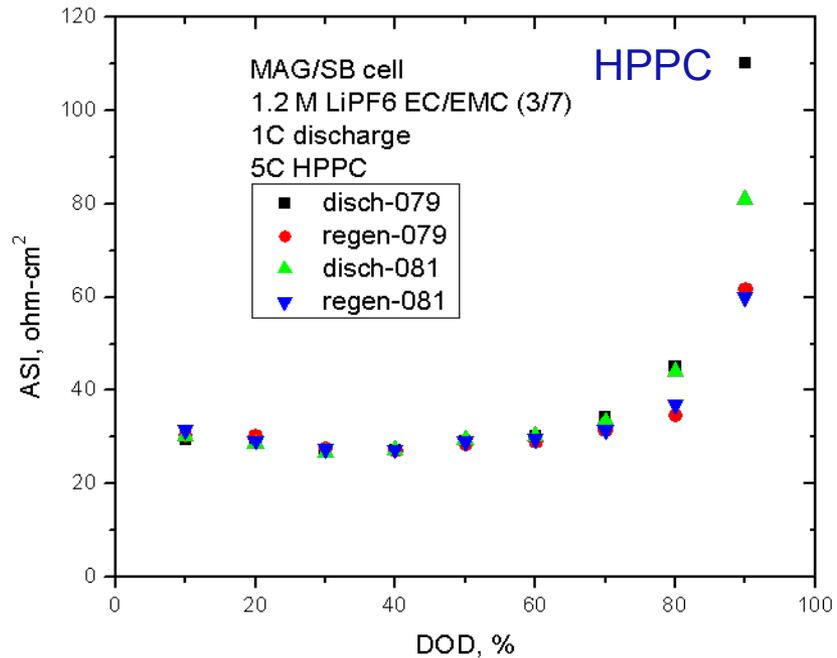


High Capacity of SoBright Material



- The reversible capacity between 4.3 V and 3.0 V is about 162 mAh/g with 14% capacity loss during first cycle.
- The cutoff voltage can be extended to 4.5 V and 4.7 V with more than 170 and 180 mAh/g reversible specific capacity, respectively.

Power Capability and Thermal Stability



- The ASI is about 30 ohm-cm², which is comparable to NCM material.
- The thermal stability of SB blend is comparable to NCA material.

Plans for Next Fiscal Year

- Investigate materials from different suppliers for high energy (PHEV) and high power (HEV) application
 - Hitachi Chemical's anode materials
 - Nano-size LiMnPO_4 from HPL
- New binder study
- New electrolyte study – Fluorinated solvent (Daikin)
- New anode materials search and evaluation
 - High energy density
 - Stable material
- Continue to evaluate advanced cathode, anode, binder, separator, and electrolyte systems as they become available from various sources.

Summary

- MES olivine can be used for high power applications. Electrode optimization is needed to achieve its best performance.
- Variations of Toda $\text{Li}_{1+w}[\text{Ni}_x\text{Co}_y\text{Mn}_z]_{1-w}\text{O}_2$ (NCM) materials indicate that the nickel rich material can deliver high energy density. The surface fluorination can further improve its thermal and cycleability.
- Tronox spinel can supply high power capability. The surface fluorination improves the thermal stability.
- SoBright NCM material has high power and high energy density. Its thermal stability is comparable to NCA (Gen2) electrode.

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